

24-1730

6/6/50
90

potash - pro.

K₂CO₃

manuf. of, P 928²

purification of, P 4361¹

KCl

solub. in EtOH in presence of other salts,

5576²

K₂SO₄

manuf. of, 4593²

15 - 1921

Kcl

152. Potassium chloride from furnace dust. C. Anderson and F. S. Moon, Jr. U. S. 1,354,642, Oct 5. Furnace dust containing K₂SO₄ and KCl is leached with a soln. containing sufficient CaCl₂ to convert the alkali metal sulfates into chlorides and KCl is recovered from the soln. thus obtained.

6-7922

K₂CO₃

KHCO₃, maint. of, P 994

K₂

from bitterns, P 3176

from cement-silica dust, P 999

K₂CO₃ - yea

(

O/

16 - 1922

75.205

994. Potassium carbonate. J. F. Harlow
U.S. 1,400,542, Dec. 10. alk. bitter
such as U.S. alk. plains brine is mixed
with $MgCO_3 \cdot 3H_2O$ and the mix. is
treated with CO_2 at a temp. of about
 20° to ppt. $KH Mg$ carbonate which is
then sep. and heated to form acid
 $MgCO_3$ and $KHCO_3$ in soln.

16 - 1922

KCl

317.6^g potassium chloride from bittern.
J. F. Harlow. U.S. 1, 452, 571, July 11.
a bittern contg. K_2CO_3 and Na salts
concd. to approx. satn. is treated
with $CaCl_2$ in proportion chem.
equiv. to the K present and heated
to 95° to ppt. $CaCO_3$ and leave the
K in soln. as KCl. The soln. is then
concd. to 0.75 its former vol. to
affct cryptn. of Na salts while
the soln. was still hot and after
sepn. of these salts the soln. is
cooled to about 20° to obtain KCl.

929. potassium chloride from ^{anal. 4th} dust. F.S. iron. U.S. 1, 402, 173, 400.
3. Current silver plate dust is agi-
tated in H₂O to dissolve Na and K
salts and the soln is treated with
sufficient CaCl₂ to convert Na and
K sulfates into chlorides. KCl is
obtained from the soln. by frac-
tional crystn.

C

O/

17-1923

K-00 - yes

K-01 - yes

K-04

manuf of, P 15362

C

Q

17-1923

K2SO4

1536 Potassium sulfate. C. H. H. 21.8.
1, 11, 16, 185, Feb. 20. CaSO₄, suspended
in an aqueous soln. of KOH is
reacted on by CO₂ to form a satd.
soln. of K₂SO₄ and the latter is
precip. from the soln. on cooling.

17-1924

U.S. 30
1/15

K₂CO₃

K₂HCO₃, manuf. by electrolysis of KCl
soln., p. 512.

K₂CO₃, soly. of sol. in aq. soln. of,
p. 437.

K₂SO₄ - no.

K₂SO₄ - no.

C

O

18-1924

K₂CO₃

512 Potassium bicarbonate by electrolysis
of potassium chloride solutions. R.
Luchy. U.S.P., 447, 086, Dec. 11. A
KCl soln. is electrolyzed and the KOH
formed in the catholyte is reacted
again with KHCO₃ to form K₂CO₃.
Fresh KCl is added to replace that
consumed by the electrolysis and
KHCO₃ is recd. from the end liquor
contg. K₂CO₃ by introducing CO₂.

18-1924

K₂CO₃

2493. Water solubility in homologous series. H. Föhner. Ber. 57 B, 516-5 (1924); cf. GA. 17, 106, and earlier papers — [on the following is of interest] — the soly. (mols. per l.) of Me, Et, Pr, and Bu also in 50 cc. soln. contg. 20 g. K₂CO₃ are 10.00, 1.54, 0.24, and 0.039 resp.; with less K₂CO₃ (5.0 and 3.5 g. per 50 cc. soln.), the ratios of the soly. of 1 also to that of its next higher homolog are of the same order of magnitude (about 4) as those of the higher also in H₂O alone.

C

O/

19 - 1925

K-403 - 200

K-41 - 200

K-504

compd. with H₂PO₄, 2178'

177 (with an 178). Constitution of phosphoric acid. J. Friedl. Rad. Acad. Sci. Zashch 248, 16-37 (1923). — [only the following is of interest] — when K_2SO_4 is dissolved in conc. H_3PO_4 and the soln. poured into alc., a crypt. precipitate, $2K_2SO_4 \cdot H_3PO_4$ is formed. If hot H_3PO_4 is added with K_2SO_4 until crystals appear on cooling, and then mixed with alc., crystals with the common $K_2SO_4 \cdot H_3PO_4$ sep.

4/6/30
JMM

20-1926

K₂CO₃ - mo.

KCl - mo.

K₂SO₄ - mo.

6/10/50
mo

21-1927

4/6/53
70

KUCO - mo.

KU - mo.

KUSO - mo.

22-1927

6/4/50
8/13

P-Tell

as sugar industry by-product, 1868

K₂CO₃ - year

KCl - year

K₂SO₄ -

from K₂Fe(CN)₆ by airt, 1860

C

d

22 - 1928

potash

11/17/28

1768. Certain by-products of the sugar industry. Their production and use in South Africa. H. C. Dymond. Planter Sugar Ind. 29, 3 (1927). - a review discussing the production of motor fuel, yeast, potash, NH_3 , amines, cyanides, etc.

22-1928

K₂SO₄

U.S.P.

2640. Separation of potassium sulfate
① from crude potassium ferricyanide
by crystallization. V. P. M'inkii
and N. P. Lavin. Trans. State Inst.
Applied Chem. (Moscow) 1927, no. 5,
8-17. — K₂SO₄ is the principal im-
purity of technical K₄Fe(CN)₆ obtained
by the calcination of animal mat-
ter with K₂CO₃ and Fe. Technical
K₄Fe(CN)₆ sometimes contains
up to 10% of this impurity, which
is due to the oxidation of the S of
organic matter, and also con-
tains a little chloride and carbonate.
The only practical method of puri-
fying this K₄Fe(CN)₆ is by crystal-
lizing from H₂O. The dissolved system
K₄Fe(CN)₆ - K₂SO₄ is found to be
remarkable in this respect that
the line of sep. of crystn. fields
is almost a st. line parallel to

22-1923

K₂CO₃

6/6/50
JL

2640. the aim of abstracts; this circum-
stance is particularly favorable to
the separ. of the two salts. Tables of
solubilities of mixts. of $K_4Fe(CN)_6$ and
 K_2SO_4 at 25° , at 40° , at 55° and at
the b.p. temps. are shown. If the
admixture of K_2SO_4 is small, the
recommendation is to sep. it by
cooling the satd. soln. of the 2
salts at high temp. If the propor-
tion of K_2SO_4 is large, it is ne-
cessary to allow it first to cool
moderately, then sep. the part
of K_2SO_4 which crystallizes and
finally cool the satd. soln. for
the separ. of $K_4Fe(CN)_6$.

C

d

23 - 1929

6/6/50
JAD

potash - zero

K₂CO₃ -

analysis: K₂SO₄ - 14.0 - 14.6766

KCl - zero

K₂SO₄ - zero

n.w.2.

24 - 1930

K₂CO₃

615

928. manufacture of potassium carbonate
société anonyme alcalina. Belg.
360,703, June 29, 1929. K₂CO₃ is ob-
tained by decompos. of an acid
double salt of Mg and K which is
itself obtained by the reaction of
a concd. soln. of a K salt satd.
with CO₂ on neutral MgCO₃. The
latter is produced by adding MgO
or Mg(OH)₂ to the CO₂-satd. soln. of
K salt.

7

24-1950

KCl

5576. Effect of added salts upon the solubility of other salts in ethyl alcohol. Ralph T. Leonard and Walter C. Schumb. J. Am. Chem. Soc. 52, 3962-7 (1930). — The solubilities of KCl, KClO₄ and Ba(NO₃)₂ in the presence of other salts in EtOH have been measured and the results compared with those predicted by the inter-ionic attraction theory. Marked deviations from the original Debye-Hückel theory are shown but results agree qualitatively with that of Bjerrum.

C

Q

24 - 1930

K₂SO₄

4593. Potassium salts. Kati - Chemist A.G.
FA. 682, 685, Sept. 5, 1929, MgO. ~~is~~
read. from crude K salts either
wholly or in part in the form
of phosphates of Mg and NH₄. FA.
682, 687 describes the recovery of
sulfates as K₂SO₄ by introducing
NH₃ into the solution freed from
phosphate of Mg.

25-1931

potash - yro

K₂CO₃

manuf of, P781², 04679

KCl - yro

K₂SO₄ - yro

6/6/50
500

25 - 1931. K_2CO_3 6/15/30

781. Potassium carbonate, alkaline (sol. anon.). Fr. 688, 493, Jan. 20, 1930. K_2CO_3 is made by introducing MgO into a soln. of a K salt under a pressure of CO_2 equal to or higher than the critical pressure at which CO_2 is sol. in all proportions in water. KCl is the preferred K salt so as to obtain $MgCl_2$ at the same time.

(

O

25-1931

K₂CO₃

11/12/31

4367. Potassium carbonate. Verein für chemische und metallurgische Produktion. Ber. 526, 322, Ser. 7, 1928. This relates to the manuf. of K₂CO₃ by treating a soln. of KCl and MgCl₂ with NH₃ and CO₂ in such a way as to form KHC₃O₃·MgCO₃·4H₂O and then decomposing the double salt. The improvement consists in using MgO, obtained from the double salt, to recover NH₃ from a corresponding amt. of the mother liquors from the mtn. of the double salt. These liquors contain NH₄Cl and some KCl, and when heated with MgO they yield NH₃ and a soln. contg. KCl and MgCl₂, which are used again. Cf. CA. 24, 1474.

26-1928 KCl 6/1/20
Z 560. Potassium chlorate and potassium chloride. E. I. Shpitalnik and Z. A. Loffe. Russ. 4354; granted Jan. 31, 1928; published July 31, 1928. Cl is made to react with K_2CO_3 soln. of such a concn. that finally KCl precipitates out at high temp. The first stage of the process in which K_2CO_3 is satd. with Cl_2 , whereby bicarbonate is produced and the max. concn. of the hypochlorite is reached, is carried out at a medium flow of Cl_2 and at a low temp.; the second stage is carried out at a gradually increasing temp. of the soln. and a simultaneously increasing velocity of the flow of Cl_2 so as to obtain an acidic soln., which favors the transfer of hypochlorite into the chlorate.

Imp

24-1930

K₂CO₃

2/10/30

4361. alkali chlorides are eliminated from solutions of crude K₂CO₃ by passing NH₃, preferably to saturation, into the soln. and decg. the ammoniacal layer which contains the chlorides.

26-1932

6/15/32

potash - yes

K₂CO₃ - yes

KCl

manuf. H, 25501

K₂SO₄ - yes

27-1933

potash - pro.

K₂CO₃

K₂CO₃, amount of, 4352?

K₂CO₃, amount of, 4359?

KCl - pro.

K₂SO₄

amount of, 51542, 5157?

27-1933

K₂CO₃

6/2/50

4357. Potassium bicarbonate. J. A. Farber

ind. A-G. Mit 386, 351, Jan. 19, 1933.

KCl reacts with CO₂, or with compts. of CO₂ and NH₃, in mists of NH₃ and H₂O, contg. at least sufficient H₂O to hydrolyze the carbonate to bicarbonate but more NH₃ than can be dissolved in the H₂O at room temp. and atmos. pressure, at temps. at which K carbonate would be decomposed under the conditions of working. In one example liquid NH₃, H₂O and powd. KCl are placed in a pressure vessel into which CO₂ at 10 atm. and room temp. is introduced.

in v. [In a 2nd a mist of liquid NH₃ and H₂O is led through powd. KCl and NH₄HCO₃ at 25 atm. and 60°]. KHCO₃ forms the solid phase in each case, NH₄Cl being contained in the mother liquor.

C

O

??

27-1933

K₂CO₃

2559 Potassium carbonate, Alfred Men-
 tel. Ser. 578, 471, Aug. 27, 1920. A
 mixt. of K₂SO₄, C and CaO is heated
 to about 1000° in N to give a pro-
 duct containing KCN, which is
 then hydrolyzed with steam at
 400-500° to produce K₂CO₃ and NH₃
 in accordance with the equation:

$$2KCN + 4H_2O = K_2CO_3 + CO + H_2 + NH_3$$

 added. NH₃ may be obtained by
 reconverting the hydrolysis pro-
 duct into cyanide and hydro-
 lyzing. The residue from the at-
 tention of the K₂CO₃ contains CaS, which
 may be treated in the known
 manner to yield CaCO₃ and H₂S.
 The latter may be oxidized to
 H₂SO₄ which may be combined
 with the NH₃.

propy.

27-1955 K-504 6/01/55
5134. Preparation of potassium sulfate
and ammonium chloride from
potassium chloride and ammonium
sulfate. K. Chirkov, N. V. Solov'ev
and M. S. Zokolova. *Kalii* (U.S.S.R.)
1955, no. 2, 19-24. — The most
complete transformation of KCl
into K₂SO₄ takes place at 700° and
with an excess of not less than
46.7% of (NH₄)₂SO₄.

C

Q

27-1953

K-504

11/10/53

5157. Potassium sulfate from potassium chloride. August 14th. 21.3. 1, 4.2, 4.8, Aug. 15. A reacting mix. of KCl and $(\text{NH}_4)_2\text{SO}_4$ is slowly passed through a zone heated at a temp. above the volatilization point of NH_4Cl and below the m.p. of K-504 and the resulting K-504 is continuously removed from the zone, the volatilized chloride is also continuously removed and is condensed, treated with lime, and the liberated NH_3 is absorbed in H_2SO_4 to prepare $(\text{NH}_4)_2\text{SO}_4$ for further continuation of the process. app. is described.

28 - 1954

potash - yr

6/6/50
7/1/50

K₂CO₃

manuf of, 1944-94

K₂ - yr

K₂SO₄

manuf of sulfate, 1934

K₂SO₄, manuf of, 1934-55?

28 - 1934 - K-504 6/15/76

4173. Potassium sulfate. Production
from potassium chloride and sulfur
acid. C. J. Fox and J. W. Threlentine,
Ind. Eng. Chem., 26, 493-6 (1934). -
Domestic economics and details
of a process.

28 - 1954

K₂CO₃

1149. Potassium carbonate and sodium sulfate. Koli. Chem. A-6. Fr. 784,517, Sept. 28, 1955. K₂CO₃ is transformed to K₂CO₃ by means of Na₂CO₃, and the Na₂CO₃ is retransformed to Na₂CO₃ by means of Na₂CO₃ in aqueous suspension, preferably under pressure and heat. The mother liquor from the Na₂CO₃ is treated with CO₂ to sep. NaHCO₃, and the mother liquor from the NaHCO₃ is cooled to sep. Na₂SO₄. The NaHCO₃ is heated in the presence of water to form Na₂CO₃ and H₂O.

28 - 1934

K₂SO₄

6255. Potassium sulfate, John A. S. Co.,
Yon, L. S. Clifford, Albert C. Case
more and Imperial Chemical Indus-
tries Ltd. Pat. 410, 330, May 25, 1934.

K₂SO₄ is made by the cyclic process:

(1) KCl, in part fresh material and
in part returned from (4), and
glaukite from (2) and (4), are
reacted in the presence of H₂O at

an elevated temp., e.g., 100°, K₂SO₄
being sep. after cooling to 5-55°;

(2) the mother liquor from (1) is
worked up by adding Na₂SO₄ and,
optionally, some KCl, to yield gla-
ukite at below 55° and a mother
liquor; (3) the mother liquor from

(2), after addn. of the mother liquor
from (4), is evapd. to sep. NaCl;

and (4) the mother liquor from (3)
is cooled to sep. KCl and glaukite
for return to (1) and a mother
liquor for return to (3).

29 - 1955

potash - geo.

K₂CO₃

system, K₂Fe(N)₆ - K₂SO₄ - H₂O - soln -
vilitis in, 82439

KCl - geo.

K₂SO₄

manif. of, P 37899

29-1935 K₂CO₃ 1/1/35
8246. The solubilities in the system po-
tassium ferrocyanide - potassium
carbonate - potassium sulfate - water
at 25°. N. A. Flückiger and E. F. Plas-
chia. Trans. State Inst. Applied
Chem. (U. S. S. R.) no. 26, 48-51. -
The solubilities are given in tables
and curves. With increasing concn.
of K₂CO₃ the solubilities of K₂CO₃ and
and K₄Fe(CN)₆ are greatly reduced.
at a concn. of 34.9% K₂CO₃ the concn.
of K₄Fe(CN)₆ falls to 0.269%, no
double point was obtained at which
the 2 salts are in solid phase.

C

O

29-1935

K-504

11/17/35

3799. Potassium sulfate, Palestine PFluch
Std. Fr. 776, 937, Feb. 7, 1935. Finely
divided. sol. anhydride or semi-
hydrated CaSO₄ is agitated care-
fully for a short time with a solution
of KCl. The syngenite is sep. from
the soln. of Ca SO₄ and the remaining
KCl and decomposed to K₂SO₄.

prob!

30-1936

potash - ver.

K₂CO₃ - ver.

KCl

rem. from Kaid Na salt, 77897.

K₂SO₄

6/1/50
J.H.

6/6/50
mp

Browning

Date Received 11/6/50

Name of Contributor Willie Smith

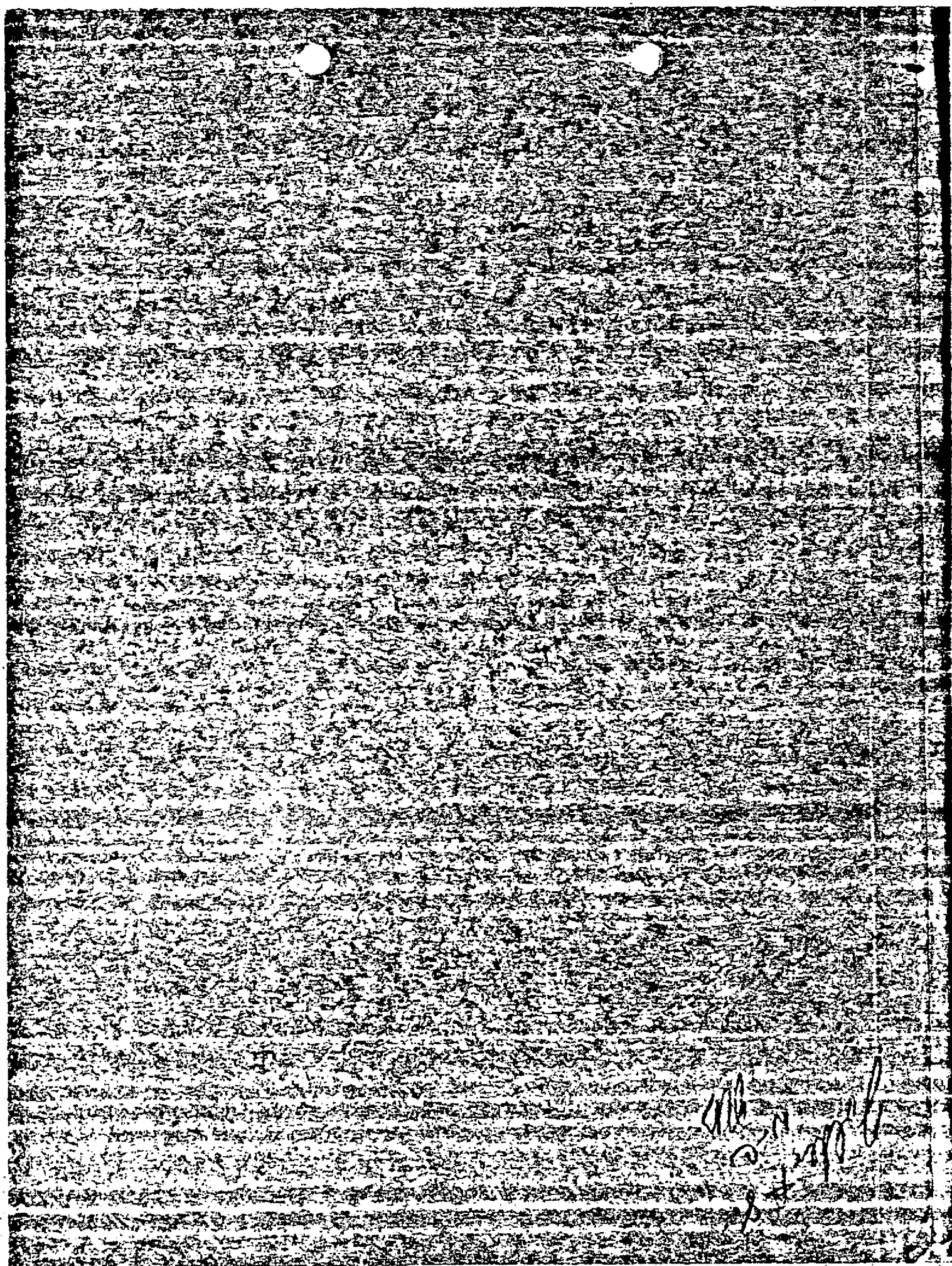
To Be Paid to Willie Smith

Description of Gift 8 of antiques

Box No. 63-4307-1-B-12 (1)

63-4307-1-B-12 (1)

63-4307-1-B-12(1) #8



SAC, PHILADELPHIA

July 7, 1950

T. SCOTT MILLER, SA

HARRY GOLD, was.,
ESP - R

65-1307-1B 12 (1) Folder No. 8

GOLD identified the material in this folder on June 24, 1950.

The letter dated September 15, 1947 from JULIAN PAUL BRODIE to BROTHMAN, GOLD said, was in connection with some work that BROTHMAN's firm was doing for BRODIE which was an attempt to develop a men's vanishing cream. GOLD explained that the firm was having difficulty in making the cream with the qualities that BRODIE desired the cream to have, such as a medicinal effect on the face as well as leaving the face with a definite tanned appearance. GOLD said this letter was written by BRODIE after it was apparent by both parties that the vanishing cream could not be produced.

GOLD said that the remainder of the material in this folder dealt with his work at the Philadelphia General Hospital and his course in pharmacology at the U. of Pa. Medical School.

TSM:EMC
65-1307

05/11/49
JL

Quantity	Master Cat. No.	Description
1	H-28900	Beckman pH meter, model G; complete with batteries, electrodes, solutions, standard cell and amp.
2	H-29040	Hypodermic Glass Electrodes
1	H-14200	Klett - Summerson Colorimeter, clinical model Two H-1420 Klett - Summerson Filters, nos. K-5 54 and K-5 64 are included
1	H-10640	Blood Volume Index Centrifuge Tubes
2	H-6350	Koch micro - Burets; standard Taper points, amp interchangeable outlet tips — capacity, 5 ml.
1	H-34640	Volumetric Flasks; standard Taper glass stoppered — capacity, 100 ml.

Price

~~245.00~~
245.00

~~245.00~~ 30.00

148.00

~~10.48~~

~~29.32~~

~~78.62~~

9/10/48

materials list for Heart Station (continued)

Quantity	Measure	Description	Price
1	each	10-18461	
		cylinders of standard type slugs covered, ones shown — 10 ml capacity	\$ 10.00

Insert p. 10

Note: The above ~~are~~ all to be ordered from The Harsco Scientific Company, 117 S. 17th St., Phila., Pa. They are standard laboratory items which sell at a fixed price, regardless of the supply house from which they are ordered.

as per 1/4

In addition, there ~~are~~ needed:

Filing cabinet

Table, solidly built, size: length, 65", width 35", height, 30"
 ~~various~~

Sink - the present sink should be raised to a height of 36" (from floor to top), and the fittings replaced with an acid-proof type.



Shelf are needed, in various locations in the room

03/10/79

Quantity

Hardware
cat no.
6404

Y

Description

Vandiyke monomeric
alloy steel apparatus

Price

2.48

6 Volt, 120 amp-hr.
storage battery for use
with Beckman model
DU electrophotometer

Trickle charger for use
with 6 Volt, 120 amp-hr.
storage battery

these materials

other ~~from~~ supply houses ~~from~~ where they may be
obtained are:

Arthur H. Thomas Co.
230 S. 7th St.
Philadelphia, Pa.

Williams, Brown & Carl
919 Market St.
Philadelphia, Pa.

material for boat station (optional)

price

6 40

5 12

2 64

13 32

6 60

16 25

distillation

plant making

capacity, 20 mt.

double catalytic

plant capacity

with plant

capacity, 100 mt.

distilling plant

plant capacity,

filter

fast tubes, Pyrex

15 mm diameter

125 mm length

cost no.

H-3200

H-11090

H-6110

H-6280

H-60940

H-61610

1/15/57

Chemical

2

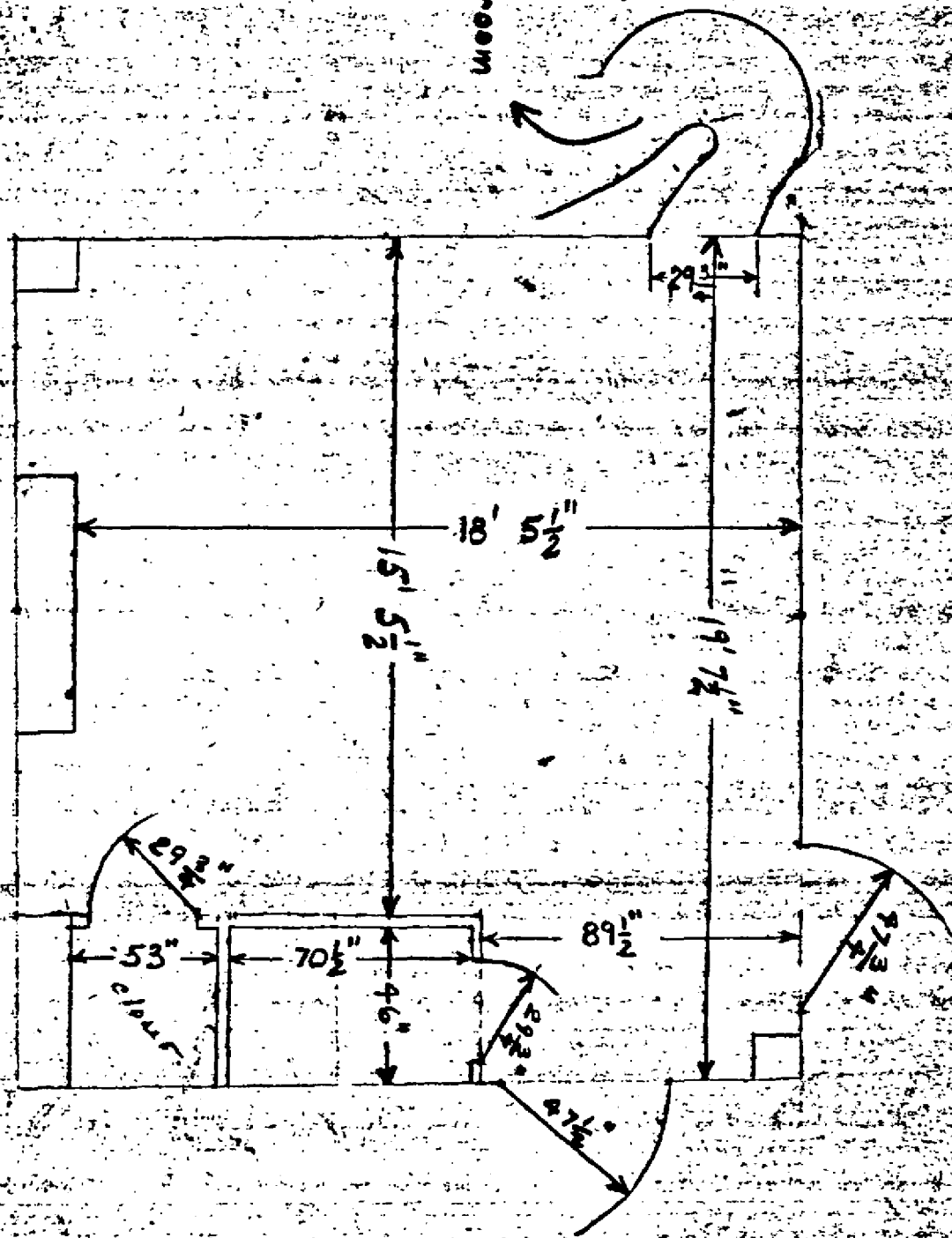
2

2

1

26

Darkroom



HEART STATION LABORATORY
SCALE 1/4" = 1 FOOT 10-19-48

6/6/50

$$\frac{0.3}{1}$$

2015

2015

$$\frac{1.5}{3} = 0.5$$

1.3

$$\frac{16}{5} = 3.2$$

$$\frac{3.6}{6.0} = 0.6$$

~~2015~~

$$\frac{10}{2} = 5$$

②

$$\begin{array}{r} 14\sqrt{} \\ 16.0 \\ \underline{3.0} \\ 19.0 \end{array}$$

2.7

0.207 mg / 10 cc
2.7 mg / 100 cc

$$3.6 \times \frac{1.5}{5} = 1.08$$

2.7

6/6/50

B.V.
B.V.
9.5

1260
8.7

f-o-c. 132 701-789 (1220)

am. f. rhine 127 420-436 (1234)

f-o-c. 151 464-477 (1244)

am. f. rhine 130 464-470
" " 129 474-478

am. f. rhine 132 801-809

determination of Na and K in serum from coma Project Patient

patient - Ralph Parker

6/6/50
JW

Potassium,

Sodium,

mg %

mg %

1	24.5 (6.30 mcg.)	345 (mcg.)
2	24.5 (3.70 mcg.)	360 (mcg.)
3	12.5 (mcg.)	355 (mcg.)
4	12.5 (mcg.)	370 (mcg.)

When used Pentothal at resuscitating facilities
to use cyclo-oxygenase
release - autonomic sensation

6/6/50
W

also, within and cyclo-oxygenase at bad heart action
to use curare & can use low cyclo-oxygenase levels

Summary Table of Effects

II Stimulants of various Cortex (only C.N.S.)

In essence convulsant drugs = analeptics
contrast depression of C.N.S. → air

P.B. Cholinergic

Caffine - the M & Vanthine derivative

atropine
+
curare } formerly popular

all of above almost entirely supplanted by
synthetic or other drugs - hardly ever used except
a severe atropine &

are

Picrotoxin - no M

poorly understood

no clear formula known

↳ natural

counteracts
poison by
bituminates

not effective against chloral hydrate &
other like drugs

are intravenously

↳ effects

slow infusion

↳ because of latent
poison in onset
of picrotoxin effect

⑧

11-1-47

metrazole & does not have latent period - was
introd as cardiac stimulant (m?)

1. used as antidote for depressants
2. notes of an an. facilitates

6/6/50
JH

Coramine (Nikethamide)

Parent of this group when the more popular

active base of action is presumably bound up
with narcotic drug action

- 1. might displace narcotic drug from ^{nerve} ~~nerve~~
- 2. " ventricular " chemically

but 1 & 2 are not true

3. action is a physiological one in which
nerve cell is acted on & a contrary action to
effect of the narcotic drug.

4. or, indirectly, by setting up strong reflexes —
nerve impulses may be aroused — a.c.
cf. cyanides. - reflexes in carotid & aortic
bodies.

most Recent Trend (analeptic action)

not only stimulate C.M.S. but \rightarrow > 4 p.
(increase circulation)

Sympathomimetic Effects

mimic effects of —?

Adrenaline, etc.

get
improvement
in
circulation

Habitual Use of Drugs

11-1-48

1/6/50
29

Simple Habituation } etc.
Physical Dependence }

addiction is a psychic affection Man - cannot
be studied in animals.

↓
disposition
chronic poisoning
with tolerance
not true def.

Dog - can not handle a dose it likes
✓ with a first but gets accustomed - yet needs
↑
morphine can be trained as to effects.

Cocaine - Dogs do not like it
can make dogs tolerant to morphine but never addicted

Tolerance - can really get to be something

whether
RbH
or Sch.

↓
100 mg morphine all day
subcutaneous only
1.5 am morphine in 40 min. I.V.
(reducing fatal dose 0.5 am).

is essentially a tissue tolerance - cells of
body have become accustomed to the presence of
drug - possibly oxidized more readily - fate
of morphine same in man.

in place of
drug

Contract Effect -

perfuse heart with digitalis (a toxin) & continuing
→ contract but if continued perfusion, heart no longer

Exp.

6/10/20

action of drug depends on diff in conc. bet.
inside & outside of cell. (i.e. - drug diffuses out)

→ Reason why addicts never take off drug
sometimes dr. when starts again.

Heroin - Diacetylmorphine - not
supposed to produce addiction - but really -
worst of all known

Every one of artificial morphine derivatives is known
to cause addiction

Cyph. - only ~~one~~ ^{partially} useful

Colony (U.S.P.H.S.) Tex. Ky.

1. Withdraw drug → abstinence symptoms
↓
"cold turkey" - shivers
nausea
convulsions
collapse

Not small amount of morphine → & symptoms
↓
disposition

→ this way to test drugs for addiction (i.e.
Demerol or methadone)

Drastic test but there is no other way at
present

- a - can't study animals
- b - can't try on population
- c - to utilize colony (to test new synthetic
analgesics)

↓
now found which is
addiction producing

type no. could be made

next

circumstances used for effect on nervous system (as with these)

1. historically remembered ones (Claude Bernard)

6/6/50
20

2. List of things on nervous system.

- a. local anesthetic
- b. sedatives

- c. convulsant & anti-convulsants
- d. analgesics

3. also - some which produce bad effects on nervous system (e.g. -
carbamazepine
strychnine)

4. + series for action on C.N.S. leads to examining the outcome
over background of pharmacology

II.

1. General

A. Inhalation

1. Etvo - Entenmil 1948

- profoundly & completely unconscious
- muscle completely relaxed
- no conscious or unconscious movement
- but, do not stop breathing or circulation

Surgical anaesthesia

so too far -> paralysis of C.N.S. -> death

Etvo (etc) selectively depresses C.N.S. (mouth
vent)

also, selective within nervous system

2. Goals

a. diminished consciousness & drowsiness - can
still perform

b. stage of excitement & delirium - partly conscious
or unconscious - good deal of irritability

(remember about Crawford long duration effect)
to end of 4th P. red blood 1840

(first authentic case of intracerebral anaesthesia)

c. surgical anaesthesia - all symptoms quiet down
& patient is breathless much less - muscles completely

is and depression depression of C.N.S.

- (1) + most higher centers are involved in - memory, thinking, etc.
- (2) next, lower centers (primitive part of CNS.)
- (3) paralysis of spinal cord \rightarrow muscular relaxation
- (4) most primitive part of CNS. - breathing slows down

I Why is the drug selective on CNS?

6/6/50
20

II Why a descending depression?

It should have been studied more than any other drugs and anesthetics

1. Reaction is not due to a chemical reaction but drug + a. sensitivity of tissues - no common chemical decomposition
amides, etc., organic of amine, etc.,
etc., etc., amine, etc., etc.

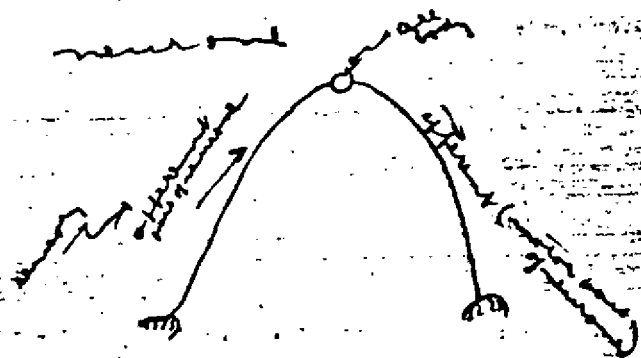
2. prompt & ready reversibility ~~characteristic~~ \rightarrow reversible

3. sensitivity of tissues in which reacts could be varied
could produce on bacteria, plants, sea animals, etc.
streaming, contraction of heart of dog - can work on
fermentation of sugar (147, E to H) + all of above

4. Degree of effect depends on concentration of anesthetic agent

ordinarily non-irritation CNS - as first worked on spinal
living animals.

\hookrightarrow Fundamental organs of CNS



as transfer of impulse across nerve membrane

\hookrightarrow impulse (not continuity like nerves)

1. impulse most susceptible part of nerve cell
and also most susceptible in all body

① Action of Drugs on Circulation

11-15-49

Reasons

1. Disturbances of C.S. not common which M.D. deals
2. Not variety & no of drugs used
3. Tremendous mass of info available (cannot be measured objectively without difficulty)

Basic underlying principles

1. Envir of human being really consists of sea water environment around cells (cf pericardial cavity)

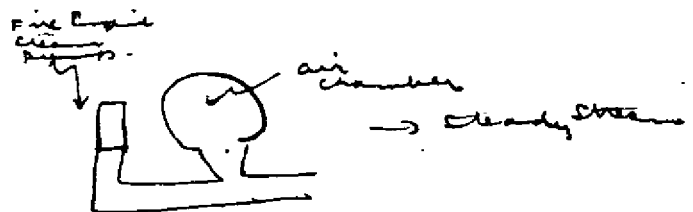
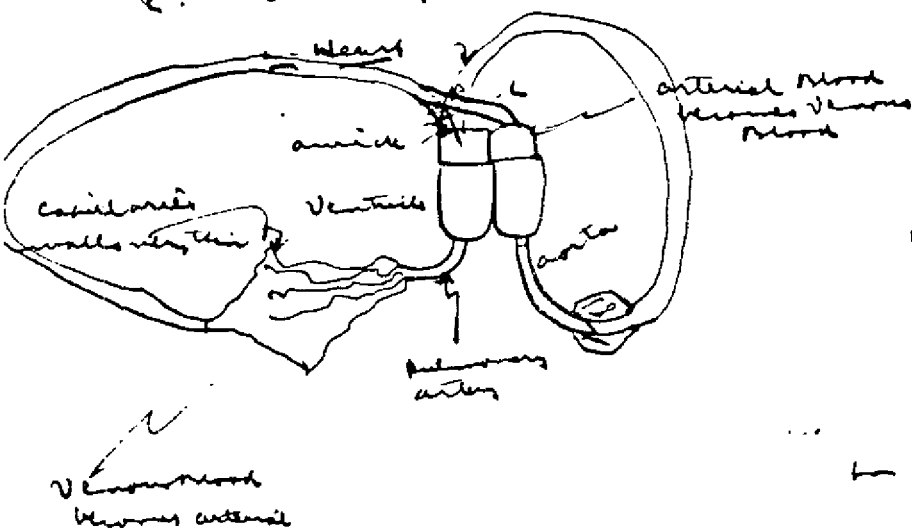
Diffs

- a. can measure
- b. red cells
- c. oxygen reserve
- d. can remove waste of metabolism
- e. white cells for dealing with infection.

Can study

- a. blood level & time administration of drugs — cf. myocardial

- b. are dealing with complex tubing system



↳ blood vessels have elasticity of blood vessels.

- a. can't feel directly — must feel definite sequence of observations
- b. must feel constant output — otherwise at edema, anastomosis
- ↳ outputs of both ventricles must be the same

3. Have pressure in tubes (major factor in controlling circulation)

- a. cardiac output of heart
- b. resistance of tubing system (cut down outflow) ↳ causes
- c. viscosity of liquid (↳ viscosity & red matter present)

- ⑤
1. only impulse is carried into a muscle
 2. rest where nerve fiber is (high resistance & no flow)
 3. ?

6/6/50
720

more cells involved in higher parts of CNS.
also more synapses involved
∴ the greater no. of units involved, the greater the
sensitivity to narcotics.

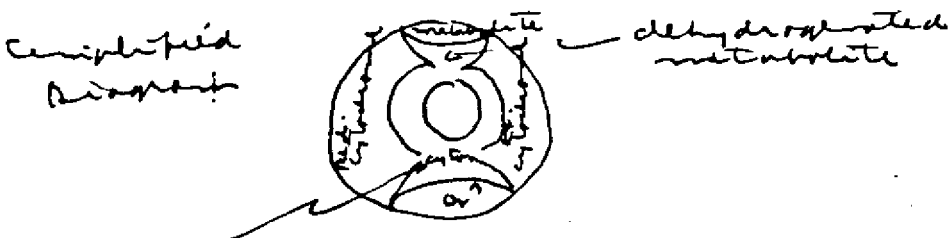
Theories of narcosis (anesthesia)

1. no all-inclusive physical-chemical explanation
2. now look upon drugs as inhibitors (relatively non-toxic)
of cell excitations

∴ at less free energy available for rest. cell functions
also ∴ at unstable state of excitability.

Recent work shows that // similar to state of
brain & degree of functional activity of brain

Keilin's exp. of oxidation in cell. - see diagram



cytoplasm oxidase - low CN works

on a small
complete snow ball or blanketing of cell - then not
on dehydrogenated of agent

So Summary

1. narcotics act on dehydrogenation of cell & cut down
on uptake
2. not only selectivity on CNS? id. hi
a. high or requirement of CNS - brain req
b. brain more susceptible to lack of O₂ (molecular O₂)
from bloodstream
c. brain probably contains more cells than rest of body
put together
d. abundant blood supply of CNS - brain gets
more blood than any other body tissue except
kidney

Relation of Denervation Facilitation due to decreased or ~~the~~ uptake

muscle - nerve all as (chemical) amplifier

Ex acetylcholine cycle.

↓
chemical mediator

a.c.h. - liberated at periphery of nerve.

nerve impulse → K ions

↓
(impulse)
disturbance

(a.c.h. is stored but is
sensitive to K ions)

a.c.h. cell receptor substance is choline esterase

|
|

↓ free energy contrib by destruction of a.c.h.

But must replace a.c.h.

choline - from phospholipids in all body tissues

acetyl
oxidation
pyruvate $\xrightarrow{O_2}$ acetate
CO₂
acetyl-CoA

or
from
citrate or
hydrolysis of
acetoacetate

need also tri-energy phosphate (for comb of
choline + acetate) → a.c.h.

or for oxidation of pyruvate

a.c.h. can be considered as prototype of
unstable chemical cnds which transmit neural
impulse & amplify
read
unstable chemical cnds

Consideration of Cereb. used in Producing anesthesia A- (minor)

Survivors of 1000 = three

6/6/50
20

→ Reason = most of 1000s had undesirable side effects

- a. bad odor or taste
- b. marked depression of heart
- c. bad effect on liver or kidney
- d. cannot control degree of anesthesia produced

- A- only results that can be taken by intubation
can be used to produce surg. aneth. —
reason, can control rate of absorp. & elimin. by
~~breathing~~ (ways of lungs)
- ↳ { control (inhalant) gas in intubated air
 - { control in alveolar (exhaled) air
- ∴, have complete control by operator
- 3. Signs & symptoms should be very readily
recognized by anesthetist
 - 4. current trend is to use combination
 - 5. expect semi anesthetic in E-t-o — use others
because of disagreeability & slowly after feeling

B. never Intrac. Anest.

- 1. pleasant to take — just prick
 - 2. But do not block perception of pain — at all
not analgesic
- so as surgeon works patient may make movement
& anesthetist will move →
- 1. Removal of pain stimulus
 - 2. Anesthetist may → respiratory failure
- 3. Anesthesia is nil for patient — aneth. by
reflex — patient falls asleep in paralyzed state
↓
not even needle prick

But now we shall do → unconscious
then use nitrous oxide to complete

(6)

10-25-69

Cy clopropane now favorite of anesthetists

- (1) use 2.0% in O₂ (lighter than air)
- (2) Has made thoracic surgery feasible.

6/6/50
JP

D. Simple Sedatives - to produce sleep

(1) action like 1st stage of surgical anesthesia

assume all act in same way (more confusion of
significance than explanation)

Cont

- a. Expose brain (light ventral)
- b. can with no stimuli (elect. responses) at dose of
diff. areas in cortex
- c. > ventral & dorsal areas
- d. But Et = 0 & all responses - all complete

But

chloralose (p. 5) & can pick up impulses through
C.N.S.

Simple Sedatives 1,000's tested

Derivatives of barbituric acid have practically
superseded all other classes of drugs.

Reasons

- 1. Very low toxicity - do not damage vit. analogy
not or cause damage to
kidney or heart
- 2. Easy to take - capsule
- 3. Variety of choice - partial with respect to duration
of action
long acting - not if want to wake
unrehearsed
& for anesthesia etc.
- 4. 6 hrs for removal of intestines
- 5. action short - almost exclusively of
toxic convulsions etc.

10-25-48

p. 3

2

all - conspicuous for producing state of ^{accompanied} euphoria -
 cases go away, feel good
 also produce considerable ~~than~~ amt. of dilation of
 blood vessels - feel warmer at time.
 But ^{the} lowers body temp & (if taken within) then
 get pneumonia.
 worse of all - habituation - most conspicuous of
 all mentioned so far
 Only few are seriously addicted to barbiturates
 Check now for better info as how these things work
 & not for "ideal" derivative.

Crude

1. Truth serum - Pentothal → depresses higher centers
 but not with dose individual concludes more
 freely. also used in mental hospitals for
 cases of schizophrenia & dementia praecox
 → small dose → quite lucid (as release of
 internal tension)
2. Action of serum in body (Partic barbiturates)
 a. Excrete 80-90% of drug acting barb. unchanged
 in urine (Pentobarbital, etc).
 & rest of it shorter acting than at home
 coming out in excreta - what happens?
 Infectious Hepatitis?? Look up.
 associated in liver
- c. But shorter acting ones - they are some
 in animals & humans. - presumably
 not detoxified in liver etc
- d. now excretion S^R - Columbia
- e. excretion studies - for di-dilum

p. 5

Anti Convulsants

convulsions - excitation of C.N.S. has spread beyond
 bounds - all stimulation gone on with board -
 overall rings all bells & diversified react.
 of all body muscles

Intake Dose of NaCN in Dog - goes thru all
 symptoms of ~~initial~~ typical epileptic (high/low
 day)

all ~~over~~ ^{derivate} ~~carpids~~ ^{carpids} ~~action~~ ^{action} bodies - so cannot
 also ~~least~~ ^{least} ~~and~~ ^{and} ~~nothing~~ ^{nothing} happens

on other hand (

Convulsion may be due to increased intensity of stimulus (over-decrease to spread of stimulus) (cause of)

now, How prevent convulsion

6/6/50
21

E p.s. when convulsions are already in (i.e., tetanus, etc.)

note the - latter effects of barbiturates
3 orders to 3 mins in longer acting
barbiturates in n.w.
pentothal - short acting
phenobarb - long acting

For chronic use - epilepsy - phenobarbital
disadvantage -> drowsiness
bromides not used much due to ->
bromide psychosis.

most recent ones

D.H.S. (Dilantin Hydantoin etc)

Tridione (most recent)

phenitoin - convulsions drug.

now prove dose (also) in brain of animal (also)

set threshold dose (current)

then -> drug

of try current

-> "screening" test - keep only promising

try again & discard

try again & discard

-> 1 remaining

Dilantin - mental & psychomotor n.s.

did not make individual study

reflexes - normal

silencing - major conclusion

tridial - minor (some saving of comms.)

ver 1. habitual use of signs

6/6/50
JP

and may interfere with circulation of

blood vessels (widely)

also, in some individuals carotid sinus is sensitive of external pressure
minutes
Trolley car accident in Boston

6/6/50
JMM

I B 2 carotid & aortic bodies close to carotid sinus & aortic arch or base
(like autonomic sensory roots)
hemorrhoidal organisms can be stimulated by (a) (b) & (c)
(b) only reserpine & amesophylline act as so in monkey -
will remove aortic to circulation time - takes deep
breath suddenly - reserpine least likely of (a) (b) & (c)
to do harm - "drug of choice"

3. 4. Experiments
→ reserpine doses of nicotine to animals & induces
h.p. initially, then continues to rise, then falls,
finally → paralysis.

Drug for working transfer similar to am & reserpine found in
Tetrahymena & etc. C. d. m. m.

4. Committee on the Control of Endings (do they really exist or not)
produced by ergot group of alkaloids
new drugs dihydroergot (ergot) for hypertension
early returns of. now being tried
dihydroamine favorable

5. muscarinic & muscarinic drugs → vasodilatation
& muscarinic action is removed by action of atropine
Atropine - drug the antidote which does not touch
muscle to relieve death → to remove effects
(swelling, dilation of heart muscles etc.)

6. 1. Vomiting due to distal action of local drug
also loss of appetite (first reaction)
2. Veratrum alkaloids used for h.p. for use by
reserpine in heart itself for (a) & (b) — (c) in lungs.

7. circulation unobstructed to muscular activity - now unimp
then is first reflexed out up - (how much)
8. Drug action on ?

S.N.S. when stimulated is supposed to liberate or form
sympathetic (a chemical mediator) \bigcirc 6/6/57

Heart beat - Heart has a rhythm of its own - i.e., does
autonomous
not depend on nerve impulses. (e.g. - see heart of frog
& pump along time)

sympathetic
nerve
acceleration



Sinoatrial node

impulse begins at sinoatrial node

no communication bet. muscles of atricle &
ventricle so impulse would die out
if not for AV node

atrio-ventricular node

interval $\frac{1}{10}$ sec
(no longer than $\frac{3}{10}$ sec
if so fast heart stops)

fine muscles
spread all over
& whole ventricle
contract

(i.e., no left bundle branch
block)

Inhibitory nerve - X (Vagus nerve)

so, have four extra cardiac pairs of nerves to regulate action output of
heart by blood pressure

at { normal output - 4000 cc/min
blood pressure - 120 mm Hg for top level (systolic)
(70) 80 mm Hg - base level (diastolic)

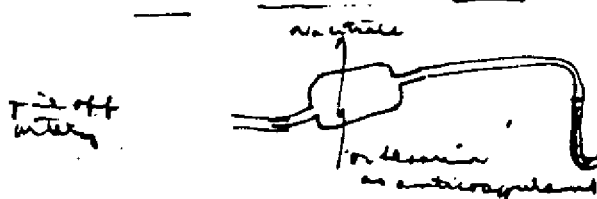
min 20.0 - 8 to 10 liters/min

B.P. { syst. - can go to 200

baroreceptors
(pressure mechanisms can
operate as regulatory mechanism)

normal - 90 mm Hg
affected by


How measure B.P. - animal



Kymograph drum

Must do not get both systolic & diastolic b.p.
and delay

6/6/50

 I had two skins & amputated suitably
(deceased during war)
do not move much (but amputated) &
so had only little thought of only a
few drops of blood & used only inner
a small needle (also set on clothing)

Print of test wire depends on cross section
m-g.

D. unis inplatable cap + 2 claws.
parts on ^{1st} joint press to obliterate pulv in arm - when 2d
sees round (septohipres) then enters longer for rest →
disaster
air press falls

the
when and six around gets down down & then another

can do on rats tail - don't listen of water circulation
under microscope.

1. Drug actions on almost stuff - $>$ or $<$ activity } next with
v. m. " • Tones Caliber of Blood Vessels
3. " " • Velocity of blood - no aims. blood action that are recognized - ability of SCN to look b.p. when sleep.

11. Finally suggest why act on mechanism of control & regulation -
 which act on ~~neurotransmitter~~ ^{neurotransmitter} mechanism that
 control circulation.

6/4/50
 20

I-A. outline

1. Cardioinhibitory center stimulus - digitalis & drugs
 p. 2 I-A-1

other drugs which produce this are those ^{vertebrate} ~~vertebrate~~ ^{nerve} ~~nerve~~
 produces marked slowing of heart, dilation of blood vessels &
 ? respiration.

C.C.D. not

r. decreased interaction bet cardioaccelerators & cardioinhibi
 centers - most marked by action of nerves → > decreased
 dilation & > cardiac output

I-A.

2. V.C.S & 3

induced not breaking of quiescence etc & b.p. rises then
 this is favorable sign - getting ready to pump. But not
 true for benzylamine type - raises b.p. by peripheral type action
 vasoconstrictor after being stimulated.

V.C. - no. imp - better is control

B In the Blood Vessels

1. when b.p. rises → inhibitory impulses (in ?)
 a. slow heart
 b. dilate blood vessels
 c. inhibit respiration

↳ ^{inhibitory} ~~inhibitory~~ ^{terminated} ~~terminated~~
 ↳ receptors and
 optic nerve conduction

4
 Del intervention in experimental animal

b.p. → 200
 fibrillation - valves fail completely
 pulse gone 20 → 200

AND is that all way to study high b.p. without arterial banding
 and on eq. → other mechanisms coming into play & b.p. - low

Control of Circulation

1. Identify the controlling circula - should maintain homeostasis (control all internal environment)

mainly about by

a. rate output of heart

b. size caliber of blood vessels

} but really can't do this for long.

Compensate - cardiac output & vascular tone are regulated by two main types of influences.

a. chemical influences

b. Vaso motor influences

6/6/50
20

Chemical - blood vessels of all parts of body are dilated by all kinds of metabolism

a. \rightarrow O₂ tension

b. \rightarrow CO₂

c. \rightarrow acidity

d. \rightarrow t¹

e. acetylcholine

f. K ions

etc.

So, the more active a cell becomes the more the blood vessels dilate, i.e. local automatic control.

Vaso motor influence - diff than chemical

a. Vasoconstrictor

b. Vasodilator (belong to ~~sympathetic~~ ^{parasympathetic} nervous system)

Relay to ~~sympathetic~~ ^{parasympathetic} nervous system

(initial factor in warning body for emergencies)

So involve whole S.N.S.

So, some blood vessels in body that are already constricted by S.N.S. i.e. some that are not (e.g. blood vessels supplying abdomen, skin)

But not brain & heart - fear for flight

new blood

muscles,

don't need now (already in S.N.S.)

cf. action of adrenaline & epinephrine

Vasodilator control - Relay to parasympathetic nervous system (in the main antagonistic to S.N.S. in its action)

usually for function of individual isolated organs (e.g. salivary glands, etc.)

when stimulated

usually

is via To adrenergic

2-15-47

Na & K by Flame Photometer (perkin-elmer)

4/4/50
20

A. Blood Prep

1. Mix 100
2. Centrifuge → Serum

B. Std

1. Na with K
2. w/ K with Na
3. w/ TCA

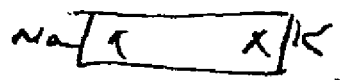
C. Sample



1. 4cc TCA (10%)
 2. add 2 cc sample (undiluted) serum
 3. add 4 cc distilled H₂O (dist) (water)
 4. Fill
- What 40 or 42



(5) K stds 4, 5 & 6.



X X
Dist H₂O

5. (as per dist H₂O (machine) - only Na with K)
6. Remove F.A. - Remove funnel
7. Turn dial 0
8. for start set to 0 with water knob K
9. Run 1st one (H₂O in dist) then each std & sample
It should agree with std

4 am / 100

1 20

39 / 100

3000 / 100

part.

(2) 4 4

24

$\frac{1}{2} + 9 \frac{1}{2} 1404$

Urine

1-100 mi

Tag v the window

6/6/50
20

Protein
(serum)

ask the technician to use a colorimeter
(10.1 cc serum) (0.1 cc reagent)
run up & down the tube
read at 500 mμ
Make it using same rubber stress for
all tubes
centrifuge 5 min

6/6/50

Klett-Sumerson Photochrometer

use 540 mμ filter (adjust to zero for each filter)

use Klett colorimeter tubes (individually calibrated)

run blank 1st then samples.

Rinse tubes 1st in H₂O water then dry & read

Table for values

Kjeldahl serum of known values

$$\frac{1000}{50}$$

$$\frac{1000}{50}$$

(

~~Handwritten scribbles~~

4/6/50
70

(0)

$$\frac{1000}{100}$$

→ 5.0

$$\frac{10}{20}$$

~~Handwritten scribbles~~

$$\frac{10}{20}$$

2.5

2

$$\frac{10}{20} \times 5 = 2.5$$

$$\frac{10}{20} \times 5 = 2.0$$

$$\frac{10}{20} \times 5 = 2.5$$

$$\frac{10}{20} \times 5 = 3.0$$

$$\frac{10}{20} \times 5 = 2.5$$

$$\frac{10}{20} \times 5 = 4.0$$

$$\frac{10}{20} \times 5 = 2.0$$

$$\frac{10}{20} \times 5 = 2.0$$

$$\frac{10}{20} \times 5 = 1.0$$

60 → 4.20

10 → 5.0

10

60 → 4.20

$$\frac{1000}{20} \rightarrow 50$$

C.A. 41, 2150^a (1947)

J. Biol. Chem. 167, 499-513 (1947)

6/6/50
210

Pauline M. Wald

The Flame Photometer for the measurement of
Na and K in Biol. materials

also,

C.A. 40, 4756¹ (1946)

J. Biol. Chem. 163, 429-434 (1946)

P. 24. H.

notes on chem. detn of Na & K.

C.A. 41, 6595^c (1947)

J. Biol. Chem. 168, 641-649 (1947)

R. Overman & A. K. Davis (this Journ. coll. med.)

Application of Flame Photometry to Na & K detns in
Biological fluids

O & D model Perkin - Elmer flame photometer
model 17 for detn of Na & K in blood, plasma
red blood cells & urine with approx. the same
accuracy as by chemical methods.

C.A. 41, 2767^f (1947)

see 0.0005 mg/ml.

Biochem. J. 40, 828-831 (1947)

R. J. Sumner (Postgrad. med. school, London)

colorimetric detn of Mg in plasma or urine by means of
Titan yellow

29-8

(

9.5

)

2.6

253 x 1.7

$$\begin{array}{r} 7.0 \\ 3 \overline{) 450} \\ 8.3 \end{array}$$

$$\frac{190}{2}$$

$$\frac{25}{1} = 25$$

$$\frac{75}{1} = 75$$

$$\begin{array}{r} 2.5 \\ 2.5 \\ 1.5 \\ 3 \overline{) 6.5} \\ 2.1 \end{array}$$

$$\begin{array}{r} 5.0 \\ 3.0 \\ 6.0 \\ 3 \overline{) 190} \\ 6.3 \end{array} \quad 6.5$$

6/6/50

$$\begin{array}{r} 7.0 \\ 6.5 \\ 6.5 \\ 3 \overline{) 10.0} \\ 6.6 \end{array}$$

$$\begin{array}{r} 3.0 \\ 2.0 \\ 1.5 \\ 3 \overline{) 6.4} \\ 2.1 \end{array}$$

$$\begin{array}{r} 6.0 \\ 4.5 \\ 4.5 \\ 3 \overline{) 16.0} \\ 5.3 \end{array}$$

$$\begin{array}{r} 9.5 \\ 5 \\ \hline 47.5 \end{array}$$

11.0

$$\begin{array}{r} 1.0 \\ 1.0 \\ 0.0 \\ 2 \overline{) 2.0} \\ 0.0 \end{array}$$

$$\frac{7.5}{4} = 1.875$$

$$\begin{array}{r} 7.5 \\ 7.5 \\ 7.5 \\ 3 \overline{) 22.5} \\ 7.5 \end{array}$$

19.0 x 5

75

2nd conclusion ~~less~~ ^{more} sensory fibers ~~conduct~~ ^{conduct} than motor fibers (afferent)

Counterpart - leg goes to sleep - can move leg but can't feel any

also Further

6/6/50 20

being knocked out before pressure applied by anesthetic - or reason smaller fibers (matter of diffusion)

tested local anesthetic tested on p. 5

Citric acid like barbiturates - adequate but will not likely get ideal

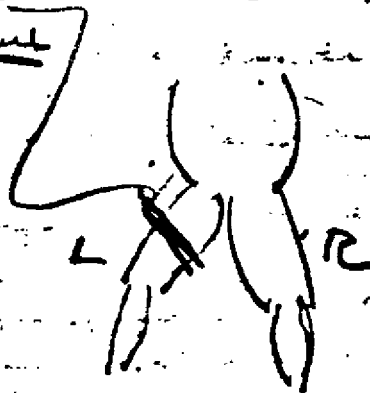
note p. 5 - Skeletal neuromuscular blocking agents

muscle relax but need to breathe

Curare or curare-like action

Famous test of Claude Bernard

- 1. Held Curare to Frog - Respiratory Paralysis & hypoxia
- 2. Application on 5th rib in intercostal space
- 3. put rubber band around trachea to obstruct ventilation before curare



aria reflex L R
2 2'
after curare 2 0

control response in reflex

Curare - produces paralysis but is not anesthetic for "pith" animal & then cut down & study all the cells

protected side still reacts normally

a. General (U.S.A) not active (the) Decision to discontinue
 but? ~~non-therapeutic~~
 major danger of addiction
 Ad. part ?? → looks effective smooth muscle.

6/6/50
 JH

b. amide done
 most effective synthetic analgesic
 most ~~strong~~ effective analgesic yet free from danger of addiction

over 10.4

(miscellaneous depressants)

1. Carbocaine (depress. hypoglycemic)
 ↳ has a strong like action on autonomic system
 we can depress C.V.S. & produce sleep
 ↳ rather slow to work
 1 day as time - mild pleasant N₂-richness & air-richness
 (found out during war)
 but results largely due to suggestion? to definite
 ability in h. to perception of air-richness.
 also used as truth serum.

2. amorphous H₂O - solid out H₂O from machine with
 cones and
 cones
 a. ~~has~~ something effect to some (only 15 mg)
 v. but has not removed the depressant effect - (air) can
 still drunk

3. Butobarbital - produces convulsions (not animal in
 any position)
 used as qd sedative for horses (machine actually
 excites)

4. my Cull - animal doesn't matter
 as long as dose → respiratory failure
 (but must inject - can't find other indication
 would it reason for conc. effect of amorphous)

can produce intense burning by O₂
 (small can be slightly in reaction)

If normal defense mechanism break down

p. 5. Contraindications

(1) no therapeutic effect

6/6/50
JLD

P. 5. Local anesthetics

- produce insensibility to pain at site which applied
 (not like gen. anesthetics which are carried thru body)
 - act locally because they are applied locally (e.g. numb area)
 - Have mitigation for nerve tissue.
 - Do not use to produce gen. anesthesia
 - (1) not convulsions \rightarrow collapse
 - (2) collapse without any relaxation convulsions.
- Do not want these substances absorbed.

How used

1. terminal anesthesia - usually to numb constriction and
 be constriction relaxation of

simple
surface applications

drug

2. infiltration - base of under skin so can
 (affect) tissue to disturb contact with nerve endings
subcutaneous

usually
very dilute solutions (1% or less)
proceed is not desired

absorption - do not want

1. Reduce local constriction \rightarrow add contraction - excessive
collapse or adrenaline

outline

6/6/50
21

F. analgesics

(direct or block reception of pain)

How done

1. block applied pain impulses (end nerves or local anesthetic)
(depression in a sense nerve to block conduction)

2. nerve conduction within "CNS"
(salicylic acid series)

↳
aspirin etc

p. 4 "2" also are antipyretics (but only in
presence of fever - not for normal temp).

3. modifying reaction to pain impulses = reduction
of euphoria (reaction of comfort & well-being)

i.e. - morphine (reaction to pain is modified)
characteristic of opiates

Dangerous condition - can likely lead to habitua-
tion.

no explanation for habituation

mechanical blockade differs

codeine - weakly

idion - stronger (4 words as far as addiction)

↳ adding drug to eff

Drawbacks

a. Danger of addiction

b. For reducing effects on smooth (involuntary)
muscle

→ nausea
vomiting
constipation

c. ~~Control~~ Natural source of these drugs (opium
plant) has got out of control.

d. ~~Human~~ plant provides for edible oil - but also
contains morphine (all with opium)

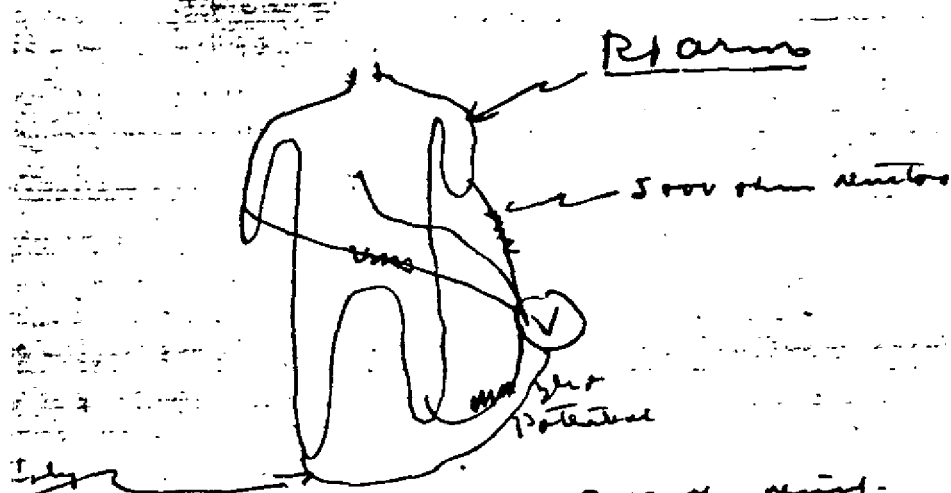
3rd table a middle p. 44

E.K.G.

C.L. not satisfactory
 C.F. used in many cases but not ideal.
 C.R.

6/6/50
 JMS

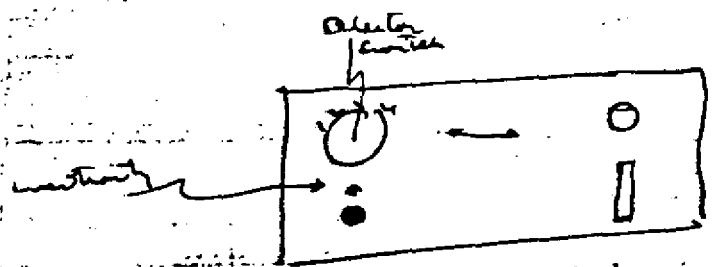
V. Unipolar leads



Waters lead 500 ohm resistor.
 Goldberger leads no resistor - gives accurate

1. EKG leads
2. unipolar leads (as above)
3. Unipolar leads
4. Bipolar leads
5. augmented unipolar leads

How take E.K.G.



Automatic switch for recording unipolar leads

E.K.G. machine

E-K-G.

6/6/50
JP

1. Detect & Treat Arrhythmia
2. Detect & Treat
3. Detect & Treat Myocardial Disease
4. Occasionally Etiological Diagnosis

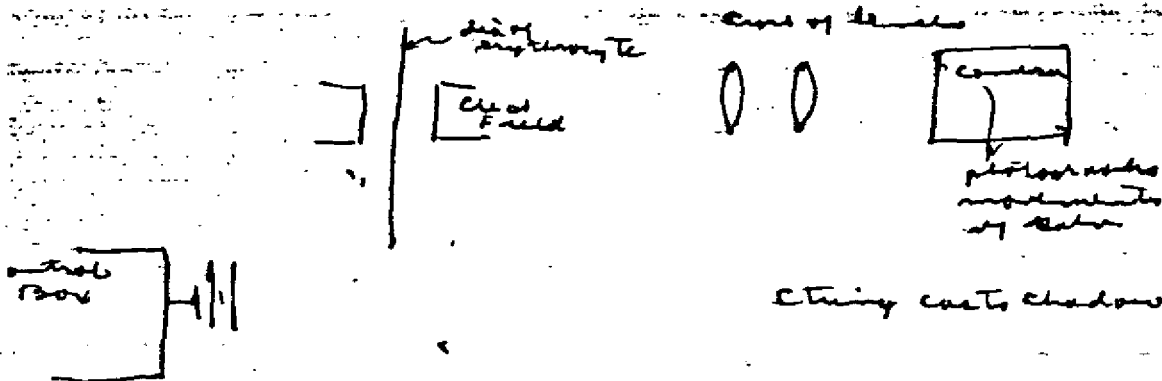
ECG Diagram

cardiac circuit with Fluoroscope

1937 Ludwig →
1902 Finis used

ECG Tube

Quartz Filter - coated



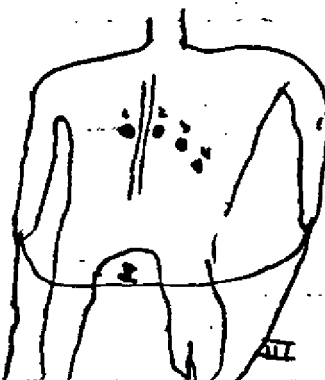
Triples
1. ECG tube - camera

2. mirror →
3. view window
4. V. tube



Preceded leads

Pos. 1 - 12 mm scale as
Pos. 2 - 12 mm scale as

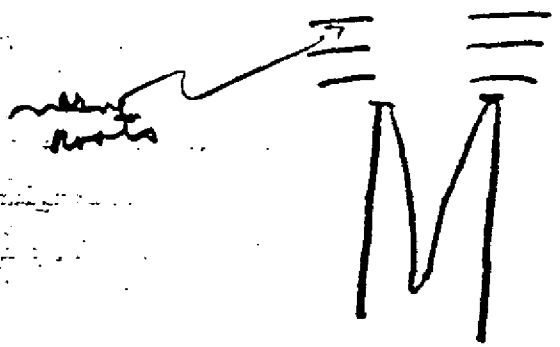


Question may help how does anesthetic effect
 due to ?

6/6/50
 720

3. Conduction Anesthesia

block conduct of nerve impulses along a nerve trunk
 (same anesthesia in peripheral distribution of nerve)
 or still in existing nerve (of
 spinal anesthesia (lumbar region) - inject drug
 substance in central canal space - & drugs
 pass up & down & with nerve roots (block
 conduction)



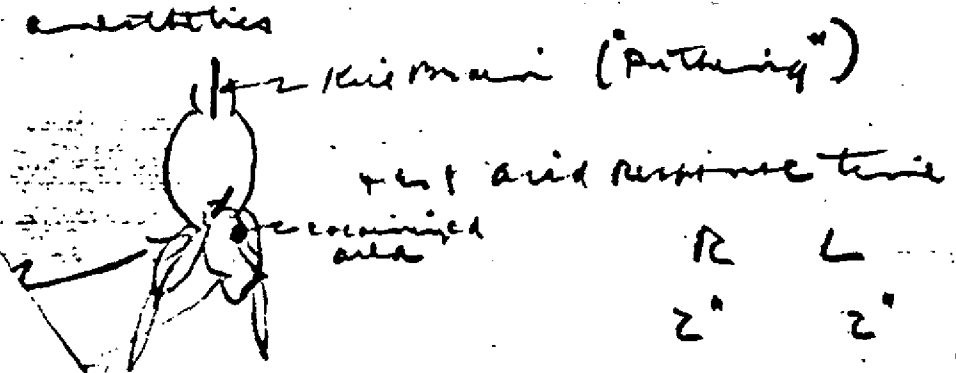
add glucose (smaller heavier than
 spinal fluid)
 & hold patient head down

can ender perform miltoid operations in head
 pain in tongue - but not safe

surgeons nowadays will not use spinal anesthesia
 for head higher than abdomen (higher than diaphragm)

Reason get to thoracic region & peripheral
 sympathetic nerves & get down to p. ->
 collapsed

Difference in sensitivity of diff nerve fibers to ~~anesthetics~~
 anesthetics

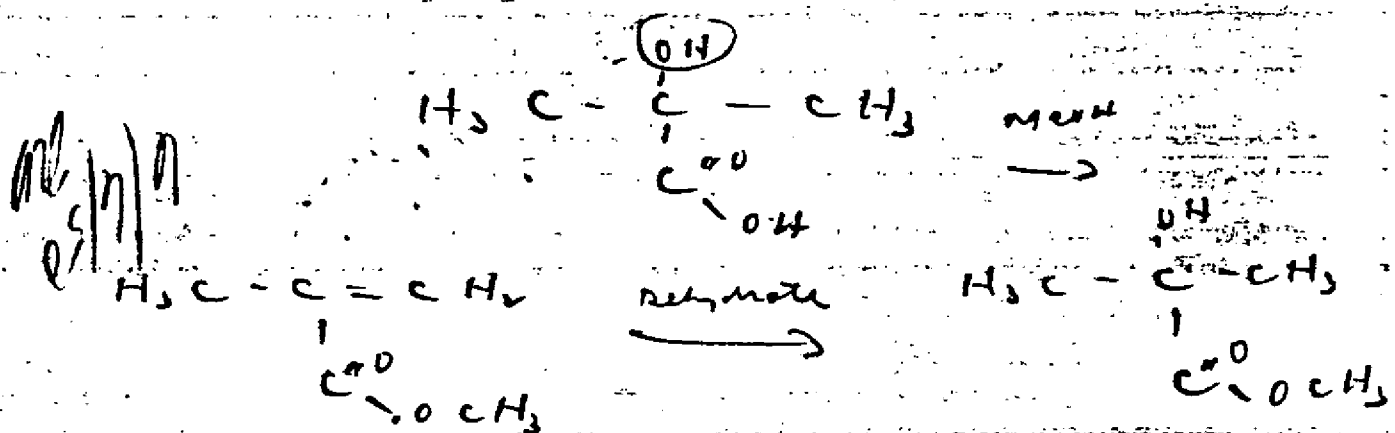
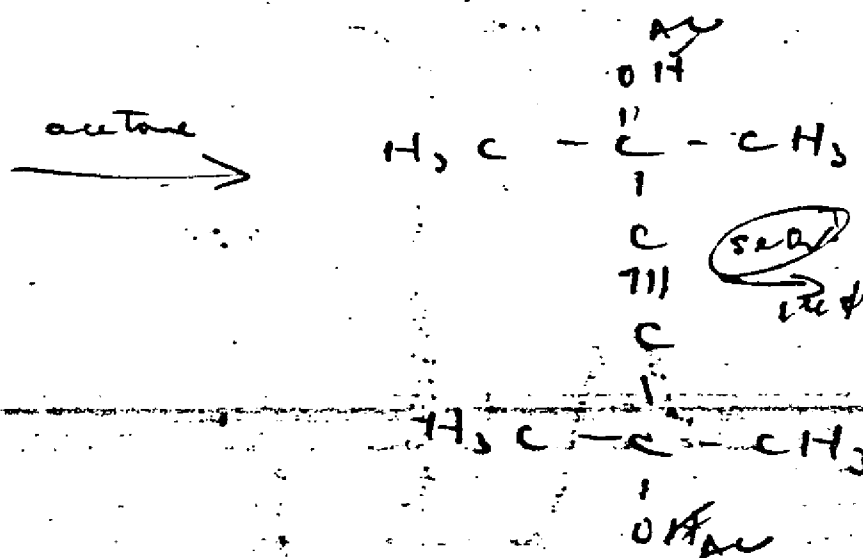
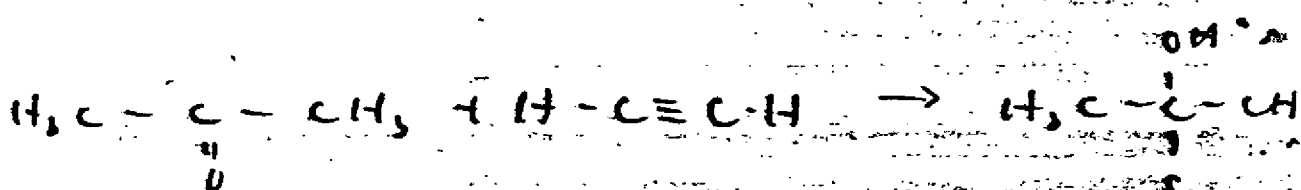
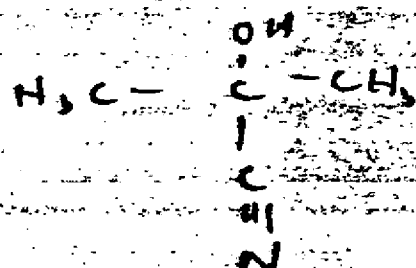
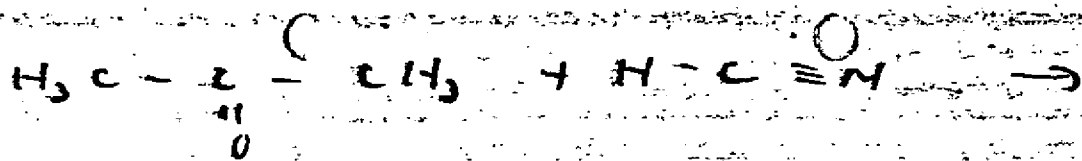


R L
 2" 2"

put cotton + pressure

an apply electrode
 (not a bit of current but a little bit)

1 Over 2



File cabinet

Link (??)

✓ cables

✓ network

✓ Keitt

✓ Table

✓ Vessels

✓ pH app.

✓ Door

✓ ~~Wires~~

✓ Wires

air

part

Antennae



6/6/50
JMS

422

Salmon

10-16-47

K → C-Mood & urine O

Introduce
to
follow in Herb Kuhn

Kamden's
Wise Book

unimpaired
Casts

Control diet

6/6/50
NB

→ K in Diabetic case Louis was his
adaptation one of K → normal → low

3. pH on Vomiting casts

4. P

• mercuric casts

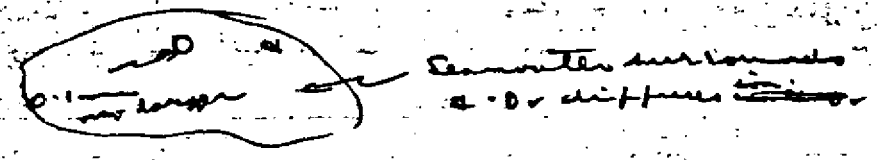
• about at least 2 or 3 more - use fractional
urine & urine

Effect of Drugs on Respiratory System

ventilation - gas exchange between living organism & its environment, and intake of O_2 & exhalation of CO_2 .

in physiology - refers to respiratory movement in higher animal or man

1. single cell (protozoic organism)



6/6/50
TM

2. combination of cells

pumps gas with air

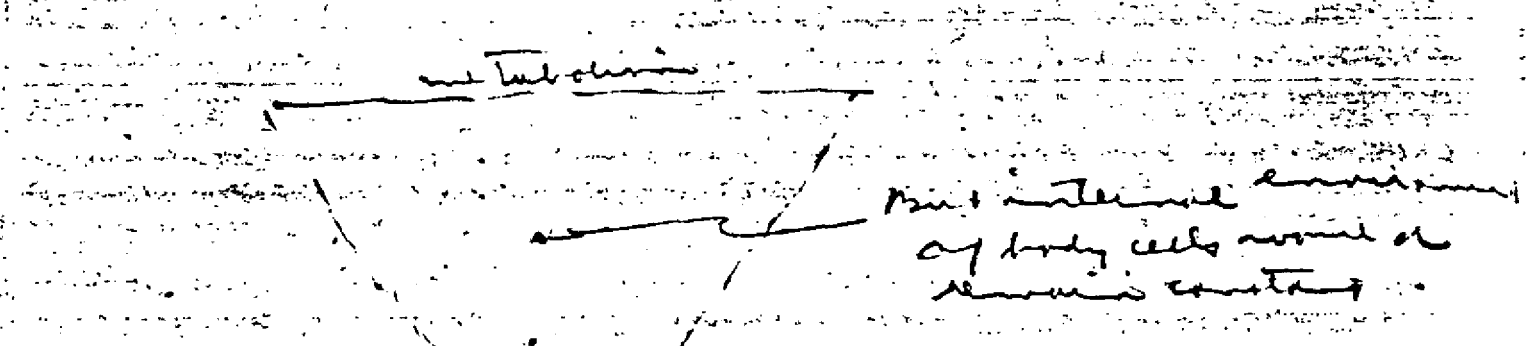
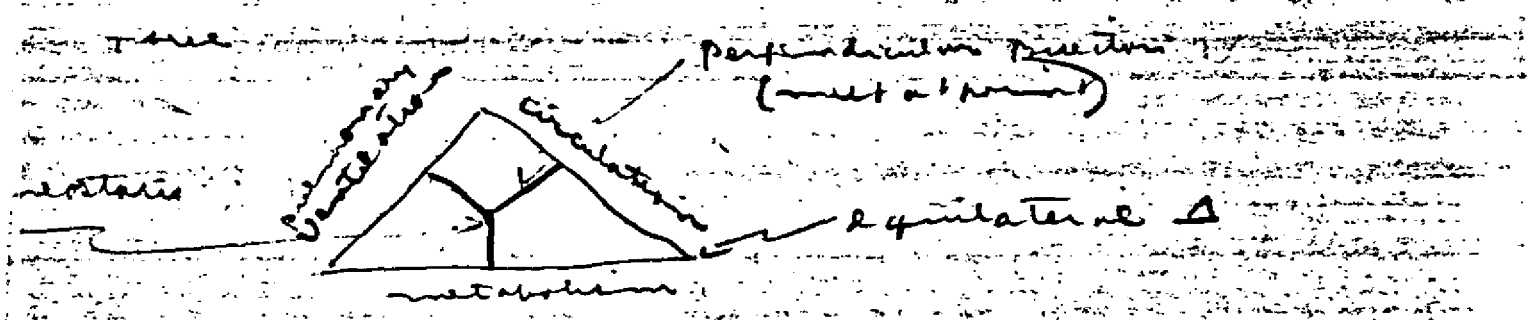
used transport mechanism

3. then needed to > area of contact

in aquatic organism - gills

in air breathing animal - lungs - thin layers of air over tremendously large area

4. then come rhythmic respiratory movements (of air in & out of lungs)



Krebs cycle - part 1

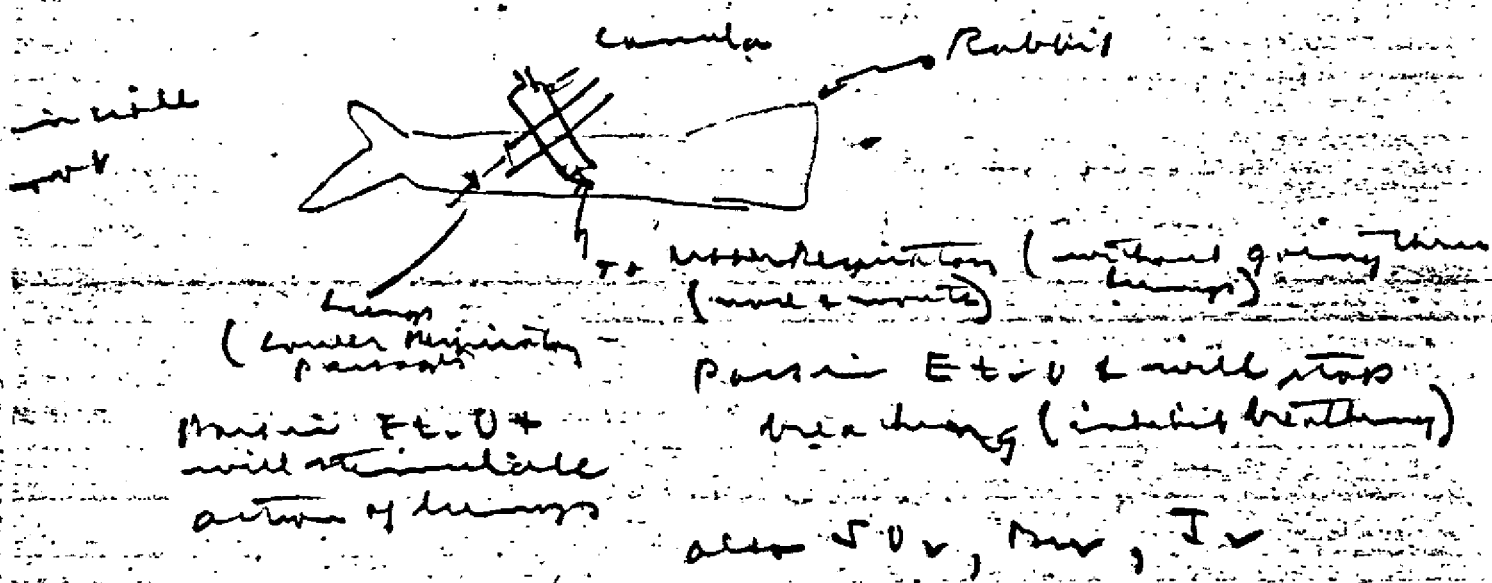
→ Fixation of CO_2

6/6/50
710

- Reduced O_2 tensions are not stimulants to respiratory centers - actually is a depressant.
- If O_2 can't depress ventilatory center but it's extent is slight & range is narrow
- CO_2 → no inhibitory activity of center rhythmically

How CO_2 works

- Hering - Breuer reflex - checks expansion of air when lungs are filled & satisfactorily
- Kretschmer reflex - distention of lungs will check inspiration



Carotid & aortic nodes are stimulated by air

low O_2 — low O_2 most powerful stimulus

> CO_2

> acidity

Conduction

Carotid & aortic nodes stimulated by CO_2 of O_2 drops

in blood (like mitochondria)

4. CO₂ rising in coming to be a recognized clinical entity.

6/6/50
JUN

↓
Tall in Trade

↓
completely
estimate englobes with
but ~~had~~ ^{had} provision for
removal of CO₂ & as
by ~~has~~ ^{has} ~~some~~ ^{some} tension may go
to 110 torr or more. ^{Hy. pressure}
will determine that he looks fine

check by
arterial blood
specimen

5. Treatment of Cough

↓
(method of getting rid of irritant)
action

- a. Inspiration of air
- b. Expiratory relaxation

can use (depressant drugs)

- a. Codeine & heroin
- b. Morphine & ^(as it produces depressant)

or
Try to remove cause of irritation

- a. Idea of depressant coating ^{very} ~~very~~ ^{of} ~~of~~ ^{upper} ~~upper~~ ^{respiratory} ~~respiratory~~ ^{membrane} ~~membrane~~
(cough suppressants) (coughers)
- b. Expectorants (Idea - ~~engage~~ ^{engage} ~~relaxation~~ ^{relaxation} of ~~respiratory~~ ^{respiratory} ~~tissue~~ ^{tissue})

↓
no real scientific background

non-scientific basis
I. D. Allen etc

PHN cl
PHN cl

Parke Davis - Volstead

of sympathomimetic drugs

↳ mimics effect of
sympathetic nervous system

apinephrine
adrenaline
epinephrine
6/6/50

maybe the stimulation of apinephrine may
be even more effective than n.v. - relaxation
effect - explain?

hemoglobin

Fe^{++}

with hemoglobin

Fe^{++}

can no longer combine
with O_2 -
dark gray color of prussian-
blue + white (can also
be caused by iron drugs)

also, for combining with hemoglobin

$CO : O_2 = 1 : 2.10$

CO mixed

- also {
1. Subliming hell good
 2. very pink color (CO hemoglobin)
 3. Doesn't give any warning

CO poisoning (+ O_2 tension is normal)
(not enough O_2 present)
arterial O_2 is normal
but O_2 does not present

anemia - 4.5 - 5.0 million red blood cells/cu
mm

a severe anemia may be related to low O_2

now only treat by blood transfusion or $FeSO_4$
 Fe gluconate (easier to take)

11-8-47

1. Hypoxemic anoxia - difficulty due to circulation - i.e. cardiovascular failure

can be a - anemic (arterial)

b - congestive (venous)

6/6/50
JP

3. overutilization anoxia - more demand for O₂ than can get - i.e. when 1 mile - occurs at this level

4. histotoxic anoxia - tissues especially which activate O₂ can no longer take it up. i.e. caused by H₂O₂ & cyanide.

what does it look like??

a - Simulates drunkenness (first Mergat 50 yrs ago)

and individuals vary from nervous to aggressiveness & are completely unaware of their poor judgment (air pilots in war)

b - Can even deceive physicians - later (due to Enigma) says he felt better

during war

a. Ways of getting O₂ other than by mouth - more successful if swallowing H₂O₂ & other chemical methods for improving altitude tolerance.

b. Nicotinamide, carbohydrate (attempts to train catfish diet) & a.f. altitude is shown as trained humans - can't train

6/6/50
gm

$$\begin{array}{r} 3 \overline{) 23.0} \\ 7.7 \end{array}$$

27.5

31.5

$$\begin{array}{r} 8.5 \\ 7.5 \\ 9.0 \\ 3 \overline{) 25.0} \\ 8.3 \\ 8.2 \end{array}$$

$$\begin{array}{r} 3 \overline{) 22.0} \\ 7.3 \end{array}$$

30.5
30.0

$$\begin{array}{r} 3 \overline{) 29.0} \\ 9.7 \end{array}$$

$$\begin{array}{r} 3 \overline{) 27.5} \\ 9.2 \\ 30.5 \\ 30.0 \end{array}$$

$$\begin{array}{r} 29.0 \\ 29.0 \\ \hline \end{array}$$

300
295

195

290

$$\begin{array}{r} 3 \overline{) 29.0} \\ 9.7 \end{array}$$

$$\frac{3}{1} = \frac{60}{100} = \frac{160}{250}$$

$$3 \rightarrow 9$$

(10)

$$\frac{160}{100} = 1.6$$

$$3 \rightarrow 9$$

310

$$3 \rightarrow 9$$

$$5 \times \frac{4}{3} = \frac{20}{3} = 6.66$$

48.0

2/19/77
W

Materials list for Heart Station

Quantity	Marshaw Catalogue No.	Description	Price
1	H-28900	Beckman PH water, model G; complete with batteries, electrodes, solutions, standard cell and cup.	\$245.00
2	H-29049	Hypodermic Glass Electrodes	\$30.00
1	H-14200	Klett - Summerson colorimeter, clinical model. Two H-1428 Klett - Summerson fitters, nos. K-S 54 and K-S 64 are included.	\$148.00
12	H-10640	Blood volume index centrifuge tubes	\$ 10.48
2	H-6350	Koch micro-Burets/ standard taper points, two interchangeable outlet tips - capacity, 5 ml.	\$ 29.30
12	H-24640	Volumetric Flasks; standard taper glass stoppered - capacity, 100 ml.	\$ 28.62
6	H-18460	Cylinders; standard taper glass stoppered, pyrex glass - 10 ml. capacity.	\$ 10.20
1	6404	Van Dyke nonmetric blood gas apparatus	\$248.00

Materials list for Heart Station

05/19/0

<u>Quantity</u>	<u>Marshaw Catalogue No.</u>	<u>Description</u>	<u>Price</u>
1	H-28900	Beckman PH water, model G; complete with batteries, electrodes, solutions, standard cell and cup.	\$245.00
2	H-29040	Hypodermic Glass Electrodes	\$30.00
1	H-14200	Klett - Summerson colorimeter, clinical model. Two H-1428 Klett - Summerson filters, nos. K-S 54 and K-S 64 are included.	\$148.00
12	H-10640	Blood volume index centrifuge tubes	\$10.48
2	H-6350	Koch micro-Burets/ standard taper joints, two interchangeable outlet tips - capacity, 5 ml.	\$ 29.30
12	H-24640	Volumetric Flasks; standard taper glass stoppered - capacity, 100 ml.	\$ 28.62
6	H-18460	Cylinders; standard taper glass stoppered, Pyrex glass - 10 ml. capacity.	\$ 10.20
1	6404	Van Slyke monometric blood gas apparatus	\$248.00

Materials list for Heart Station (cont.)

<u>Quantity</u>	<u>Marshaw Catalogue No.</u>	<u>Description</u>	<u>Price</u>
1		6 volt, 120 amp. hr. Storage battery for use with Beckman model DV Spectrophotometer.	
1		Trickle charger for use with 6 volt, 120 amp. hr. storage battery.	

NOTE: The above can all be ordered from the Marshaw Scientific Company,
117 S. 17th Street,
Philadelphia, Penna.

They are standard laboratory items which sell at a fixed price, regardless of the supply house from which they are ordered.

Other supply houses where these materials may be obtained are:

Arthur H. Thomas Company,
230 S. 7th Street,
Philadelphia, Penna.

Williams, Brown and Earle,
918 Chestnut Street,
Philadelphia, Penna.

In addition, there are needed:

- 1 - Filing cabinet
- 1 - Table, solidly built, approx. size: length, 65", width 35", height, 36".
- 1 - Sink - the present sink should be raised to a height of 36" (from floor to tap).

Materials list for Heart Station (cont.)

<u>Quantity</u>	<u>Marshaw Catalogue No.</u>	<u>Description</u>	<u>Price</u>
1		6 volt, 120 amp. hr. Storage battery for use with Beckman model DV Spectrophotometer.	38.45
1		Trickle charger for use with 6 volt, 120 amp. hr. storage battery.	17.25

NOTE: ~~The above can all be ordered from the Marshaw Scientific Company,~~
117 S. 17th Street,
Philadelphia, Penna.

~~They are standard laboratory items which sell at a fixed price, regardless of the supply house from which they are ordered.~~

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- 1 - Filing cabinet
- 1 - Table, solidly built, Approx. size: length, 65", width 35", height, 36".
- 1 - Sink - the present sink should be raised to a height of 36" (from floor to tap).

Materials list for Heart Station

Quantity	Harshaw Catalogue No.	Description	Price
24	N-5800	Pyrex Beakers - capacity, 250 ml.	\$6.40
2	N-11090	Double cast alloy Buret clamps	\$5.00
1	N-6110	Mohr Buret - capacity, 100 ml.	\$2.54
2	N-6289	Dispensing Buret	\$13.32
1	N-60940	Buret Support, Fisher	\$ 6.60
36	N-61610	Test tubes, Pyrex 15 mm. diameter 125 mm. length	\$ 1.62
<hr/>			
		Mechanical Refrigerator - capacity approx. 6 cu. ft.	

05/19/77
M

2.

Materials list for Heart Station

<u>Quantity</u>	<u>Harshaw Catalogue No.</u>	<u>Description</u>	<u>Price</u>
24	N-3800	Pyrex Beakers - capacity, 20 ml.	\$6.48
2	N-11090	Double cast alloy Buret clamps	\$5.00
1	N-6110	Mehr Buret - capacity, 180 ml.	\$2.54
2	N-6280	Dispensing Buret	\$13.32
1	N-60940	Buret Support, Fisher	\$ 6.60
36	N-61610	Test tubes, Pyrex 15 mm. diameter 125 mm. length	\$ 1.62
1		Mechanical Refrigerator - capacity approx. 6 cu. ft.	

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OF THE STATE OF PENNSYLVANIA
PHILADELPHIA 3, PENNA.

SEC 562 PL&R



Mr. Harry Gold
4823 Kindred St.
Philadelphia 24, Pa.

6/6/50
24

THE AMERICAN INSTITUTE OF ELECTRICAL ENGINEERS
THE ENGINEERS CLUB

1317 Spruce Street, Philadelphia 7, Pa.

Attention R. E. Watson

I enclose a check for \$_____ for the SYMPOSIUM ON SOME RECENT INTER-
RELATED ADVANCES IN PHYSICS AND ELECTRICAL ENGINEERING.

NAME _____

Phone _____

ADDRESS _____

Make checks payable to "A. I. E. E.—Philadelphia Section"

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The American Institute of
Electrical Engineers
&

The Franklin Institute

ANNOUNCE A SYMPOSIUM ON

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Advances in Physics and
Electrical Engineering

to be held at

THE FRANKLIN INSTITUTE
Benjamin Franklin Parkway at 20th Street
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PARKWAY ENTRANCE

PROGRAM

Thursday, October 28, 1948, 7:30 P. M.

**"The Present Status of Fundamental
Particles in Physics—A Survey"**

Speaker to be announced

Tuesday, November 9, 1948, 7:30 P. M.

**"The Medical Use of Radioisotopes for
Tracing and Therapy"**

RICHARD H. CHAMBERLAIN, M. D.

Department of Radiology, Hospital of the Univer-
sity of Pennsylvania

Tuesday, November 30, 1948, 7:30 P. M.

**"Some Dividends for Fundamental
Physics from War-time Investments
in Microwave Technology"**

DR. CHARLES H. TOWNES

Associate Professor of Physics, Columbia
University

Wednesday, December 8, 1948, 7:30 P. M.

**"The Synchrotron Accelerator Project
at Brookhaven"**

DR. G. KENNETH GREEN

Physicist, Accelerator Project, Brookhaven
National Laboratory

ALL LECTURES will be held in the
Lecture Hall of The Franklin Institute
beginning at 7:30 P. M.

The fee for this symposium will
be \$3.25 for members of either The
Franklin Institute or The American
Institute of Electrical Engineers. The
fee for single lectures will be \$1.00,
and the fee for non-members for the
series will be \$4.00. Attendance will be
limited to 350.

Please use the reply card on the
back of this page for registration.

6/6/50
gjo

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 Ed 12 Edition 12

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 Ed 2

MEMBER'S NAME Harry Gold

JULIAN PAUL BRODIE

September 15, 1947

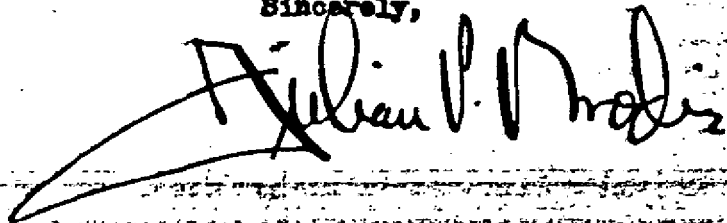
6/6/50
20

Dear Mr. Brothman:

I think you will agree with me that I have been more than patient in waiting for you to develop a man's vanishing cream which you were supposed to complete for me within three months or less from September 13, 1946. There must be an end to all things, and I think in all fairness to me that you should either complete the job or advise me frankly that you cannot proceed and return the \$650 I paid you.

Under the circumstances, I will make this final request that you deliver the completed product to me within ten days from date hereof or, if you are unable to do so, to return my \$650, and let us call the deal off.

Sincerely,



Mr. A. Brothman
A. Brothman and Associates
85-03 57th Avenue
Long Island City, New York

File this return with Collector of Internal Revenue on or before March 15, 1947. Any balance of tax due (item 9, below) must be paid in full with return. See separate instructions for filling out return.

Page 1

FORM 1040

Treasury Department
Internal Revenue Service

U. S. INDIVIDUAL INCOME TAX RETURN

FOR CALENDAR YEAR 1946

1946

or fiscal year beginning _____, 1946, and ending _____, 1947

EMPLOYEES.—Instead of this form, you may use your Withholding Statement, Form W-2, as your return, if your total income was less than \$5,000, consisting wholly of wages shown on Withholding Statements or of such wages and not more than \$100 of other wages, dividends, and interest.

Name _____
(PLEASE PRINT. If this return is for a husband and wife, use both first names.)

ADDRESS _____
(PLEASE PRINT. Street and number or rural route.)

(City or town, postal office number) (County) (State)

Occupation _____ Social Security No. _____

Do not write in these spaces

File Code _____
Serial No. _____

District _____

(Cashier's Stamp)

6/16/50
JTB

List your own name.
If married and your wife (or husband) had no income, or if this is a joint return of husband and wife, list name of your wife (or husband).

List names of other close relatives (as defined in Instruction 1) with 1946 incomes of less than \$500 who received more than one-half of their support from you. If this is a joint return of husband and wife, list dependent relatives of both.

Your Exemptions

1.	Name (show title)	Relationship	Name (show title)	Relationship
Your name		*****		

Your Income

Enter your total wages, salaries, bonuses, commissions, and other compensation received in 1946, BEFORE PAY-ROLL DEDUCTIONS for taxes, dues, insurance, bonds, etc. Members of armed forces and persons claiming traveling or reimbursed expenses, see Instruction 2.

2.	First Employer's Name	Where Employed (City and State)	Amount

3. Enter here the total amount of your dividends	Enter total here →
4. Enter here the total amount of your interest (including interest from Government obligations unless wholly exempt from taxation)	
5. If you received any other income, give details on page 2 and enter the total here	
6. Add amounts in items 2, 3, 4, and 5, and enter the total here	

How to Figure Your Tax

IF YOUR INCOME WAS LESS THAN \$5,000.—You may find your tax in the tax table on page 4. This table, which is provided by law, automatically allows about 10 percent of your total income for charitable contributions, interest, taxes, casualty losses, medical expenses, and miscellaneous expenses. If your expenditures and losses of these classes amount to more than 10 percent, it will usually be to your advantage to itemize them and compute your tax on page 2.

IF YOUR INCOME WAS \$5,000 OR MORE.—Disregard the tax table and compute your tax on page 2. You may either take a standard deduction of \$500 or itemize your deductions, whichever is to your advantage.

HUSBAND AND WIFE.—If husband and wife file separate returns, and one itemizes deductions, the other must also itemize deductions.

Tax Due or Refund

7. Enter your tax from table on page 4, or from line 12, page 3	
8. How much have you paid on your 1946 income tax? (A) By withholding from your wages (B) By payments on 1946 Declaration of Estimated Tax	Enter total here →
9. If your tax (item 7) is larger than payments (item 8), enter BALANCE OF TAX DUE here	
10. If your payments (item 8) are larger than your tax (item 7), enter the OVERPAYMENT here	

Check (✓) whether you want this overpayment: Refunded to you ☐ or Credited on your 1947 estimated tax ☐

If you filed a return for a prior year, what was the latest year?

To which Collector's office was it sent?

To which Collector's office did you pay amount claimed in item 8 (B), above?

Is your wife (or husband) making a separate return for 1946?

If "Yes," write below:

Name of wife (or husband)

Collector's office to which sent

I declare under the penalties of perjury that this return (including any accompanying schedules and statements) has been examined by me and to the best of my knowledge and belief is a true, correct, and complete return.

(Signature of person (other than taxpayer or agent) preparing return)

(Date)

(Signature of taxpayer)

(Date)

(Name of firm or employer, if any)

(If this is a joint return of husband and wife, it must be signed by both)

Do not deduct (1) You determine your tax from the tax table on page 2. (2) Your total income is \$5,000 or more and you claim the \$500 standard deduction. If husband and wife filing together at end of year file separate returns and one itemizes deductions, the other must file his or her return on Form 1040, and must also itemize deductions.

DEDUCTIONS

Describe deductions and state to whom paid. If more space is needed, list deductions on separate sheet of paper and attach to this return.

Amount

Contributions		\$	
		\$	
	Allowable Contributions (not in excess of 15 percent of item 6, page 1)	\$	
Interest		\$	
		\$	
	Total Interest	\$	
Taxes		\$	
		\$	
	Total Taxes	\$	
Losses from fire, storm, shipwreck, or other casualty, or theft		\$	
		\$	
	Total Allowable Losses (not compensated by insurance or otherwise)	\$	
Medical and dental expenses		\$	
	Net Expenses (not compensated by insurance or otherwise)	\$	
	Enter 5 percent of item 6, page 1, and subtract from Net Expenses. Allowable Medical and Dental Expenses. See instruction for limitation.	\$	
Miscellaneous (See instructions)		\$	
		\$	
	Total Miscellaneous Deductions	\$	
TOTAL DEDUCTIONS		\$	

TAX COMPUTATION—FOR PERSONS NOT USING TAX TABLE ON PAGE 2

1. Enter amount shown in item 6, page 1. This is your Adjusted Gross Income.	\$	
2. Enter DEDUCTIONS (If deductions are itemized above, enter the total of such deductions; if adjusted gross income (line 1, above) is \$5,000 or more and deductions are not itemized, enter the standard deduction of \$500).	\$	
3. Subtract line 2 from line 1. Enter the difference here. This is your Net Income.	\$	
4. Enter your exemptions (\$500 for each person whose name is listed in item 1, page 1).	\$	
5. Subtract line 4 from line 3. Enter the difference here.	\$	
6. Use the tax rates in instruction sheet to figure your combined tentative normal tax and surtax on amount entered on line 5. Enter the tentative tax here. (If line 5 above includes partially tax-exempt interest, see Tax Computation instructions).	\$	
7. Enter here 5 percent of amount entered on line 6.	\$	
8. Subtract line 7 from line 6. Enter the difference here. This is your combined normal tax and surtax. (If alternative tax computation is made on separate Schedule D, enter here tax from line 12 of Schedule D).	\$	
IF YOU USED THE \$500 STANDARD DEDUCTION IN LINE 2, DISREGARD LINES 2, 4, 5, AND 7, AND COPY ON LINE 12 THE SAME FIGURE YOU ENTERED ON LINE 6.		
9. Enter here any income tax payments to a foreign country or U. S. possession (attach Form 1116).	\$	
10. Enter here any income tax paid at source on tax-free covenant bond interest.	\$	
11. Add the figures on lines 9 and 10 and enter the total here.	\$	
12. Subtract line 11 from line 8. Enter the difference here and in item 7, page 1. This is your tax.	\$	

Do not use this page if your income is wholly from salaries, wages, dividends, and interest

Page 2

Schedule A.—INCOME FROM ANNUITIES OR PENSIONS

1. Cost of annuity (total amount you paid in)	\$		4. Total amount received this year	\$	
2. Amount received tax-free in prior years			5. Excess, if any, of line 4 over line 3		
3. Remainder of your cost (line 1 less line 2)	\$		6. Enter line 5, or 3 percent of line 1, whichever is greater (Attach separate schedule for each additional annuity if needed)	\$	

Schedule B.—INCOME FROM RENTS AND ROYALTIES

1. Kind of property	2. Amount of rent or royalty	3. Depreciation or depletion (explain in Schedule F)	4. Expenses (explain in Schedule G)	5. Other expenses (explain in Schedule G)
	\$	\$	\$	\$
	\$	\$	\$	\$
	\$	\$	\$	\$
Net profit (or loss) (col. 2 less sum of cols. 3, 4, and 5)	\$	\$	\$	\$

Schedule C.—PROFIT (OR LOSS) FROM BUSINESS OR PROFESSION. (Farmers should obtain Form 1560F)

(State (1) nature of business; (2) business name)

1. Total receipts

COST OF GOODS SOLD

(To be used when inventories are an income-determining factor)
(Enter the letters "C" or "I" on lines 2 and 3 if inventories are valued at either cost, or cost or market, whichever is lower)

2. Inventory at beginning of year

3. Merchandise bought for sale

4. Labor

5. Material and supplies

6. Other costs

(explain in Schedule G)

7. Total of lines 2 to 6

8. Less inventory at end of year

9. Net cost of goods sold (line 7 less line 8)

10. Gross profit (line 1 less line 9)

OTHER BUSINESS DEDUCTIONS

11. Salaries and wages not in line 4

12. Interest on business indebtedness

13. Taxes on business and business property

14. Losses (explain in Schedule G)

15. Bad debts arising from sales or services

16. Depreciation, obsolescence and depletion (explain in Schedule F)

17. Rent, repairs, and other expenses (explain in Schedule G)

18. Amortization of emergency facilities (attach statement)

19. Net operating loss deduction (attach statement)

20. Total of lines 11 to 19

21. Total of lines 9 and 20

22. Net profit (or loss) (line 1 less line 21)

Schedule D.—GAINS AND LOSSES FROM SALES OR EXCHANGES OF CAPITAL ASSETS, ETC.

1. Net gain (or loss) from sale or exchange of capital assets (from separate Schedule D)

2. Net gain (or loss) from sale or exchange of property other than capital assets (from separate Schedule D)

Schedule E.—INCOME FROM PARTNERSHIPS, ESTATES AND TRUSTS, AND OTHER SOURCES

1. Name and address of partnership, syndicate, etc.

Amount

2. Name and address of estate or trust

Amount

3. Other sources (state nature)

Amount

4. Total

Total income from above sources (Enter as item 5, page 1)

Schedule F.—EXPLANATION OF DEDUCTION FOR DEPRECIATION CLAIMED IN SCHEDULES B AND C

1. Kind of property (if building, state nature of which construction)	2. Date acquired	3. Cost or other basis (do not include land or other nondepreciable property)	4. Annual depreciation claimed in year at end of year	5. Depreciation allowed (or allowable) in prior years	6. Remaining cost or other basis at end of year	7. Estimated life (in years) or other basis in years	8. Estimated remaining life (in years) beginning of year	9. Depreciation allowable this year
		\$	\$	\$	\$			\$
		\$	\$	\$	\$			\$
		\$	\$	\$	\$			\$

Schedule G.—EXPLANATION OF COLUMNS 4 AND 5 OF SCHEDULE B, AND LINES 6, 14, AND 17 OF SCHEDULE C

1. Column of Line No.	2. Explanation	3. Amount	1. Column of Line No.	2. Explanation	3. Amount
		\$			\$
		\$			\$
		\$			\$

If you use this table, tear off this page and file only pages 1 and 2

TAX TABLE

FOR PERSONS WITH INCOMES UNDER \$5,000 NOT COMPUTING TAX ON PAGE 3

Read down the shaded columns below until you find the line covering the total income you declared in Item 6, page 1. Then read across to the column headed by the number corresponding to the number of persons listed in Item 1, page 1. Enter the tax you find there in Item 7, page 1.

If total income is from - 4, page 1, is -		And the number of persons listed is - Item 1, page 1, is -				If total income is from - 4, page 1, is -		And the number of persons listed is from 1, page 1, is -									
All kind	Net less deductions	1	2	3	4 or more	All kind	Net less deductions	1	2	3	4	5	6	7	8	9	10 or more
80	\$550	\$0	\$0	\$0	\$0	\$2,225	\$2,250	\$288	\$193	\$98	\$3	\$0	\$0	\$0	\$0	\$0	\$0
850	575	1	0	0	0	2,250	2,275	292	197	102	7	0	0	0	0	0	0
875	600	5	0	0	0	2,275	2,300	296	201	106	11	0	0	0	0	0	0
900	625	10	0	0	0	2,300	2,325	300	205	110	15	0	0	0	0	0	0
925	650	14	0	0	0	2,325	2,350	305	210	115	20	0	0	0	0	0	0
950	675	18	0	0	0	2,350	2,375	309	214	119	24	0	0	0	0	0	0
975	700	23	0	0	0	2,375	2,400	313	218	123	28	0	0	0	0	0	0
100	725	27	0	0	0	2,400	2,425	318	223	128	33	0	0	0	0	0	0
725	750	31	0	0	0	2,425	2,450	322	227	132	37	0	0	0	0	0	0
750	775	35	0	0	0	2,450	2,475	326	231	136	41	0	0	0	0	0	0
775	800	40	0	0	0	2,475	2,500	330	235	140	45	0	0	0	0	0	0
800	825	44	0	0	0	2,500	2,525	335	240	145	50	0	0	0	0	0	0
825	850	48	0	0	0	2,525	2,550	339	244	149	54	0	0	0	0	0	0
850	875	52	0	0	0	2,550	2,575	343	248	153	58	0	0	0	0	0	0
875	900	57	0	0	0	2,575	2,600	347	252	157	62	0	0	0	0	0	0
900	925	61	0	0	0	2,600	2,625	352	257	162	67	0	0	0	0	0	0
925	950	65	0	0	0	2,625	2,650	356	261	166	71	0	0	0	0	0	0
950	975	70	0	0	0	2,650	2,675	360	265	170	75	0	0	0	0	0	0
975	1,000	74	0	0	0	2,675	2,700	365	270	175	80	0	0	0	0	0	0
1,000	1,025	78	0	0	0	2,700	2,725	369	274	179	84	0	0	0	0	0	0
1,025	1,050	82	0	0	0	2,725	2,750	373	278	183	88	0	0	0	0	0	0
1,050	1,075	87	0	0	0	2,750	2,775	377	282	187	92	0	0	0	0	0	0
1,075	1,100	91	0	0	0	2,775	2,800	382	287	192	97	2	0	0	0	0	0
1,100	1,125	95	0	0	0	2,800	2,825	387	291	196	101	6	0	0	0	0	0
1,125	1,150	100	0	0	0	2,825	2,850	391	295	200	105	10	0	0	0	0	0
1,150	1,175	104	0	0	0	2,850	2,875	396	299	204	109	14	0	0	0	0	0
1,175	1,200	108	12	0	0	2,875	2,900	401	304	209	114	19	0	0	0	0	0
1,200	1,225	112	17	0	0	2,900	2,925	405	308	213	118	23	0	0	0	0	0
1,225	1,250	117	22	0	0	2,925	2,950	410	312	217	122	27	0	0	0	0	0
1,250	1,275	121	26	0	0	2,950	2,975	415	317	222	127	32	0	0	0	0	0
1,275	1,300	125	30	0	0	2,975	3,000	419	321	226	131	36	0	0	0	0	0
1,300	1,325	129	34	0	0	3,000	3,050	427	327	232	137	42	0	0	0	0	0
1,325	1,350	134	39	0	0	3,050	3,100	436	336	241	146	51	0	0	0	0	0
1,350	1,375	138	43	0	0	3,100	3,150	445	344	249	154	59	0	0	0	0	0
1,375	1,400	142	47	0	0	3,150	3,200	455	353	258	163	68	0	0	0	0	0
1,400	1,425	147	52	0	0	3,200	3,250	464	361	266	171	76	0	0	0	0	0
1,425	1,450	151	56	0	0	3,250	3,300	474	370	275	180	85	0	0	0	0	0
1,450	1,475	155	60	0	0	3,300	3,350	483	379	284	189	94	0	0	0	0	0
1,475	1,500	159	64	0	0	3,350	3,400	492	388	292	197	102	7	0	0	0	0
1,500	1,525	164	69	0	0	3,400	3,450	502	397	301	206	111	16	0	0	0	0
1,525	1,550	168	73	0	0	3,450	3,500	511	407	309	214	119	24	0	0	0	0
1,550	1,575	172	77	0	0	3,500	3,550	521	416	318	223	128	33	0	0	0	0
1,575	1,600	176	81	0	0	3,550	3,600	530	425	326	231	136	41	0	0	0	0
1,600	1,625	181	86	0	0	3,600	3,650	539	435	335	240	145	50	0	0	0	0
1,625	1,650	185	90	0	0	3,650	3,700	549	444	343	248	153	58	0	0	0	0
1,650	1,675	189	94	0	0	3,700	3,750	558	454	352	257	162	67	0	0	0	0
1,675	1,700	194	99	4	0	3,750	3,800	568	463	361	266	171	76	0	0	0	0
1,700	1,725	198	103	8	0	3,800	3,850	577	473	369	274	179	84	0	0	0	0
1,725	1,750	202	107	12	0	3,850	3,900	586	482	378	283	188	93	0	0	0	0
1,750	1,775	206	111	16	0	3,900	3,950	596	491	387	291	196	101	6	0	0	0
1,775	1,800	211	116	21	0	3,950	4,000	605	501	396	300	205	110	15	0	0	0
1,800	1,825	215	120	25	0	4,000	4,050	615	510	406	308	213	118	23	0	0	0
1,825	1,850	219	124	29	0	4,050	4,100	624	520	415	317	222	127	32	0	0	0
1,850	1,875	223	128	33	0	4,100	4,150	633	529	424	325	230	135	40	0	0	0
1,875	1,900	228	133	38	0	4,150	4,200	643	538	434	334	239	144	49	0	0	0
1,900	1,925	232	137	42	0	4,200	4,250	652	548	443	342	247	152	57	0	0	0
1,925	1,950	236	141	46	0	4,250	4,300	662	557	453	351	256	161	66	0	0	0
1,950	1,975	241	146	51	0	4,300	4,350	671	567	462	360	265	170	75	0	0	0
1,975	2,000	245	150	55	0	4,350	4,400	680	576	471	368	273	178	83	0	0	0
2,000	2,025	249	154	59	0	4,400	4,450	690	585	481	377	282	187	92	0	0	0
2,025	2,050	253	158	63	0	4,450	4,500	699	595	490	386	290	195	100	5	0	0
2,050	2,075	258	163	68	0	4,500	4,550	709	604	500	395	299	204	109	14	0	0
2,075	2,100	262	167	72	0	4,550	4,600	718	614	509	405	307	212	117	22	0	0
2,100	2,125	266	171	76	0	4,600	4,650	727	623	518	414	316	221	126	31	0	0
2,125	2,150	271	176	81	0	4,650	4,700	737	632	528	423	324	229	134	39	0	0
2,150	2,175	275	180	85	0	4,700	4,750	746	643	537	433	333	238	143	48	0	0
2,175	2,200	279	184	89	0	4,750	4,800	756	651	547	442	342	247	152	57	0	0
2,200	2,225	283	188	93	0	4,800	4,850	765	661	556	451	350	255	160	65	0	0
						4,850	4,900	774	670	565	460	359	264	169	74	0	0
						4,900	4,950	784	679	575	470	367	273	177	82	0	0
						4,950	5,000	793	689	584	480	376	281	186	91	0	0

Date Received: 12/1/50

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To Be: [illegible]

SAC, PHILADELPHIA

July 7, 1950

T. SCOTT MILLER, SA

HARRY GOLD, was.,
ESPIONAGE - R

65-4307-1B 12 (1) Folder No. 9

GOLD advised on June 24, 1950 that the material in this folder are notes and material in connection with a course given by Dr. CARL SCHMIDT, Professor in Pharmacology at the University of Pennsylvania Medical School. GOLD said that he took this course in the Fall of 1948 which was given by the Philadelphia Section of the American Chemical Society and paid \$10.00 to take the course.

GOLD said he thought that he needed this course in connection with his position at the Philadelphia General Hospital.

The notes in this folder are in the handwriting of HARRY GOLD.

TSK:ELC
65-4307

CHEMOTHERAPY

(Specific treatment of diseases due to living organisms by chemical agents that are more toxic to the tissues of the invading organisms than to those of the host).

I. Anthelmintics (drugs used against parasitic worms).

A. Against intestinal parasites

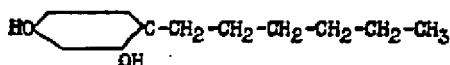
Natural (vegetable) products: Aspidium, ⁽¹⁾ ~~homogranate~~ ⁽²⁾ (pelletierine), chenopodium, santonin, thymol, fig latex (ficin).

Synthetic products:-

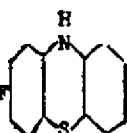
a. Carbon tetrachloride USP C-Cl_4

b. Tetrachlorethylene USP $\text{Cl}_2\text{C:CCl}_2$

c. Hexylresorcinol USP

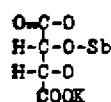


d. Phenothiazine NF

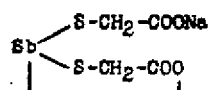


B. Against tissue parasites (leishmaniasis, filariasis etc.)

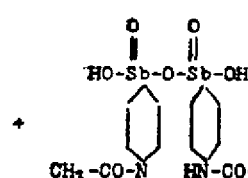
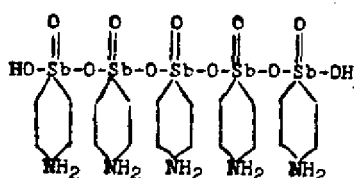
Antimony Potassium Tartrate USP
(Tartar emetic)



Antimony Sodium Thioglycollate USP



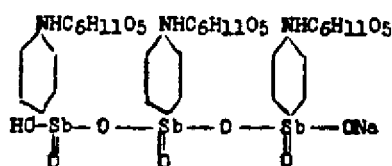
Ethylstibamine MNR
(“Neostibosan”)



+ $\text{H}_3\text{SbO}_5 + (\text{C}_2\text{H}_5)_2\text{NH}$

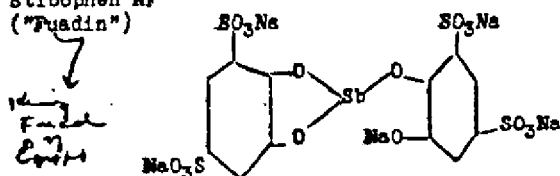


Stibamine Glucoside MNR
(“Neostan”)

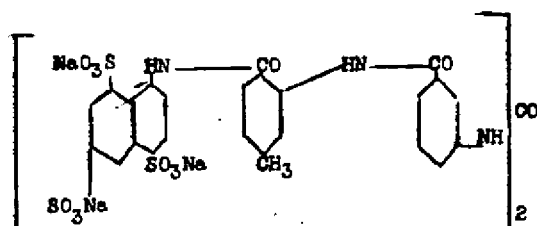


6/4/50
200

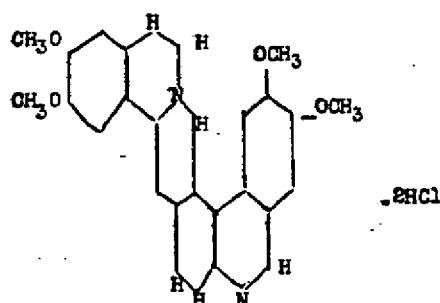
Stibophen NF
("Fusidin")



Suremin Sodium USP
(Bayer 205, Germanin, Naphuride)
 $C_{51}H_{34}O_{23}S_6Na_6$



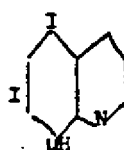
C. Against amebae
Emetine Hydrochloride
($(29H_{40}N_2O_4 \cdot 2HCl)$) USP



Chiniofon USP
("Yatren")



Diiodo-hydroxyquinoline NMR
("Diodoguin", "Yodoxin")

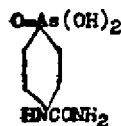


Iodochlorohydroxyquinoline NF

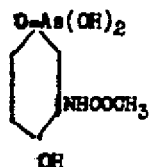


6/6/50
JSC

Carbarsone USP



Acetarsone NF
("Stovarsol")



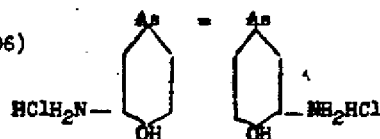
II. Against Syphilis

A. Preferred treatment at present is Penicillin (see Antibiotics under Bacterial Chemotherapy)

B. Organic Arsenic Derivatives

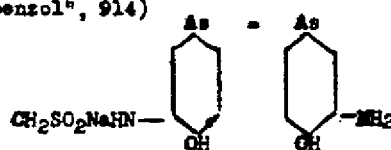
Arsphenamine USP

("Salvarsan", "Arsenobenzol", 606)



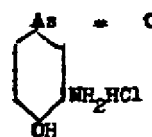
Neocarsphenamine USP

("Neosalvarsan", "novarsenobenzol", 914)



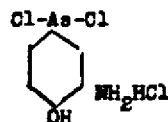
Oxyphenarsine Hydrochloride USP

("Mapharsen")



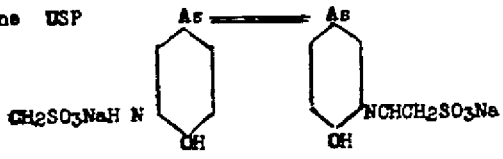
Dichlorophenarsine Hydrochloride USP

("Chlorarsen")

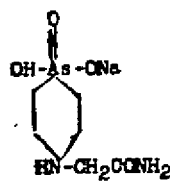


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20

Sulfarsphenamine USP

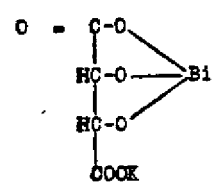


Tryparsamide USP

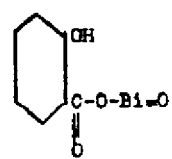


C. Bismuth Preparations

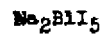
Bismuth Potassium Tartrate USP



Bismuth Subsalicylate USP



Iodobismuthite Sodium NNR
("Iodobismitol")



Sobisminol Mass NNR

Bismuth Camphocarboxylate NNR

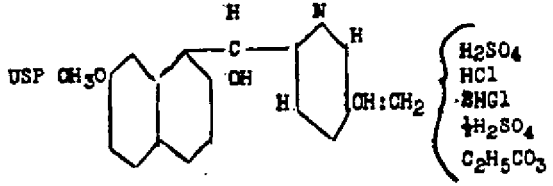
Bismuth Ethylcamphorate NNR

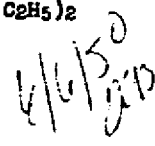
Bismuth Sodium Triglycollamate NNR

Quinine Bismuth Iodide NNR

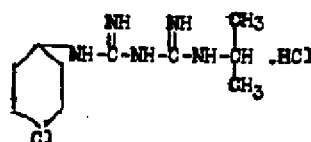
III. Against Malaria

Quinine (Bisulfate)
($\text{C}_{20}\text{H}_{24}\text{N}_2\text{O}_2$) (Hydrochloride)
(Dihydrochloride)
(Sulfate)
(Ethyl carbonate)

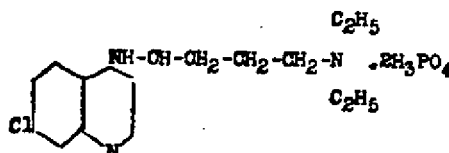




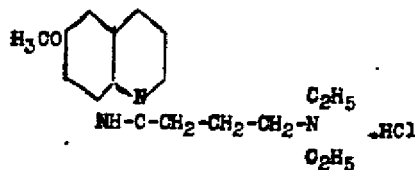
Chlorguanidine Hydrochloride NNR



Chloroquine Diphosphate NNR
("Aralen", SN7618)



Pamaquin Naphthoate NF
("Plasmoquin", "Aminoquin")



IV. Against Bacteria

A. For local use (antiseptics, disinfectants, germicides)

1. Phenol, cresol, resorcinol, picric acid etc.
2. Alcohols (ethyl, isopropyl) *and water the EtOH*
3. Aldehydes (formaldehyde and methenamine) *formaldehyde & methenamine*
4. Acids (HCl , HNO_3 , H_2SO_4 . Particularly H_3BO_3 , benzoic, salicylic, acetic and mandelic acids)
5. Halogens and halogen - containing compounds.
Chlorinated lime, sodium hypochlorite, chloramine-T, dichloramine-T, chloroacodin ("azochloramid"), halazone, succinylchlorimide. Iodine, iodoform, thymol iodide, iocamfen, vioform.
6. Oxidizing agents
(Peroxides of hydrogen, sodium, calcium, zinc. Perborates. Permanganates. Chlorates.)
7. Heavy metals and derivatives.
Inorganic mercuric chloride, oxide, cyanide, iodide.
Organic - mercurochrome, merthiolate, meroresin, metaphen, merphenyl.
Silver, silver nitrate and picrate. Colloidal silver protein, chloride and iodide
Zinc oxide, chloride, sulfate and stearate.
Copper, copper sulfate.
8. Surface-active (detergent) agents
Benzalkonium chloride USP ("Zephiran")
Cetyl Pyridinium Chloride MNR ("Ceepryn")
Benzethonium Chloride MNR ("Phemerol")

was as
normal as
used to be
by next in
somewhat
orthodox
investigations
that were

4/6/50
JLB

Hexylresorcinol USP
Sodiumtetradecyl sulfate MNR

9. Dyes

Azo compounds

Scarlet Red NF (Sudan IV)

Dimazon, pyridium

Acridine derivatives

Acriflavine NF

Dymixal MNR

Proflavine NF

Triphenylmethane (Rosaniline) Derivatives

Methylrosaniline Chloride USP

(Gentian violet, methyl violet, crystal violet)

Carbolfuchsin MNR

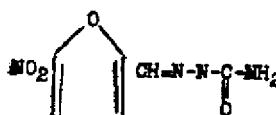
10. Miscellaneous

Sulfur and derivatives.

Volatile oils, camphor and thymol; chlorbutanol. (certain)

Nitrofurazone MNR

("Furacin")



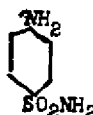
with of course...
...quite effective

Tyrothricin (see Antibiotics)

B. Antibacterial Agents Used Systemically

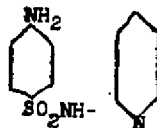
1. Sulfonamides

Sulfanilamide USP



Sulfapyridine NF

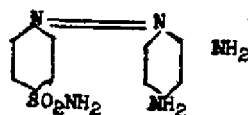
("Dagenan", M and B693)



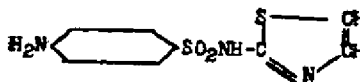
Prontosil NO

("Streptozon", "Rubiazol")

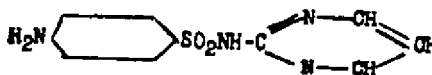
fresh
or
your



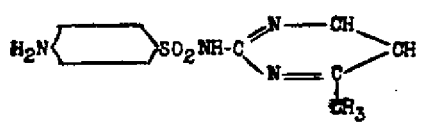
Sulfathiazole USP



α -Sulfadiazine USP

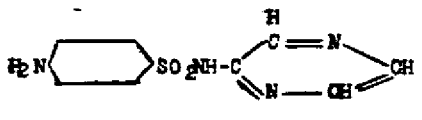


4- Sulfamerazine USP

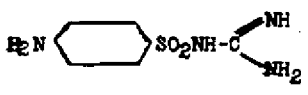


6/4/50

6- Sulfapyrazine MNR



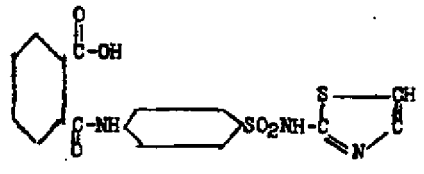
2- Sulfaguanidine USP



4- Succinylsulfathiazole USP
("Sulfasuxidine")



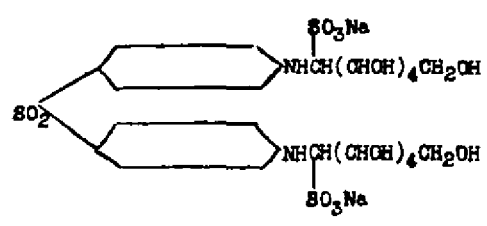
7- Phthalylsulfathiazole MNR



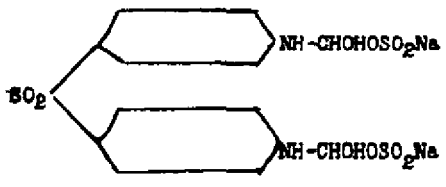
9- Sulfamylan NO
("Marfanil")



4- Promin NO



2- Diasone NO



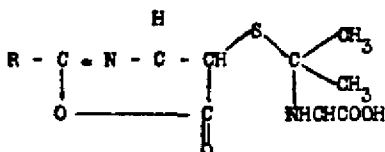
7- Promizole NO
Paraaminobenzoic acid MNR
("Paba")



6/4/50

2. Antibiotics

Penicillin USP



In Penicillin F (British Penicillin I) $R = CH_3-CH_2-CH = CH-CH_2-$

In Penicillin G (British Penicillin II) R = CH₂ -

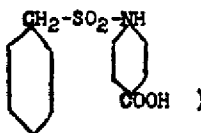
In Penicillin I (British Penicillin III) R = HO-CH₂-

In Penicillin K (No British equivalent) $R = (C_6H_{13})CH_2-$

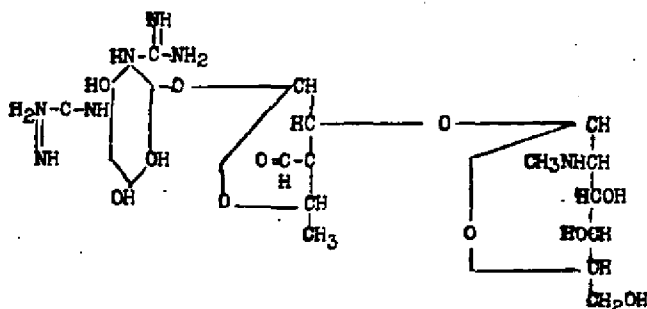
Used as Na or Ca salt

(To slow renal excretion of penicillin:

Caronamide ("Retentin") NO



Streptomycin INR



Streptidine

$$\text{C}_8\text{H}_{18}\text{N}_6\text{O}_4$$

Streptose

$$C_6H_{10}O_5$$

Methylglucosamine

$$\text{C}_7\text{H}_{15}\text{NO}_5$$

Used as Sulfate or Hydrochloride

Tyrothricin INNR

(Mixture of polypeptids. Used locally only)

Bacitracin NO (Polypeptides)

Aurocmycin NO

("Duomyoin")

Chloromycesin NO

Aerosporin (Polymyxin) NO

6/6/20

Therapeutic Value of Principal Antimicrobics

<u>Penicillin</u> <u>Best agent vs.</u> <u>Streptococci</u> <u>(hemolytic and anaerobic)</u> <u>Clostridia</u> <u>Pneumococci</u> <u>Meningococci</u> <u>Gonococci</u> <u>B. anthracis</u> <u>Spirochetes</u> <u>(syphilis, yaws)</u> <u>(moderately effective vs. Adihomycosis)</u>	<u>Bacteraemia</u> <u>Like penicillin in general, but early promise in syphilis not borne out (probably valueless here).</u> <u>Effective vs. Some penicillin-resistant strains. Also effective vs. amebiasis (value here uncertain)</u> <u>Difficulties over quantity production and toxicity (damage to liver and kidney reported in animals and man)</u> <u>These may or may not be due to impurities.</u> <u>(wait and see)</u>	<u>Streptomycin</u> <u>Best agent vs. Tubercula</u> <u>Miliary T-b</u> <u>and T-b meningitis</u> <u>E. coli infections (urinary tract peritonitis, etc.)</u> <u>H. influenzae infections (pneumonia, meningitis)</u> <u>Moderately Effective vs. Pulmonary and renal T-b.</u> <u>Penicillin-resistant strep., staph.</u> <u>Questionable vs. Bacillary dysentery</u> <u>Typhoid</u> <u>Undulant fever</u> <u>Chronic T-b</u> <u>Ineffective vs. Amebas.</u> <u>Clostridia</u> <u>Blackhead</u> <u>Malaria</u> <u>Ule. colitis</u> <u>Viruses</u> <u>Spirochetes</u> <u>Toxicity</u> <u>Like penicillin, more serious: injury to auditory nerve (may be permanent or even permanent).</u>	<u>Aureomycin (Diamycin)</u> <u>Preliminary data indicates great value vs. Typhoid</u> <u>Rocky Mt. fever</u> <u>Undulant fever (acute and chronic)</u> <u>Questionable vs. Virus pneumonia</u> <u>Typhoid fever</u> <u>Toxicity</u> <u>Only temporary anemia thus far reported.</u>	<u>Chloromycetin</u> <u>Preliminary data indicates great value vs. Typhoid</u> <u>(typhus, scrub typhus)</u> <u>Rocky Mt. fever</u> <u>Undulant fever.</u>	<u>Aerosporin (polymyxin)</u> <u>Preliminary data indicates great value vs. Perfringens</u> <u>Typhoid</u> <u>Influenza</u> <u>Colon infection</u> <u>Local use only (ineffective by mouth, dangerous by injection).</u> <u>Value demonstrated vs. Ulcers of skin</u> <u>Mastoiditis</u> <u>Empyema</u> <u>Sinusitis</u> <u>(Local bactericidal use as good or better.</u>
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Lowest of all chemother. agents. Sensitivity reactions (skin rash, asthma, etc.). No serious results reported.

MAJOR ANTIBIOTIC AGENTS

NAME	SOURCE	DATE	SPECTRUM	PRACTICAL VALUE
Bacteria				
Pyocyanase	Pa. Aeruginosa	1889	Gram pos. and neg. cocci and rods	None (very toxic)
Pyocyanin	Pa. Aeruginosa	1924	Mainly Gram pos.	None (very toxic)
Tyrosinase	B. Brevis	1939	Gram pos. and neg.	++ (local use only)
Gramicidin S	B. Brevis	1944	Gram pos. and neg.	++ (local use only)
Neotetracin	B. Subtilis	1945	Gram pos. and neg. cocci	+++ (production troubles)
Subtilin	B. Subtilis	1946	Gram pos. and neg. cocci, diph., +b.	++ (early information)
Bumycin	B. Subtilis	1946	T-b, diph., fungi	++ (local use only)
Aerospirin (Polymyxin)	B. Aerospira (polymyxa)	1947	Gram neg. rods	+++ (preliminary)
Molds And Fungi				
Pentothill acid	Pen. Puberulum	1915	Gram pos. and neg. cocci and rods	None (no information)
Pentothill	Pen. Notatum	1929	Gram pos. and neg. cocci rods, spiro.	+++ (unquestionable)
Citrinin	Pen. Citrinum	1931	Non-selective	None (very toxic)
Gliotoxin	Trich. lignorum	1936	Gram pos. cocci, fungi	None (very toxic)
Claviformin	Pen. Claviforme	1942	Probably identical, active	None (toxic - may have some value for local use)
Clavadin	Asper. Clavatus	1942	Gram pos. and neg. cocci, some fungi	None (toxic - may have some value for local use)
Patulin	Pen. Patulum	1942	Mainly Gram pos.	None (toxic to liver)
Pumiligin	Asper. Pumilatus	1942	Gram pos. cocci and rods	None (toxic)
Aspergillilic acid	Asper. Flavus	1945	Similar or identical	None (no information)
Flavodin	Asper. Flavus	1945	Activity like Pentothill	None (no information)
Flavocidin	Asper. Flavus	1945		
Gigantic acid	Asper. Giganteus	1945		
Cinetomycin	Cinet. Cochliodes	1944	Mainly Gram pos. cocci and rods	None (inactive in vivo)
Actinomyces				
Actinomycin	Act. (undent.)	1924	Gram pos. and neg. cocci	None (toxic)
Actinomycin	Act. Antibioticus	1940	Gram pos. cocci and rods	None (toxic)
Streptothricin	Act. Lavendulae	1942	Mainly Gram neg. cocci and rods	++ (local use only)
Streptomycin	Act. Griseus	1944	Gram pos. and neg. cocci and rods, T-b.	+++ (toxicity?)
Neomycin	Neo. Coelatae	1946	T-b Group	++ (no information)
Chloromycetin	Streptomycetes--?	1947	Typhoid, typhus some virus uses	+++ (preliminary)
Aureomycin	Strept. Aureofaciens	1948	Lymphogran. Bruceiellae, virus pneumonia	+++ (preliminary)
Actidione	Strept. Griseus	1948	Pathogenic yeasts	++ (no information)
Others				
Lysozyme	Animal cells	1922	Gram pos. and neg. cocci and rods	None (glycolytic enzyme)
Chlorallin	Algae	1944	Gram pos. and neg. cocci	None (no information)
Canavallin	Jack beans	1944	Gram pos. and neg. cocci	None (no information)
Allidin	Garlic	1944	Gram pos. and neg. cocci	None (no information)
Tomatin	Tomato	1946	Pathogenic molds and fungi	None (no information)

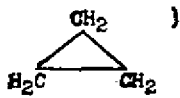
REPRESENTATIVE DRUGS USED CHIEFLY FOR ACTIONS ON CENTRAL NERVOUS SYSTEM

(USP = U.S. Pharmacopoeia XIII; NF = National Formulary VIII; MNR = New and Nonofficial Remedies 1948; NO = Not Official)

I. Depressants of Nervous System

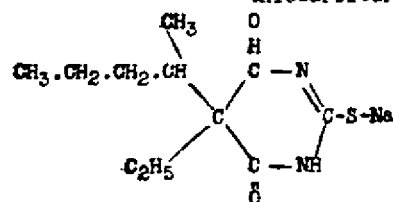
1. General (as opposed to Local) Surgical Anesthetics

A. Inhalation Anesthetics

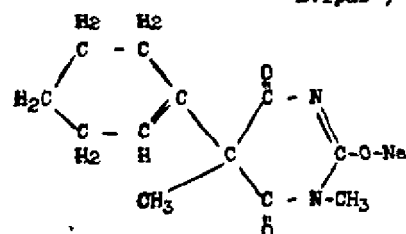
<u>Ether</u> (diethyl oxide, $C_2H_5.O.C_2H_5$)	USP
<u>Vinyl Ether</u> (divinyl oxide, $CH_2:CH.O.CH:CH_2$)	USP
<u>Chloroform</u> (trichloromethane, $CHCl_3$)	USP
<u>Ethyl Chloride</u> (monochloroethane, $CH_3.CH_2.Cl$)	USP
<u>Nitrous Oxide</u> (nitrogen monoxide, N_2O)	USP
<u>Ethylene</u> (dimethylene, $CH_2:CH_2$)	USP
<u>Cyclopropane</u> (trimethylene, )	USP

B. Intravenous Anesthetics

<u>Thiopental Sodium</u> (Monosodium salt of 5-ethyl-5-(1-methylbutyl) thiobarbituric acid-"Pentothal")	USP
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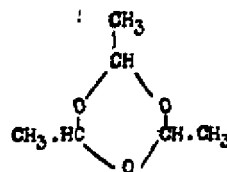
<u>Hexobarbital Soluble</u> (monosodium salt of 1,5 dimethyl-5 cyclohexenyl barbituric acid-"Evipal")	MNR
---	-----



C. Rectal Anesthetics

<u>Tribromoethanol</u> (tribromoethyl alcohol, $CBr_3.CH_2.OH$)	USP
--	-----

<u>Paraldehyde</u> (paracetaldehyde)	USP
--------------------------------------	-----



11-29-48

②
7 main types of substances available for renal clearance
(no renal excretion)

1. mannitol, Urea, creatinine, Na_2SO_4 D₅ (100%)
2. cleared completely by kidney (eliminated 100%)
 { p-aminosalicylic acid
 Diodrast (organic iodine and etc. 50% I. by and)
 90-95% cleared.

anti diuretics - check secretion of aldosterone
 diuretics - >

and of
 still use AV and theine & theophylline diuretic & NaCl but
 mostly are osmotic.

Skatol Vasodilator (blood vessels)

1. CO on brain & constricts blood vessels elsewhere in body.
2. Paralysis, Kellin & Kellidin dilate coronary blood vessels more than they do elsewhere - within arterial.
3. Cytochrome (no bold name & etc) - but controlled tests failed to > deficient blood flow in heart.
4. need vasodilators mostly in occlusion of arteries.

Autonomic Pharmacology

ch 13

1. Drug Chaperones (mimic activity effects of autonomic)
2. auto neural cells (as well as, 1,000 & autolytic)
3. drugs display specific antagonisms
4. blood problems produce a different type of problems of all pharmacology
 also:
 stimulation of blood in nervous system - of autonomic
 group of neuroleptics

11-69-48

6/6/50

note 5 & P₁ have homologs in administration of H₂O —
liver & di. excretion.

Part 2

in fact
untrue

Now blocking agents are supposed to act (analogous to a) dark room, etc. antagonist compounds with same receptor as which agonist and but only prevents reaction - no other agonist effects, and in

⑤

11-29-43

P-5
Set attention - barbers
eyeballs - asphalt chemical gates (Toungate)

sig. glands
attention (yawning)

Resist. + rest
muscle (hunched)
attention (H.P. hind)
concentration sturdy < sturdy >

Ident
as is

slow dent - set more involved as good as details

slow very
skin as quick

* if present all only way for these involves to all these

now 12 Cynopsis - Drugs

① sketch 4 plains (H.P. Ford) - late in 1944
extracts
clinical supra clinical glands - study published in 1945
good > destruction
at 4 P

② discussed in 1947 - for the clinical travelling
from clinical course

Pharmacology propose monomer - due to dog - dog did
not identify the drug 14 months later
as first looked as these "unidentified"

③ Ephedrine - strong test of the drug - under drug
dr. could have had synthetic drug in place of natural drug
ago by pharmacology experiment

(6) 11-24-48
 Benzylamine - inferior to epinephrine
 ↓
 does not raise b.p. as markedly
 does epinephrine
 but less harmful to g.d.

2. Tyramine - aliphatic
 naphazoline - histamine-like effects.
 N.O. = non official

- P3
- ① not potent
 - ② N.O.
 - ③ found naturally - formed by decarboxylation of Tyrosine
 → thought for a while to be cause of essential hypertension
 - ④ not longer there
 - ⑤ according to recent reports is quite potent
 OH groups → brief effects
 alkyl → longer effects

epinephrine - the cause of a O.H. rapidly oxidized in body
 epinephrine - stable in body, less potent, & handled by
 diff. enzymatic systems than epinephrine

Uses of Sympathomimetic Drugs

1. Greater practical value - add to local anesthetic
 when to delay the absorption into system - i.e.
 get vasoconstriction effect to delay absorption
 system & prolong effect - epinephrine in
 the same amount

⑦

11.29.48

- ✓ Second use - To relax bronchial spasms (To relieve asthma) - Epinephrine will relax most bronchial spasms - except that after a while its effects become nil.

Can be taken by mouth only as ^{water} solution - otherwise ^{with air} ~~can be taken by mouth~~ but effects trouble with both is in that > heart beat & raises d.p.

As far as we know, which acts specifically on bronchial muscles & still has no bad effects & can be taken by mouth

2. decrease congestion of membranes of nose - Epinephrine
 too much in children (~~it is~~ epinephrine in oil)
 → potency pneumoniae
epinephrine inhalation convenient

use of Epinephrine inhalation may actually be due to decongestion similar to epinephrine inhalation
 ✓ in asthma

3. use effect on blood vessels in liver and kidney
 - to keep b.p. from falling too low.
epinephrine, adrenaline, etc.

these
strong drugs act

1. stimulate sympathetic blockade - but this is on a low plane
 2. may alter rate of sympathetic (destruction of) - not proved
 3. most probably a drug simply replace sympathetic combination with fixed receptors

11-9-48

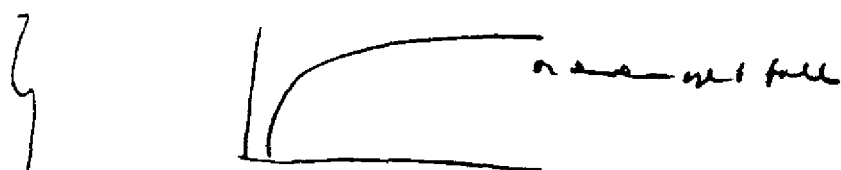
(2)

How Test Effects of these Drugs (Circulation, Excretion & 2)

1. Smaller (4 test) - measure rise in b.p. of anesthetized animal or with (indirectly)
2. measure effect on isolated intestine 10/15/48
 sympathetic - \leftarrow inhibits activity

3. Effects on blood vessels - vasoconstriction
 - inject drug into artery while measuring platysma muscle
measure signal of arm or leg.

to chlorpheniramine - loss of effect on \rightarrow dose



one of difficulties in testing these drugs.
 How get around - keep dose down to minimum (4 still get effect)
 rise in arterial tension - come in for short period.

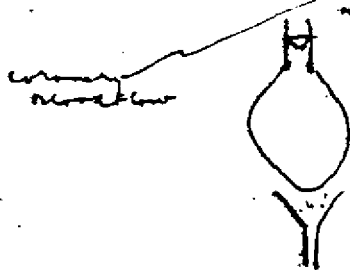
4. uterine muscle record (in dose of ?)
 rat & cat - inhibited
 rabbit - \rightarrow
 5. bronchial muscle
 6. salivary gland of cat - very characteristic response in mucousy saliva
- next sympathetic blocking agents

11-22-49 Action of Drugs on Heart

Methods

1. Perfusion (one of oldest & best) - remove from animal's body (by perfusion?) & continue to beat as perfused - use modified Langendorff's bath (balanced salt solution with various ions & Ca & Mg & K & some buffer?)

2. Perfusion of mammalian heart - use mainly for action of drugs on coronary artery - use Langendorff method



use heart of rabbit, guinea pig, cat, dog

3. Heart lung - method - remove blood from heart & pass it through a lung

Other - Quantitative

1. ECG - as usually with figure of heart beat

no loss as to amount of blood put out. no advantage - no oxidation of any

2. Infundibulograph - record heart



3. Cardiometer - put heart inside glass or steel rigid chamber



orientation of the recording device is to be decided

2. Cut out Paravertebral muscle - exposed in same way as ?
 a. duct by ?

most ventral side of all of last four

3. In vivo measurements of cardiac output

a. Fick principle - determine O₂ intake of blood
 in flowing through.

Ex

O₂ intake = 250 cc/min

(1)

(a. venous) venous blood = 15% O₂

(2)

(a. arterial) arterial = 20% O₂

(3)

av. diff = 5% O₂

$$\text{So } \frac{250 \text{ cc/min}}{0.05} = 5 \text{ l/min}$$

∴, need to collect arterial & venous blood

only venous blood - collection of mixed venous blood

∴, ventral venous, put needle in at vent. of heart & collect (see sketch)

c. now with cardiac catheterization

c. arterial blood - real problem (cause of blood coming
 by all arteries)
 diff from venous blood

but is method of choice

d. Ballistocardiograph - ideal for all standard tests but
 are - accuracy.

Flashed red (wood) on which, water like - measure wave
 front of beat (used, also in air transmission devices)

Min

calibrate empirically (see sketch)

③

4-22-48

4. For dose for type of normal value \rightarrow formula (unavailable for normal heart, but not abnormal kind) any calibration method would

6/4/50
9/50

disadvantage of Fick principle is 10 min time required for assay.

10. C.B. is gd because it measures each heart beat by itself - so gd for very rapid & advanced drug methods.

7. Stewart (or dye-dilution) principle - calc from color diln (artificial blood) - Has been used with dyes & more recently NaCl (measure increased electrical conductivity)

8. Potentiometer - Krombach (Tamm et al.) - Fall to use of measure of shadow of heart - measure electrostatically - very special method, very complex & requires skilled operators

9. methods for measuring cardiac efficiency

Energy \rightarrow $\frac{\text{work done}}{\text{unit of energy}}$

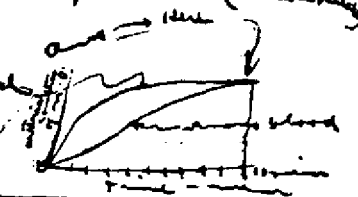
Energy used = 0 - used by heart muscle

Volume of cor. blood flow \times $(V_1 - V_2) = \text{vol of blood flow}$
Volume of A-V diff

Let efficiency tell you what $\frac{1}{2}$ - now can measure in dogs & (in last yr) in humans - made possible by insertion of catheter into coronary sinus (measuring by fluoroscopic control)

For measurement - Kety method

1. At catheter 15% nitrogen bubble
2. to be for analysis
3. to be for analysis
4. to be for analysis



$$\frac{A-V \text{ diff}}{1.75 \text{ cm time}}$$

11-22-48

11/2/50

are integral or diff

to can use in interval dog or man to measure mechanical efficiency of heart.

results from no for - can't control control period.

But action of drugs on heart involves abnormal conditions

1. Digitalis has effect of non. heart just opposite to that of drained heart \rightarrow decreases cardiac output, ex. adrenergic

But for patients with heart disease (stroke heart, decomp. etc) - & get increased output & better breathing.

So this dig. was studied before other conditions are taken into

2. Reason for "1" - heart & failing - great Cng. Phosphoryl

more the more the more the more blood is in the heart (within certain limits)

\rightarrow at outside & sides up into stretch the contract (like with rubber bands)

\therefore dig. stretches heart muscle too much when it is again efficient

or non heart go beyond limits.

3. What dig. does

a. decreases output of failing (or decomp.) heart

b. slows down heart rate

(1) \rightarrow activity of cardio. center

(2) modulates block bet. auricles & ventricles.

\rightarrow that some nodes
more in order

C

Q

11-22-43

1. auricular fibrillation - may be 1000 or 1500 times/min.
for auricular rate & time (some of them) get thru to ventricles
& 7 kind pulse that is. Pulse goes from 125 to 150 to 20 to 25.
For most the slowing is much less and the rhythm doesn't run.

2. Dig not effective when
a. cardiac decompensation
→ auricular fibrillation

6/6/50
JTB

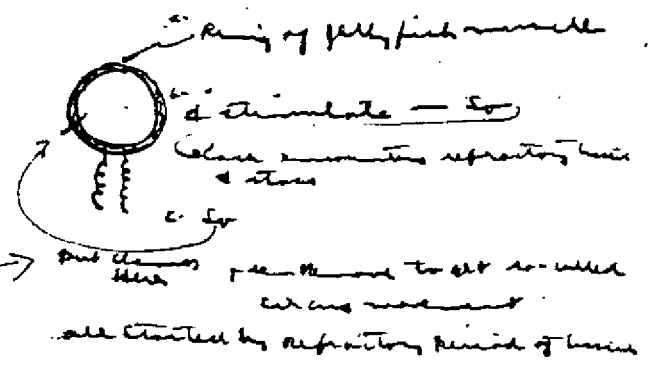
But Dig. does not remove auricular fibrillation but does
make ventric. rate slow.

4. But Quinidine will alter auricular fibrillation
→ and/or atabrine (resort)

5. 1/2 of coils of auric. Fib. stored by quinidine
(but use cautiously for already diseased heart as quinidine
↓ depresses for heart muscle)

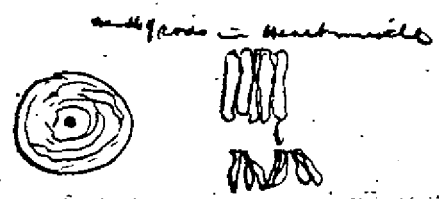
Experiment of How Quinidine & atabrine work in auric. Fib.

↓ of Sir Thomas Lewis
as done in this country



to block
conduction of
nerve impulses that
carry equally &
spread out.

to in Human heart



again
a. block to impulse
b. Atrioventricular node is slowly
↓

(6)

11-22-49

1. How do these drugs act on this phenomenon

CHP

1. big dose stops
2. but Q prolongs refractory period & won't respond & fibrillation dies out

WFO

some drugs increase excitability of heart muscle very markedly.

to when ventricles fibrillate then get death (Fib. of ventricle is irreversible & fatal) - usual cause of death in electrocutions.

1. chloroform (in low concn) > reduce excitability of heart muscle & when get 2 output then can get
↓
main
excite
etc

ventricular fibrillation

by 1. toxicology & 2. removal adhesion on heart

b. failure of one into adhesion

as calculated
outlet.

c. something can happen under benzol & also under cyclohexane anesthesia

2nd

list page 2

1. chloride output & other stuff
2. Ventricular fibrillation (Vfib) - fibrillation much more than they are now
- arrhythmic mechanism - normally some disease (acute coronary)
- 1st lot of quinidine leads to lower so content of blood. 2nd lot of quinidine leads to lower so content of blood.
3. The above - appear in distinction with other from these

4/12/00

- 1. ^{also} decrease etc
- 2. K⁺ salt decreases heart muscle -
 - helps to stop ventricular
 - fibrillation as decrease tendency to stop ventricular fibrillation

5. 1. A-V block (currently - anterior block)
2. ~~anterior~~ not inside fibres themselves

(not on charts) showing tendency now

For Completed Heart Failure (absence of clots & too little volume)

not organic material to produce a proper division
satisfactory - does not find to account for extra
work of that muscle.

Computer Means Failure - too much flims for least to handle

D. ~~Pituitary tumor~~ produced strong constriction of
growing artery.

continued 1. But do not give pituitary press. & w. — do not
expect. (i.e., as in contraction of uterus after child-
birth to stop hemorrhage.)

to idea is water out for the actual answer is.

- Dilatation of heart muscle - N arteries with abnormalities of coronaries

cholesterol derivatives. (and cholesterol etc. other formulae)

~~papaverine~~ - opium alkaloid

synanthropic - hydric str.

all of above - continuous blood vessels on whole, found only in
invertebrates

Don't know - Drugs which dilate blood vessels usually
are usually disincenting bc cause of their overall effects.
Giving rules to other side of H2O.

② By nitroglycerine - dilatatory coronary arteries was only a small effect of 7 mm of Hg.

11-22-47

Notes

→
Action of Drugs on Peripheral Circulation

which is possible - Vol of blood in Tends to keep at level only
Tendency of ^{arteries} is maintained - i.e. pressure
constricts itself
attains its approach - for blood

For Heart - coronary branches → Angina pectoris
in heart due to pressure

For coronary occlusion - heart muscle will degenerate &
get infarction.

For skeletal muscle - can get bad stuff in legs to
get gangrene.

For kidney - renal circulation - with blood supply in
which get substances produced which give &
maintain life & p.

to,

Two procedures

can

1. > Blood supply - attracted to arteries & physical but
has only limited therapeutic value.

difficulty - vasodilator drugs act too profoundly,
only spasmolytic (K⁺) are constrictor elsewhere & danger
to coronary

Obstruction to →

note (relaxation - organic obstruction hindering of blood supply)

methods for measuring Vol of blood flowing thru
individual. Think of all part of body - to
find out how output is distributed.

1. In animals - most direct way, put in some sort
of flowmeter & pass blood thru & then back to
body

renal circulation - pressure of urine - so -
directly from vessels

Experiment - R.O.
and location for dist
to

a. Thermocouples

was - control flow & < . to see how gradient bet
skin & blood, when >
flow & tissue (collection
skin is constant)

⑨

11-22-48

but must measure in air. Temp Room & arterial activity
from chemotaxis neuroendocrine ^{of subject}

transmittable
not function

M

6/6/50
JAN

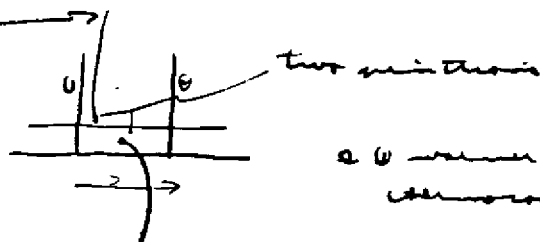
Can we only for direction of action of

2. Variation on "1"

transmittable "Phine" in memory

body doesn't really do what it should

the temp
current
to
sent blood
indication



a 0 warmer than 1 & at under than
chemotaxis

Difficulty here - could not control the temp
current & other difficulties.

Phine why can't we be building with
even phenomenon used word.

3. major clinical - use Hairer Heater to test blood
quality.

but all of "1" & "2" are same condition - diff in
to set two functions depends on many conditions
which diff the blood 1

phenomenon used word - across in use to calibrate
thermometer.

Form 101

PHILADELPHIA GENERAL HOSPITAL
HEART STATION

Name _____ Dept. _____ Date _____ No. _____

Last Name First

⑦

(C)

11/15/47

12-15-47

p. 15 d. 4 d. - Can actually cure malaria - but
can be toxic - do not get released.

CEPH. - work on dechlinings (not on man)
also Phenol mixtures (close to malaria to Ch. 1. 1/2 way)
ultimate Ex. 1 on man (red. P. at 1000, 111.)

Really definitive Expt. - report of complete control.

actually value of this work during war did not relate
perhaps much to malaria as to that of Ch. 1. (study of
surgical systems)

2nd 140

p. 15

Vs. P. 1. 1. 1.

Pasteur - really killed by French Nurses.

1st result of

antiseptic use of surgery (Lundholm's)

sanitary of D. 1. 1. 1. - Expt. of Ph. 1. - same
much more.

now

antiseptic use - Nurses, surgeons etc. sterilized etc.
do not see infection of wounds as old - healing
by 1st bacteria

1. H. 1. 1. 1. 1.

Warrior food

Expt. - local application

2. No. 1 - on skin is Expt. of W. 1. 1. 1.

Expt. - Expt. of water Expt.

Ex - used for Expt. - Expt. Expt. Expt.

3. Ex - used for Expt. Expt. Expt.

4. Expt. Expt. Expt.

5. Expt. Expt. - Expt. of W. 1. 1. 1. Expt.

6. miscellaneous

12-3-48

6/2/50

p. 6

B dystonias A & C agents
(one of more millions chapters)

Production of ^{oil} drugs (dystonically) produced at American
Drug Works.

Frederick's Paper at Zimeldrop

Am: on the stage, on (now called a paratone) - and other
then Chloroform in

Am: the med. work Feb. 1935 (Hundreds of Patients)

Really tried first on Patients

Epinephrine - excited in clinic - not test/ready
↓
further work

1. - instead of most reliable but

1. - methemoglobin (cyanosis) in blood

not down in vision light.

Epinephrine
only 4 cases of future Handed - 1st effective

chemotherapeutic agent as pneumonia - but

→ vomiting - some discoloration.

Epinephrine - largest spectrum of metabolism

but is also one of most toxic

highly metabolized by

B. 1

a, b, & c.

Epinephrine - the antibiotic

1. cannot be used on skin - critical to not use and may
of maintaining ^{Epinephrine} blood drug -

2. sedation for getting blood into blood in di

(6)

12-13-48

17

d

e

f

not obs. in intact tract.
seen though in the resp. v.

1/1/50

9 not antagonized by p.a.b. acid
while it is all material in best for
infected wound.

not used in t.b. & in deproy, particularly
+ f

activity in sulfonamides etc now finished - so
fleming & (F. deproy) had difficulty in manufacturing
mil. manufacturers.

now trend back to S. & T. and develop with
S & antibiotics.

1. pure hipp
2. with C.H. H. H.

Sulfonamides (penicillins & strepto)

1. when used in human medicine they are
bacteriostatic & not bactericidal & not
bacteriophages

in humans in
which can be used

2. Log phase on their action - bacteriostatic action
not exerted immediately.

3. complete ^{more bacteria for} ~~agony~~ some essential and toxic to which
necessary for the proliferation of the bacteria
↓ with longer system in the bacteria
by p.a.b. acid.

Explanation of Log Phase.

exists form of explanation - not really, as, as
we are deproy. longer system
and - as by constitutive inhibition.

12-20-48

antibiotics - anti(microbial activity)
(anti)

- bio (of biological origin)

etc

1. extreme potency
2. hi-specificity
3. extremely low toxicity

development (or history) of antibiotics

1. 2nd world war
2. arose to situation where many body accidents were
3. need for something in nature: answer.

Penicillin - How could organisms do it now
(after penicillium mold of penicillin)

1. anthrax bacilli in soil disappear in few weeks
2. sp. something in air to which destroy pathogenic organisms

1899 - Chamberlin & (Lowe?)

penicillium pyogenum secretes substance which is toxic to ?

1929 - Fleming's famous contamination
(considered religious turn of mind)

med. officer in WWI. - heavy respiratory disease due to infected wounds

penicillin in WWI

1930 - "mold" produced by *Penicillium*

biological - product of animal cells
↳ proteolytic enzyme

1940 - clinical investigations of

1. clear space on petri dish culture contains by air bound mold
2. cultured ~~clonally~~ mold → penicillin production

yellow extract

rose in England
one of few that → ~~antibiotic~~
substance

not so same subformic acid → 10000

not Fleming's mold

found out that penicillin killed others in mouth but not
multiplying bacillus

to use to exp. i.e. from mold substance

12-20-48

- 4/4/50

appetite atrophied
 gone
 etc

1. ^{name} ~~Excretion~~ can make it bactericidal & not merely bacteriostatic
2. Toxicity of P is much lower than that of sulphonamides
3. none of Toxicity reactions of P are very severe - unpleasant but not dangerous
4. broader spectrum of activity - covers almost all
 - ↳ effective *streptococci*, *gonococci*, etc
5. acts more rapidly - no lag phase
6. not rendered inactivated by penicillins & tetracycline antibiotics

Disadvantages

1. changing cell culture method of media
2. difficult to store & keep well
3. doesn't work as well by mouth - absorbed as readily as is excreted by kidney \rightarrow acid destruction
4. under ~~system~~ reaction & is destroyed by penicillins are
5. produces local irritation (i.e., added procaine)
(Gard, etc)
6. Retention - Caronamide \rightarrow connects with \uparrow for renal excretion mechanism.

Present Disadvantages

7. people are developing sensitivities (rash, asthma)
8. patients may become resistant
under a course of therapy $\left\{ \begin{array}{l} \text{population based sensitive} \\ \text{patients become resistant} \end{array} \right.$
9. useless against many ~~types~~ \rightarrow common organisms
 - a. Gram neg. rods of Typhoid bacillus
 - x - typhoid bacillus
 - c. acid fast organisms (?) - Tubercle bacillus
Lact. bacillus

Techniques for exam

- " paper chromatography
- " Craig's counter-current exam.
2. growth of ~~exam~~ mycology
 \rightarrow Green Waksman at Rutgers
 \rightarrow (1944) Streptomycin

discovered as result of patient's & painstaking search.

penic. and tubercule

Streptomycin - basic substance - soluble as salt or sulfate

only for penic.

Effect of \uparrow - acts on Typhoid like rods
(Tubercle bacillus
typhoid bacillus
typhoid bacillus)

12-20-48

U/C/50 MB

retained with the P
to need no blocking agent

vs. strepto

1. Toxicity much like penicillin
pharmacokinetics and effects

→ aggregation apparatus
aggregates
present in animals

2. development of resistance bacterial strains

meningococci { → strepto & all bacteria in
(and S)
used in (bacteria) → S
→ used (trace)

concrete tubercle bacilli infection
(but must give subcutaneously at first)

but resistance to tubercle bacilli is temporary
(most of us who I said again)

limited to acute tuberculosis
(gallstones)

not effective therapy for (tuberculosis meningitis)
(encephalitis)

2nd Hr

Baritracin (named after the little girl named Tracy)
produced by Frank Melloway (Lingen of N.Y.)

searched for antibiotics that would do for skin
the infected wounds what maggots did for
osteomyelitis

initially grown in culture which used in T.
(growth) synthesis of baritracin similar to that of strepto

slow forming rod

Has wider spectrum (includes gas gangrene bacilli)
not inhibited by penicillin
not effective as depth of wounds

(6)

12-20-68

Drugs (wide open fields for)

agents that can be used locally as well as systemically.

need

1. Effective
2. Not toxic
3. Kind of desired activity
4. Can be produced in quantity, practically

6/6/50
M

Craving not at all troubled about synthesis of the actual ingredients

Chemists - really a different way of looking at ~~the same~~ ^{new} things known previously

Wide open fields

1. Virus diseases - antimony can be used as various pneumonia
2. Use to control infections in animals & plants
3. Possibility of therapy of this type in neoplastic disease
 ↓
 (can do)
 ↓
 undesired cells
 ↓
 really not in future
 but something favorable in long run.

omitted by areas

1. Insecticides - Rodenticides
2. Poisonings
3. Biochemical mechanism of drug actions.

Taking better drugs is usually more difficult than that of producing them

also for task of explaining what they do.

Trends

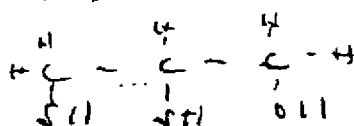
1. Biochemical Basis of Drug Actions
 a. less common as pharmacological action
 particularly on some basis other than toxic & other
 up till recently unsuccessful
 in attachment on

TO A L (Anticancer activity))
contains As

12-20-47

6/6/50
gwp

Anticancer molecule



Smith by Peter & colleagues at Oxford

SH combines with glutathione which is essential
could be activity of new era - calling shot beforehand
& timing it come true

used for Ch & Pt personally as well & brilliantly
no As.

" it can find out more about other cell mechanisms
than we now know, then can build and try
work effectively

cf Results on exposed systems, involved in
action of mutagen on cells
mutagens.

like all pure science -

" can't tell what will happen (Results unpredictable)
" side effects can be very high.

Clinical Pharmacology
methods for quant. study of drug action in man.
great financial support

U.S. P.H.S.
Rockefeller

among
animal patients and from 1944 to 1947

1. cardiac output

2. blood volume

3. rate of oxygenation

4. arterial blood flow

5. renal

kidney disease

many methods developed during war & are now in use
here.

and examples - Fresh Nutrition Research

1. As

2. As

3. As

4. As

5. As

clinical

②

10-20-68

6/6/50
for

new course - setting up organization & developing results.

Curriculum of new course

level of activity best suited to biologists

better done in kindred labs - better done than in permanent work in these labs.

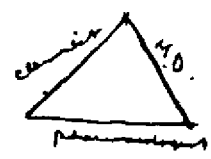
major problem is that of personnel.

Family - toxicity studies (industrial)

personnel H & H

Donors - Group

laboratory (Schmidt's Home Town) - Phasing H & H



equilateral

Excellent example of making of resources.

next steps are expansion of effort (& financial support)

1 LF 1ST 12 CR 2 151 3T 1
1 LF 1ST 12 CR 2 151 3T 1
1 LF 1ST 12 CR 2 151 3T 1

C

Q

11-29-48

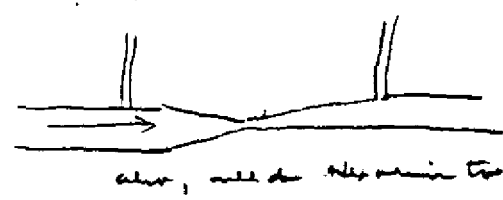
Methods for blood flow (the ones)

1. Lin. Rang. Model (for measuring venous flow)



6/4/50
JIB

2. Venturi principle

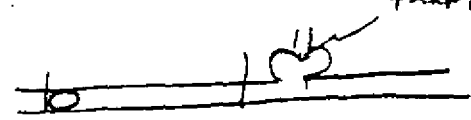


3. Rotameter - also need for low viscosity factor

4. Surface plate meter

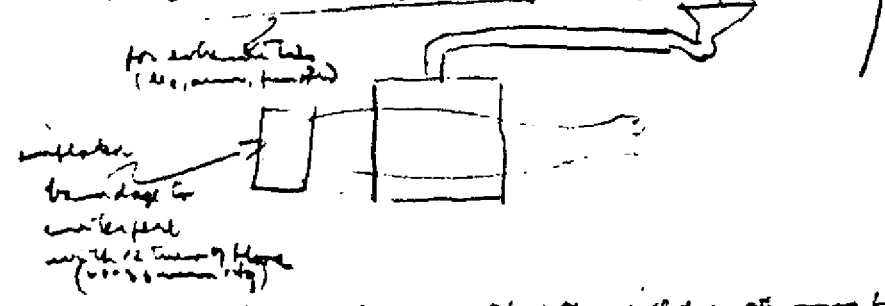
5. Bubble flow meter - also need low viscosity factor

trap for filtering out bubbles
(or that bubble is critical for measurement)



Methods for measuring, including, not for measuring

1. occlusion plethysmography



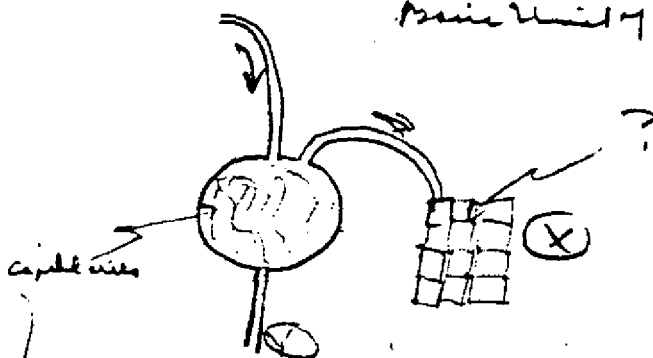
2. Kety's method - also need low viscosity factor

used for carefully to find for only transient effect - not for
the long and short of circulation

5. clearance methods (for Kidney) ⁽²⁾

11.29.47

Main Unit of Kidney Function



→ dialyze (into glomerulus) protein for filtrate

output 100-120 cc / min of glomerular filtrate

all but $\frac{1}{2}$ to $\frac{2}{3}$ / min is reabsorbed

↓
filtered urine (retrograde)

secretion in urine (Y)

Reabsorption in blood stream (X)

Cr
creatinine normally reabsorbed
creatinine excretion in urine 100%

no. of radic units = 2,000,000
in normal kidney

6. new clearance method

conc. in blood = vol of blood

= conc in urine = vol of urine

a

vol of blood = only unknown factor

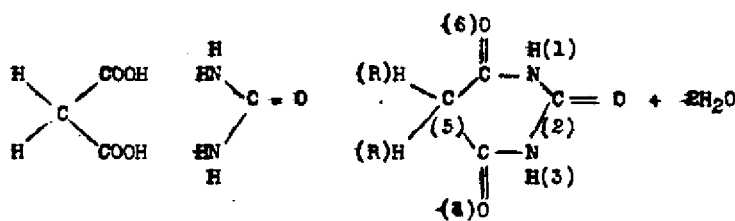
$$C = \frac{U_{\text{creatinine}} \cdot V_{\text{urine vol}}}{P_{\text{creatinine}}}$$

$$= \frac{U \cdot V}{P}$$

P (conc in blood) → really P for plasma

D. Simple Sedatives (to be taken by mouth to combat insomnia and restlessness not associated with severe pain)

1. Derivatives of Barbituric Acid



Malonic acid + urea \longrightarrow Barbituric acid

Hundreds of derivatives have been made and tested: 4 are in USP, 13 in MNR. Most important difference among these is in duration of action:

Long acting (more than 8 hours)

<u>Barbital</u> (diethyl barbituric acid-"Veronal")	USP
<u>Phenobarbital</u> (phenyl-ethyl barb. acid-"Luminal")	USP
<u>Alurate</u> (allylisopropyl " ")	MNR

Intermediate (6-8 hours)

<u>Amytal</u> (isocamyl-ethyl " ")	MNR
<u>Neonal</u> (n-butyl-ethyl " ")	MNR
<u>Dial</u> (allyl-allyl " ")	MNR
<u>Ipral</u> (isopropyl ethyl " ")	MNR
<u>Hostal</u> (B-bromallyl isopropyl " ")	MNR
<u>Sandoptal</u> (allyl-isobutyl " ")	MNR

Short (4-6 hours)

<u>Pentobarbital</u> (methylbutyl ethyl barb. acid) ("Nembutal")	USP
<u>Phanodorn</u> (cyclohexenyl-ethyl " ")	MNR
<u>Ortal</u> (n-hexyl ethyl " ")	MNR
<u>Seconal</u> (allyl-methylbutyl " ")	MNR
<u>Vinbarbital</u> (methylbutenyl-ethyl " ")	MNR

Ultra-short (less than 1 hour)

<u>Thiopental</u> ("Pentothal" - see above)	USP
<u>Hexobarbital</u> ("Evipal" - see above)	MNR

6/6/20

2. Derivatives of Ethyl Alcohol

<u>Alcohol</u> (C_2H_5OH)	USP
<u>Diluted alcohol</u> (about 49% C_2H_5OH)	USP
<u>Spiritus Frumenti</u> (whiskey-about 50% C_2H_5OH)	NF
<u>Spiritus Vini Vitis</u> (Brandy " " ")	NF
<u>Tribromoethanol</u> ($CBBr_3.CH_2OH$ - see above)	USP
<u>Chloral Hydrate</u> (Trichloroacetaldehyde, $Cl_3C.CH(OH)_2$)	USP
<u>Chlorobutanol</u> (Trichlorotertiary butanol-"chlorotone") $Cl_3C.C(CH_3)_2.OH$	USP
<u>Paraldehyde</u> (paracetaldehyde-see above)	USP
<u>Chloralose</u> (chloral + glucose)	NO

3. Sulfones, urethanes, carbamides - seldom used now.

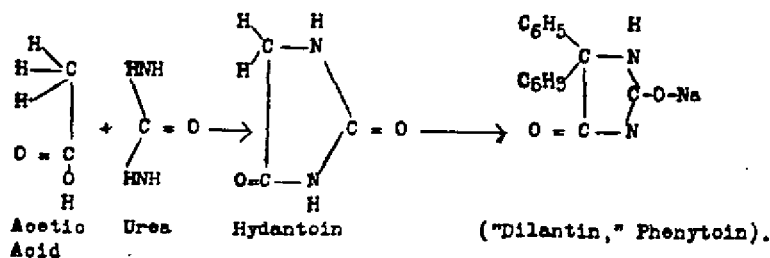
E. Anticonvulsants

General anesthetics in anesthetic concentration

Barbiturates (especially phenobarbital for chronic use, thiopental for acute emergency)

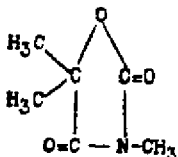
Bromides (Na, K, NH_4 , Ca) USP

Diphenyl hydantoin sodium USP



Trimethadione (3,5,5-trimethyl-2,4-dioxo-1,3-diazolidine-2,4-dione-"Tridione")

MNR



F. Analgesics

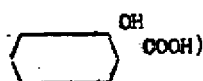
1. With effects on cerebral and smooth muscle functions - morphine, codeine, etc.

USP

6/4/50

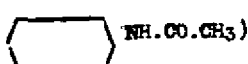
2. With antipyretic effects

Derivatives of salicylic acid
(O-hydroxy benzoic acid,



USP

Acetanilid (acetylaniline,



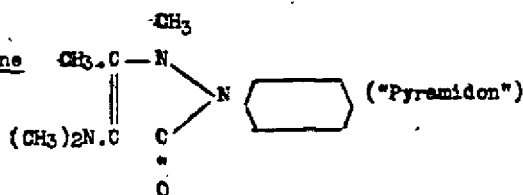
USP

Acetophenetidine, C_2H_5O



USP

Aminopyrine



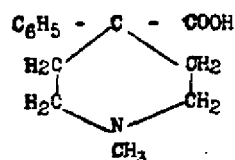
USP

(Dimethylamino dimethyl benzyl pyrazolone)

3. With antispasmodic (atropine-like) effects

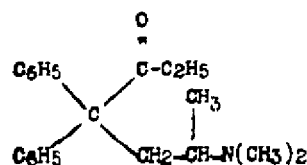
Merperidine (1-methyl-4-phenyl-piperidine-4-carboxylic acid.
Isonipocaine, pethidine, "Demerol", "Dolantin")

NNR



"Amidone" (1,1-diphenyl-1-(dimethylaminoisopropyl)-
"Methadon" butanone-2)

NO



G. Miscellaneous central nervous depressants

Scopolamine Hydrobromide

USP

Apomorphine Hydrochloride

USP

Bulbooxamine

NO

Magnesium salts (only when injected)

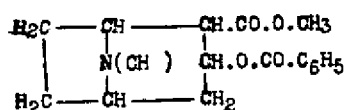
Cannabis (Marihuana, Hashish)

DRUGS USED MAINLY FOR THEIR EFFECTS ON THE PERIPHERAL NERVOUS SYSTEM

A. Local anesthetics

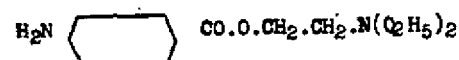
Cocaine (benzoyl methylecgonine)

USP



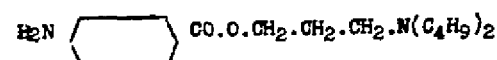
Procaine (diethylamino-ethyl ester of para-aminobenzoic acid)
("Novocaine")

USP



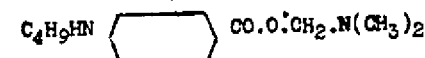
Butacaine (dibutyl aminopropyl ester of para-aminobenzoic acid)
("Butyn")

USP



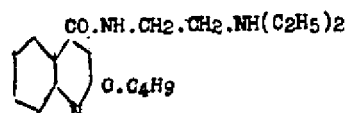
Tetracaine (dimethylamino-ester of butylpara-aminobenzoic acid)
("Pontocaine")

USP



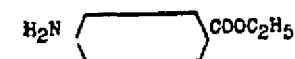
Nupercaine (2-butylxyquinoline carboxylic acid-diethyl-ethylenediamidamide)

MNR



Ethylaminobenzoate
("Benzocaine")

USP



Butylaminobenzoate
("Butesin")

USP



B. Specific neuromyal blocking agents

Curare and derivatives
("intocostrin", curarine, d-tubocurarine)

NO

6/1/50

Erythrine derivatives
(B-erythroidine) NO

Quinine methochloride and ethochloride
Large doses of choline, acetylcholine, nicotine and strychnine. NO

Quaternary alkyl derivatives of NH₄ NO

II. Stimulants of Nervous System

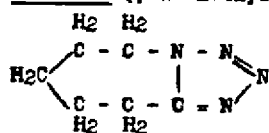
1. Convulsants (analeptics)

Strychnine USP

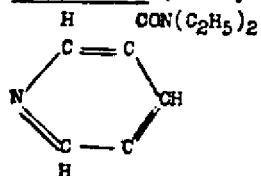
Caffeine and congeners USP

Picrotoxin (C₃H₃₄O₁₃) USP

Metrazol (pentamethylene tetrazole - "Cardiazol", Leptazol) MNR



Nikethamide (diethyl nicotinamide, "Coramine") MNR



Lobeline
(alpha lobeline) NO

2. Anticholinergic agents

Physostigmine (Eserine) USP

Neostigmine ("Prostigmine") USP

REPRESENTATIVE DRUGS AFFECTING CIRCULATION

I. Drugs acting predominantly on the nerve mechanisms that control the circulation.

A. In the Central Nervous System

1. Centers regulating heart rate.
Cardioaccelerator center stimulation - only as part of generalized stimulation of centers controlling sympathetic nervous activity by convulsant (analeptic) drugs. None used primarily for central cardioaccelerator stimulation.

Cardioaccelerator center depression - only as part of generalized depression of these centers by narcotic drugs. None used primarily for this purpose.

Cardioinhibitory center stimulation - important effect of digitalis group and veratrum alkaloids (see below); action due largely or entirely to reflexes aroused by drug in heart or lungs.

Cardioinhibitory center depression - only as part of generalized depression by narcotic.

Reciprocal innervation between cardioaccelerator and cardioinhibitory centers: increased activity of one automatically produces decreased activity of the other.

2. Centers regulating tone of blood vessels.

Vasoconstrictor center stimulation and depression like cardioaccelerator with which it is closely associated. Rise in blood pressure following analeptic (other than ephedrine-benzedrine type) is one of most favorable signs, fall in blood pressure following a narcotic one of most unfavorable.

Vasodilator center (or centers) - no important drug actions recognized.

B. In the Blood vessels

1. Pressure-sensitive nerve receptors in carotid sinuses and aortic arch - no drug actions known. Nerves can be inactivated by local anesthetics.

2. Chemo-sensitive nerve receptors in carotid and aortic bodies - stimulated by: - (a) inhibitors of oxidations like cyanides and sulfides; (b) nicotinic agents like nicotine, lobeline, choline derivatives etc.; (c) miscellaneous agents like papaverine, aminophylline, K⁺ salts etc. Depressed by excess nicotine, lobeline or choline derivatives. Nerves same as from pressoreceptors of carotid sinuses and aortic arch.

3. Sympathetic and parasympathetic ganglia (see Autonomic Pharmacology) stimulated by small doses of nicotine, lobeline, choline etc, depressed or paralyzed by larger doses. Usual effect is vasoconstriction and rise in blood pressure followed by vasodilatation and fall.

4. Sympathetic nerve endings stimulated by sympathomimetic agents (epinephrine, ephedrine, amphetamine, neosynephrine, pargoline, etc., etc.), paralyzed by sympathetic blocking agents, (ergotoxine, ergotamine, priscol, dibenamine etc.)

5. Parasympathetic nerve endings stimulated by muscarinic agents (muscarine, choline derivatives, physostigmine, neostigmine etc.), paralyzed by atropine and derivatives (scopolamine, homatropine, syntropan etc.)

6. Special nerve receptors in heart (perhaps also in liver) when stimulated by digitalis group reflexly produce increased activity of cardioinhibitory and vomiting centers.

7. Special nerve receptors in coronary and pulmonary circulations when stimulated by veratrum alkaloids reflexly produce slowing of heart, dilatation of blood vessels and inhibition of respiration.

8. (Uncertain but probable) - receptors in muscles reflexly produce increased heart rate and blood pressure during muscular exercise.

u/6/50 JAP

II. Drugs acting predominantly on the heart.

- A. Stimulation of contractility of heart muscle -
 - 1. Digitalis group: USP - digitalis, digitoxin, digoxin, lanatoside C, ouabain, strophanthin, MNR - digalen, digifolin, digilanid, digipoten, digitan, digitol, gitalin, scillaren.
 - 2. Xanthine derivatives - caffeine, theophylline, aminophyllin.
 - 3. Barium ions (purely toxic).
- B. Depression of contractility of heart muscle
 - 1. Quinine and (particularly) quinidine; atabrine.
 - 2. Narcotic drugs (particularly chloroform).
 - 3. Potassium ions.
- C. Interference with conduction of cardiac impulse.
 - 1. Between auricles and ventricles - digitalis group.
 - 2. Among muscle fibers - quinine and quinidine.
- D. Changes in coronary circulation
 - 1. Constriction - posterior pituitary extract and its pressor component (pitressin).
 - 2. Dilatation - nitrites (ethyl, amyl, sodium), nitrates (glyceryl, mannitol etc.), xanthine derivatives (caffeine, theobromine, theophylline), choline derivatives (acetylcholine, mechohyl, carbachol, urecholine), papaverine, nikethamide, epinephrine and other sympathomimetics, etc.
 - 3. Changes in cardiac work (increased by epinephrine group, xanthines, nikethamide; decreased by vasodilator drugs or any fall in blood pressure).

Chemotherapy

12-13-48

(Definition - OK)

Ideal - increasing access to tissues of parasite & no effect on tissues of host.

chemotherapy - directly associated with control of parasites, distribution of drugs to infected tissues & organs.

I. ^{P.1} oldest type - (older as it's older than now - ^{in the hand} ^{selected})
(~~chemo~~ - ~~chemo~~ - ~~chemo~~ - ~~chemo~~)

A. (a) & (b) for the moment

(f) protolytic compounds - directly to tissues of parasites

way they act { a. directly
+ directly to the control of parasites
to all the tissues
down their fold

to the - control of parasites

Cytopathic products (inactivated, they don't have any effect on the host)

a-

b-

c-

d. so as to control parasites & not the host

Idea - selected action (drug to control parasites & not the host)

D- see notes - first to be clearly

C- Amphoteric

amphoteric B - not the common drug but
more of the amphoteric B, & the amphoteric B
others. Amphoteric B will be used out completely.

Two main categories of treatment:

a. Iodine only - only by mouth

b. arsenic - other

amphoteric B (K. H. H. H.) for control of parasites & not the host

Continued - no A & I ⁽²⁾ ~~1/15/50~~
used for culture in to America. (4 places) to
produce something in developing nations

$$d_{\text{molecule}} \rightarrow \text{abundance of lines.}$$

Spirochaeta pallida (formerly)

Related Biologically To Thyranosaurus which caused
african sleeping sickness.

used 140 canisters. → till 1910.

penicillin - 1000 (1000000) units

lyph - really need cogn. power of this.
position used as lyph.

④ Malikene olive - not an insect cell.

a study C & group were hunting for coal for
transportation (colonization of Africa)

• about 1.5 mm submerged with NaOH Na₂CO₃ → di-Na
substance substance substance substance substance substance

12-13-48

as did still in rapid fatality. ^{4/1/50} ²⁵
 Co made 3,000 more → new solution (9.4)

- ① no protect
- ② used by spraying - covered dist. treatment

It's at first that needed reduced frequency for
 out stations & groups (W. side) found that needed
 smaller quantities on account of favorable temperature
 ratio. So got "in a hurry" and did for internal heat done by 1 dose

pyrethrinoid - used in neurological complications
 (poisoning) → retards more than placebo did.

C. in Mexico

not completely satisfactory - probably on way
 out of this country
 found as originators still doing it
 the rest of it - got already working body at this time

malaria

1st infection order

Amishon Maltz - brought back by one of Spanish
 → Amishon Maltz
 produced curing body - all compounds by him.

Quinine (legend of Amishon & Walter)

→ seeds smuggled out of F.A. - shown mostly in
 but not last bodies.

Alabrine - Really partially Amishon product

→ 1st time (U.S.A.) and subsequent cross work
 trying the amount & of Q → one 1;
 to parents

[illegible]

SAC, PHILADELPHIA

July 7, 1950

1. SCOTT MILLER, SA

HARRY GOLD, was.,
ESPIONAGE - R

65-4307-1B 12 (1) Folder No. 10

The above material was shown to GOLD on June 24, 1950 at which time GOLD explained the material therein.

The application for employment with the Sun Oil Company GOLD stated, he obtained in 1948 with the idea of seeking employment with this company. He said that he never filled out the application.

GOLD said MORRELL E. DOUGHERTY on behalf of GOLD, called up JOHN ASHENFELTER of the Sun Oil Company in August 1948 and told ASHENFELTER that GOLD would be in to see him relative to a position. GOLD said he did go to see ASHENFELTER but that the meeting was very unsatisfactory inasmuch as there were so many interruptions and it was suggested that another meeting be held but since the position at the Philadelphia General Hospital was about to materialize at that time, GOLD never followed through with the Sun Oil employment.

The material marked "1" GOLD stated, were notes in his handwriting on literature searches for EDDY QUICK who owned the Peacock Gold Leaf Company, in which building the Laboratory of A. BROTHMAN and Associates is located.

The last 8 pages of notes GOLD stated were notes on work at Abe Brothman and Associates relative to the Diol process. Some of this material is not in the handwriting of GOLD.

TSM:ENC
65-4307

SUN OIL COMPANY

APPLICATION FOR EMPLOYMENT

ANSWER ALL QUESTIONS COMPLETELY IN YOUR OWN
HANDWRITING. USE (X) IN SQUARES PROVIDED

Date _____

Referred by _____

FOR PERSONNEL DEPARTMENT USE ONLY

1	2	3	4	5

Remarks:

W/1/50

FOR POSITION AS _____

SOCIAL
SECURITY No. _____

OTHER POSITIONS FOR
WHICH YOU ARE QUALIFIED _____

NAME (First) _____ (Middle) _____ (Last) _____

ADDRESS (Street No.) _____ (Street) _____ (Apt. No.) _____

(City) _____ (State) _____ (Telephone No.) _____ (Years of Residence in this locality) _____

IN CASE OF ACCIDENT
NOTIFY (Name) _____ (Relationship) _____

(Address) _____ (Telephone) _____

MARITAL STATUS	NOTE: IT IS ESSENTIAL THAT THE AGE DATA BE ACCURATELY INSERTED.	NUMBER OF DEPENDENTS	
		Under 16	Over 16
Single <input type="checkbox"/>	AGE _____		
Married <input type="checkbox"/>	DATE OF BIRTH (Mo.) _____ (Day) _____ (Year) _____		
Widow <input type="checkbox"/>	HEIGHT _____ WEIGHT _____ SEX _____		
Widower <input type="checkbox"/>	OWN HOME <input type="checkbox"/>		
Divorced <input type="checkbox"/>	RENT HOME <input type="checkbox"/>		
Separated <input type="checkbox"/>	ROOM <input type="checkbox"/>		
	LIVE WITH RELATIVES <input type="checkbox"/>		

Were you ever employed by Sun Oil Company? Yes ☐ No ☐ If yes, Where? _____

Why did you leave? _____

Are you related to anyone now employed by Sun Oil Company?	Yes <input type="checkbox"/> No <input type="checkbox"/>	If yes, — Name	Department	Relationship

DESCRIBE ANY MILITARY TRAINING YOU HAVE HAD _____

EDUCATION		DATE OF	Honorable	
		Entry	Discharge	Dishonorable
PRIMARY SCHOOLING				
High School <input type="checkbox"/> or Prep School <input type="checkbox"/>				
COLLEGE OR UNIVERSITY	Day School <input type="checkbox"/> Night School <input type="checkbox"/>			
OTHER	Day School <input type="checkbox"/> Night School <input type="checkbox"/>			

Chemical Abstracts
EASTON, PA.

GOLD, HARRY
6635 KINDRED ST
PHILADELPHIA 24 PA

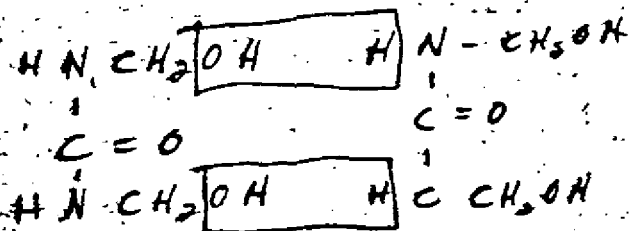
ENTERED AS SECOND CLASS
MAY 1961 AT EASTON, PA.

U/6/50
2/2/70

we planted 7 C.P. T. plants. These
 can be seen in the water
 filter and under the water net
 for several days at a time.

all

1.50 °C for 1 hr → ~~one~~ di-Hermin


$$A,5(A) =$$

4.5(A) =

6/6/50
700

vac.

water
to be
distilled

[illegible]

(12)

Lib. 2, 191, 957

Feb 27, 1940

Donald C. Edgar and Paul Robinson (to Du Pont)

Improved urea formaldehyde resins
composition (for decorative and protective
film)

4/4/50

~~1.6 x 10³~~

$\frac{370}{30}$

$$\frac{1000(0.37)}{30} = 19.7$$

30

1971

1800

1770

CH₂O

199

MAHLEPOY

pH 7.6

MAO 14

9.0

Tu, 11/11

Sta 96 km.
net 20-3000

cutting beam 12-24 km.

90 km at solid main Vitab

Filter & Bag

3500, 1000 units

all 1000 y 1000000

100 Tonne

7500000 at

15 mth range

net 1000

to 1000

permeability of films
to H_2S

6/6/50
gm

C.A. 24-30 (1927-1936)

29, 42-48¹ unimolecular films on glass in q.

25, 47-49² soly. in various solvents ✓ in q.

C.A. 31 (1937)

3987⁴ permeability of skin to ✓ in q.

C.A. 32 (1938)

nothing

C.A. 33 (1939)

nothing

C.A. 34 (1940)

7123⁵ rating coating resistance to ✓ in q.

C.A. 35 (1941)

6610⁶ permeability of skin to ✓ in q.

C.A. 36 (1942)

5107² abs. by amide IR. 4 for rate
of group II info. ✓ in q.

amine-formaldehyde test
renew

C.A. 37 (1943)

✓ 1206² stability of varnishes and resins
against

(2)
reservability

CA 37 - (cont'd)

p 741' } wood preservative impregnated
p 741' } and in 4.

CA 38 (1944)

nothing

6/6/50
ms

CA 39 (1945)

nothing

CA 40 (1946)

nothing

C.A. 37, 1206² (1943)

6/16/50
JLP

Hydrocarbonaceous materials; org. dielectric,
Trendy Vassoyuy. Elektrotekh. Inst. 1940,
no. 37, p. 82-99; Khim. Refrat. Zhur.
1940, no. 12, p. 76

m. F. Rykhalova

Stability of Resins and Varnishes against
various chemical environments

R. investigated the behavior of varnishes
from various synthetic condensed and
polymerized resins, esters, celluloses,
bitumens and natural resins in H_2S ,
 CO_2 , SO_2 , SO_3 , NO_2 and AcH vapors.
Films from alkyd varnishes, albertol,
asphalt, and combination films
from Varnish 1154 plus alkyd resin
oil-sol. resin or albertol withstood
best the action of H_2S .

Tanning

C-A 21-30 (1927-1936)

6/6/50
JMS

✓ 30, 760⁹ of gills in contact with
paper

27, P 1606⁹ prevention of ✓ m. q. oxidizing

24, P 3492² by coating metals (m) ✓ m. q. ^{large + 1} ^{acidulated}

✓ 24, P 3023³ m. of electroplated articles

25, P 2532⁶ m. of Ag ✓ m. q. 4.50 heat

23, P 593⁵ m. of metal articles ✓ m. ca. article
exposed

✓ 30, 7246⁷ of metals

✓ 27, P 4675³

26, P 3478³ of Ag ✓ m. q.

28, P 3542⁶ of Ag ✓ m. q. Ag. a. by coating

✓ 29, 2127⁴ of Ag

29, P 3075⁴ of Ag ✓ m. q. chromic and iron

27, P 1606³ of Ag & alloys ✓ m. q. ^{heat surface with}
^{HCl, NH₄ a. p. H}

23, P 1654⁴ of Ag & alloys ✓ m. q. electrodes of Cu

✓ 24, P 5119⁷ common for m. Ag

29, P 1183² common for m. Ag ✓ m. q. ^{with Cu Cl₂}
^(m. 100)
^{+ Cu SO₄}
^(for 1.5)

(E)
Tarnishing

6/6/50
710

12-23-47

21, P 2175² of Ag, etc. ✓ use ZnOAc over for tarnishing

28, 7170⁶ of Ag on silverware ✓ in Ag electrodes.

✓ 28, P 2485² of silverware, counts for m. - oxide treatment

28, P 2762^{3,4} of silverware counts for m. ✓ in Ag also in removed

29, P 8394³ of Ag wire by silver (m.) ✓ in Ag. removed due by Ag

27, 3340⁹ cloth for m. Ag ✓ in Ag. use ZnOAc & Ag salts

27, P 4692⁷ cloth for m. Ag ✓ in Ag. long fabric with PbOAc

29, P 4958¹ cloth for m. Ag ✓ in Ag. long cloth with Ag bromide

25, P 1959² resinate for m. Ag ✓ in Ag. PbOAc treat.

26, P 2934² resinate for m. Ag ✓ in Ag. PbOAc

27, P 1439⁵ resinate for m. Ag ✓ in Ag. PbOAc

24, P 4129³ m. Ag, wrapping for ✓ in Ag. counts Ag, cell

✓ 27, 2922⁷ of Ag & its reaction

25, 5652⁹ of Ag by 5 caps. ✓ in Ag. names Ag & S

24, P 3207⁷ Ag resist to tarnishing ✓ in Ag. treat with Cu, Ni, or I (oxidized)

23, P 4081³ Ag resist to ✓ in Ag. abras of Hg resist on surface (oxidized)

23, P 177⁹ Ag resist to ✓ in Ag. oxide metal treating

✓ 24, 602⁷ m. metal treating

C.A. 27, 29227 (1933)

6/6/50
JUN

überflächentechnik 10, 93-4 (1933)

H. Reinhardt

Protection against Tarnishing of Ag

Caused by H₂S or S-compounds, removed
by use of (among others) "Japan lacquer"
or similar varnishes.

see orig. article

C.A. 24, 5119 (1930)

6/6/50 PM

21.8. 1, 773, 702

aug 19, 1930

R. O. Bailey & Wm. S. Murray (to Oversea
Community Ltd)

Rendering silverware resistant to Tarnish

Polish with mixt of kerosene and
lampblack contg Iodine and free
from Tarnish producing ingredients

See orig Patent

C.A. 29, 2127⁶ (1935)

6/6/50
JW

with. Forchungsintitut Proberant

Edelmetalle 2, 61-7, 77-84, 105-12
(1934/35)

C. Raub

Treatment of Ag and its Prevention

only one that applies here is the
first one — "coating with a trans-
parent lacquer, Japanned" — to
make the surface resistant to S
compounds.

see orig. article

C.A. 36, 72469 (1936)

6/6/30
70

diffusion - Ztg 63, 374-5 (1936)

Josef Augustin

new preservation methods

Recent improvements in methods for
(among other things) preventing tarnish
on metal surfaces.

orig article

C.A. 24, 3023 (1920)

6/6/50
JW

Dec. 309, 339

June 7, 1923

H.A. Schwen & B.F. G. Smith

Prevention of Tarnishing of Electroplated
articles

S. & G. dip the article into a solution
of cellulose & soap in Am OAc and
heat at 32°C for 1-4 hrs.

CA 30, 360' (1936)

Pulp Paper Can 36, 609-13 (1935)

O. J. Schierholz

4/16/50
JNO

Tanning of Sulfur

Sulfur into of bronze powder type
— used to decorate wallpapers — tanned
due to generation of SO_2 from sulfur (or
dried by paper maker's alumin). S. & Co.
This is an extremely delicate test for SO_2
prevents by treating sulfur with $NaHCO_3$
prior to gilding.

A. 27, 46753 (1933)

6/6/50 JHC

June 20, 1933

333, 430

C. T. Huber and R. H. Scholtz (to The
Oreida Community Ltd.)

CuSO₄ is added to paper pulp to
precipitate the iron on inactive
form. The paper formed does not stain
with A 4 or other toxic metal.

A 24, 60277 (1930)

Publ. of I, no 75, 47 (1930)

1/6/50
gm

Tried - Stone - Polymer

Tanned metal tissue (Fabric)

seems non-tanned lacquer formula

see orig article (missing from files of
Eng. Soc. Library)

C.A. 32, 9222 (1938)

Trans. Faraday Soc. 34, 767-74 (1938)

T.P. Hoar and L.E. Reich

6/6/50
JH

Electrochem. Interpretation of Wagner's
Term of Tarnishing

Regard metal surface undergoing tar-
nishing as current producing cell with
the metal/film and film/attacking
substance as anode and cathode resp.,
the film being both the electrolyte and
the external circuit.

C.A. 32, 73456 (1938)

4/6/50
JED

Z. Physik. Chem. B 40, 435-75 (1939)

cf. C.A. 30, 51652

Carl Wagner and Carl Szwed

Theory of the Tarnishing Process - III

w. a measured rate of oxidation

of Cu, Ni and Ni-Au alloys

see orig articles

Tanning

6/14/50
JH

C.A. 32 (1938)

✓ 83 45⁶ of water

83 43¹ of A4 + A4 alloy ✓ in 9. side piece with
small cotton rod in
(covered 90,000 samples)

4506⁵ A4 plate peeling ✓ in 9 alloy with Rh.

✓ 90-2-27 T-200

CA 34 (1940)

6/6/50
JW

4045 of Aq and its prevention ✓
2600 Paper for pr. of Aq. ✓
and
autates

Tanning

6/6/50
JND

CA 35 (1941)

- 1008 of lacquered brass articles ✓ due to
3037 by rust or printing ink ✓ due to
✓ P 17562 pr of brass + A4
P 431 pr of A4 alloys ✓ cont with re
P 4292 pr of A4 + A4 alloys ✓ cont with re
7919 sheet of dental A4 alloys ✓

Tambling

4/6/50
JMS

26 (1942)

67592 total orientation in \sqrt{m} & me
57592 effect on static field of
brain, on a total \sqrt{m} & me

27942 timing of \sqrt{m} & me

44241 timing of \sqrt{m} & me

C-A 35, 1756³ (1941)

6/6/51

Remove H_2O by distillation, adding more NaOH if necessary, until the solution is anhydrous.

Filter the lacquer.

Then, prepare an alkyd resin thus:

6. Ident	{	phthalic anhydride	12.7 parts
		glycerin	9.0 parts
		castor oil	1.92 parts

until a suitable viscosity is reached. The resin is compatible with the lacquer and is suitable for a plasticizing agent.

7. For use, mix

40 parts, melamine lacquer

60 parts alkyd resin

and cut with NaOH to a viscosity suitable for spraying. Dry to get a thin continuous film & dry at 150°C for 15 mins. Film is not only tough resistant, but does not burn or soften (under a lighted cigarette).

C.A. - 37 (4943)

4/6/50

N. 23327 of road material, rev.

Tambling

C.A. 29 (1945)

nothing

6/6/50
JW

Tanning

6/6/50
JH

C.A. 40 (1946)

✓ 2653⁴ of 19, angle resin in pr.

P 4542¹

P 4654¹

} of 19 wall, pr.

✓

in 9

amine contents

A 40, 26534 (1946)

Industrial Finishing 22, 24, 27-8, 30,
32, 34, 36, 38, 40 (1946)

6/6/50
W

subject c. close

Synthesis in Industrial Finishing

characteristics & uses of new resins
and of the two general types of plastic
resins are noted. The softer varieties of
aliphatic resins are sometimes used as base
coats for textiles and in rubber lacquer
and specific use is to prevent skin from
turning. "chemically rubber applica-
tion and its uses are included in the
discussion."

CA 37, 2332 (1943)

6/15/52

U.S. 2,303,504

Dec 1, 1942

concerning J. Ryan (To Du Pont)

Rendering metal powders Tarnish Resistant

Particles of a lamellar disintegrated bronze powder or the like are coated with a partially polymerized urea-formaldehyde - aliphatic mono-hydric alc. condensation product by polishing at an elevated temp. not exceeding 145° F.

Orig. Patent

appl. Aug 14, 1941

Large bronze powders used by stamping operations are extensively used in the form of imitation gold transfer leaf as used in bookbinding and in general embossing work. of particular importance is high tarnish resistance. Invention provides a metallic powder of improved brilliancy and tarnish resistance. a further object is a process producing powders of greater commercial value, without sacrificing the brightness.

C.A. 37, 235-27 (1943) cont. d. 6/6/50
hiding power, or leafing characteristics of
the powder. also, can use ducting com-
mercial apt.

3. Describes usual method of preparing
bronzes powders:
- a. stamping → small miniature plates
 - b. separation of finer particles →
finally a small overall dirt, however,
at any point in the process the
powder represents a composite of
of a great variety of dirt.
 - c. Polishing operation to take care
of:
 - (1) clumps of plates
 - (2) crumpled particles
 - (3) improving brilliancy and
hiding power of particles.
 - (4) improve leafing of powder
(i.e., tendency of powder to
float & leaf). This is ac-
complished by adding a
small amt of a waxy
material such as stearic
acid, palmitic acid, can-
dellila wax, carnauba

6/16/50
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(2) cont'd

A 37, 43327 (1943)

wax and similar materials during the polishing operation. The type of agent is dependent on the use to which the powder is to be put.

4. R. has now found that by treating the powder during the polishing procedure with a small amount of a partially polymerized resinous, intra-formaldehyde condensation product dissolved in an organic solvent vehicle, are placed in a conventional polishing drum and polishing carried out until the said condensation product is uniformly distributed over the surface of the metallic powder. The final stages of the resin condensation occur during the polishing operation and probably continue for a short period thereafter.

5. Example:

course → Fine

a. 50 lbs of B-grade (F, D, B, A and AA) are put into a conventional stationary polishing drum with

CA 31, 23327 (1943) cont'd

6/4/50
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corrugated interior walls, equipped with soft brushes which rotate axially in contact with the inner wall of the drum.

b - The brushes are started and the powder are polished dry from b to 12 lbs. During this cycle the bulk density of the powder increases and the clumps of particles are broken up.

c - The second stage polishing (which can be conducted at any time following the dry polishing) is accomplished by loading a polishing drum with 50 lbs of

the powder from "b" and then (1) allowing the brushes to run for 30 mins (or until the mass has warmed appreciably and all the clumps are broken up) a temp of 145°F is reached.

(2) 4 ounces of partially condensed urea formaldehyde resin soln. are then added and the polishing process continued for 90

C-A-57, 2552 (1943) cont'd 11/13/50
minutes, with the temp. in the drum
averaging 145°F (but should not
appreciably exceed this figure).

d. The powder at the end of this treat-
ment has a slightly greasy feel but
this disappears in a few hrs. The
treated powder is very brilliant, far
surpassing that secured by prior
practice, a result which is probably
due to a combination of

- (1) dry polishing
- (2) the reducing action of the free
formaldehyde in the resin soln.
- (3) the chemical action of the acid
phosphate catalyst.

In addition to the high initial
brilliance, the powder has an un-
usual and markedly superior resis-
tance to tarnishing on exposure to
conditions which quickly tarnish
the ordinary commercial grade
of stearic acid polished powder.
It has also been found that the
resin treated powder is particularly

C-A 27, 23327 (1947) cont'd

4/17/53
RB

adapted for use in gold transfer leaf where tarnish resistance is particularly important.

9 - The resistance of the coating was further tested by attempts to extract the resin with the best known solvents for the urea-formaldehyde resins. The conc. of resin added was 0.3% but it was possible to extract only 0.0009%.

10 - A suitable resin soln. is of the type described by Edgar and Robinson in U.S. Patent 2,191,957; wherein a monohydric alcohol is used as a suitable modifying agent to produce stable partially polymerized resin solns. adapted for coating purposes and readily baked (145°F - baking?) to a tough hard film. A resin prepared according to Example II of C and R but in which *n*-butyl alcohol was substituted

C.A. 37, 23327 (1945) cont'd 4/10/60
for isobutyl alcohol and butyl and
phosphate was substituted for the
phthalic anhydride catalyst con-
stitutes a satisfactory material. The
resin soln. contained 60% total
solids by wt. Further,

i. The urea-formaldehyde con-
densation products are particularly
adapted for use here as they are

(1) light in color

(2) easily prepared

(3) stable and easily handled

7. The use of aliphatic alcohols in
the synthesis is of great impor-
tance because it has been found
to be impossible to polish bronze
powders in the presence of ap-
preciable amounts of water. There
are available commercial types
of partially polymerized urea-
formaldehyde resins in which
water is used as the carrier
but there are satisfactory here
also.

C.A. 37, 2332 (1943) cont'd 6/6/50
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- (1) Similarly, emulsions of partially polymerized urea-formaldehyde resins are unsatisfactory due to the water phase present.
 - (2) mechanical dispersions or suspensions of solid partially polymerized urea-formaldehyde resins in organic liquids in which the resin is insoluble are also unsatisfactory, as it is impossible to obtain a satisfactory distribution of the resin during the polishing.
- It is important, therefore, that the urea-formaldehyde complex used
- (1) represent an incompletely polymerized material
 - (2) be in the form of a soln in a volatile organic liquid (n. BuOH is preferred though iso BuOH can be used; EtOH leads to inferior water resistance and amyl, hexyl and octyl alcohols show slower drying characteristics).
- Partial or complete substitution of the urea may be made by using

⑨
C-A 37, 23327 (1945) cont'd 6/6/50 JMS
the substituted ureas such as alkyl,
aryl, or acid ureas. The selection
of the completely satisfactory resin
soln. can be determined only by
the appearance of the treated bronze
powder. Excessive amounts of for-
maldehyde tend to dull the powder
and excessive amts. of acidic cata-
lysts tend to pit and corrode the
powder.

It is advantageous to use sufficient
urea-formaldehyde resin to com-
pletely and uniformly coat the
disintegrated powders. The amount
of resin used varies with the fine-
ness of the powder and so the
optimum quantity will vary
from grade to grade - insuf-
ficient resin leads to poor tar-
nish resistance; and excessive
amounts interfere with the polish-
ing operation, causing clumping
or sticking together of the particles
and it becomes impossible to
handle the powder. In general,
satisfactory results have been obtained

C.A. 37, 23327 (1943) cont'd 6/4/50
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over the range of various bronze powders by the use of between 0.7% and 0.5% by wt. of solid treating resin.

Other resins, both natural and synthetic have been evaluated in comparison with the urea-formaldehyde type. Tests include:

- (1) Alkyd resins
- (2) Phenol-Formaldehyde
- (3) Vinyl resins
- (4) Cumar
- (5) Damar
- (6) Rosin
- (7) Ester resin

The majority of tests failed to markedly improve the tarnish resistance of the treated powder. None of the resins tested equaled the urea-formaldehyde type. Can use method for Sn, Ni and Ag powders.

Modification Notes

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(1) If the powder is not removed from the polishing drum (or for that at coils); the treating solution may be added directly at the completion of the dry polishing cycle.

(2) In some cases (though as a rule the preliminary dry polishing is preferable for best results) the dry polishing step may be eliminated entirely and the resin solution added at the start of the polishing operation.

3. It has 10 claims.